

UNIVERSIDADE FEDERAL DE MINAS GERAIS

Instituto de Ciências Biológicas

Programa de Pós-Graduação em Ecologia, Conservação e Manejo da Vida Silvestre

**CARACTERIZAÇÃO E IMPORTÂNCIA DA MATÉRIA ORGÂNICA DISSOLVIDA
EM ECOSSISTEMAS AQUÁTICOS TROPICAIS**

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Belo Horizonte – MG

2016

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EM ECOSISTEMAS AQUÁTICOS TROPICAIS**

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RESUMO

Embora seja reconhecida a diferença na ciclagem de carbono entre sistemas aquáticos temperados e tropicais, os estudos sobre a importância da matéria orgânica nos processos ecológicos e biogeoquímicos ainda são escassos em ambientes tropicais, principalmente no Brasil. Nosso objetivo foi caracterizar espacial e temporalmente a matéria orgânica dissolvida-DOM no lago Dom Helvécio (DH, MG) e comparar por meio de estudos também em outros sistemas aquáticos (reservatório da Pampulha-BH/MG e lago Carioca-MG) o efeito das diferentes fontes de DOM na transparência, disponibilidade de nutrientes e espectro de absorção de CDOM. Amostras de água subsuperficiais foram coletadas (2012-2014) para análise de Clor-a, sólidos em suspensão, DOC, CDOM e nutrientes. Foram obtidos dados de profundidade e pluviosidade, além de medidas in situ da radiação PAR e UV. $SUVA_{254}$, $S_{275-295}$, $S_{350-400}$, S_R , M , $a_{CDOM440}$ e $a_{CDOM254}$ foram utilizados para avaliar a qualidade de DOM. DH foi mais transparente nos períodos chuvosos estratificados e o estudo de partição de $Kd(PAR)$ mostrou que embora 79% da atenuação PAR tenha sido devido ao tripton, a variação temporal de $Kd(PAR)$ foi melhor explicada pelas variações de Clor-a. Para DOM as diferenças sazonais foram maiores que as espaciais e a pluviosidade afetou positivamente a quantidade de carbono aromático, sugerindo uma entrada de DOM alóctone com as chuvas. Houve uma redução no volume de chuvas nos anos de coleta em comparação com anos anteriores e considerando o efeito da pluviosidade observado, é possível estimar mudanças também nos padrões sazonais de entrada de DOM e nutrientes com consequências para o ecossistema. Em contraste, o eutrófico reservatório da Pampulha apresentou DOM predominantemente de origem algal. Comparando os dois sistemas, demonstramos que distintas fontes de DOM afetam a sazonalidade de nutrientes, transparência e qualidade de DOM. Por fim, avaliamos o efeito das diferentes fontes de DOM no espectro de absorção de CDOM através de experimento de mesocosmos no lago Carioca. DOM alóctone teve forte efeito na curva de absorção de CDOM entre 300 e 400 nm, enquanto DOM autóctone aumentou a absorção nos comprimentos de onda mais curtos. Dessa forma, as características ópticas (como transparência e susceptibilidade de DOM para foto e biodegradação) dependem das fontes de DOM, o que gera uma série de consequências para o ecossistema.

Palavras-chave: DOM, CDOM, matéria orgânica, coeficiente de atenuação, radiação

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ABSTRACT

1
2 Although it is recognized the difference in carbon cycling between temperate and tropical
3 aquatic systems, studies on the importance of organic matter in the ecological and
4 biogeochemical processes are still scarce in tropical environments, especially in Brazil. Our
5 aim was to characterize spatially and temporally the dissolved organic matter-DOM in Lake
6 Dom Helvécio (DH, MG) and compare through studies in other aquatic systems (Pampulha-
7 BH reservoir/MG and Lake Carioca/MG) the effect of different sources of DOM on lake
8 transparency, availability of nutrients and CDOM absorption spectrum. Subsurface water
9 samples were collected (2012-2014) for analysis of Chl-a, suspended solids, DOC, CDOM
10 and nutrients. Rainfall and depth data were obtained, in addition to in situ measurements of
11 PAR and UV radiation. $SUVA_{254}$, $S_{275-295}$, $S_{350-400}$, S_R , M and $a_{CDOM440}$ $a_{CDOM254}$ were used to
12 evaluate the DOM quality. DH was more transparent in the stratified rainy periods and the
13 $K_d(PAR)$ partition study showed that although 79% of PAR attenuation was due to tripton, the
14 temporal variation of $K_d(PAR)$ was better explained by changes in Chl-a. Seasonal
15 differences in DOM were greater than spatial, and rainfall positively affected the amount of
16 aromatic carbon, suggesting an allochthonous DOM input with rains. There was a reduction
17 in rainfall in the sampling years compared to previous years and considering the observed
18 effect of rainfall, it is possible to estimate changes in seasonal patterns also in the DOM and
19 nutrients inputs with consequences for the ecosystem. In contrast, the eutrophic Pampulha
20 reservoir presented DOM predominantly from algal source. Comparing the two systems, we
21 show that different DOM sources affect the seasonality of nutrients, lake transparency and
22 DOM quality. Finally, we evaluated the effect of different DOM sources in CDOM spectra
23 through mesocosms experiment in the Lake Carioca. Allochthonous DOM had a strong effect
24 on the CDOM absorption curve between 300 and 400 nm, while autochthonous DOM
25 increased absorption at shorter wavelengths. Thus, the optical characteristics (such as lake
26 transparency and DOM susceptibility to photo and biodegradation) depend on DOM sources,
27 which cause several consequences for the ecosystems.

28 Keywords: DOM, CDOM, organic matter, attenuation coefficient, radiation

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APRESENTAÇÃO DA TESE

A estrutura da tese foi organizada conforme descrito nos seguintes tópicos:

- Introdução: abrange uma revisão bibliográfica sobre radiação, propriedades ópticas aparentes e inerentes e matéria orgânica em ecossistemas aquáticos. Os objetivos da tese estão descritos ao final da Introdução e os objetivos específicos se encontram em cada capítulo.
- Capítulo 1: manuscrito aceito com revisões no periódico *Anais da Academia Brasileira de Ciências*. Devido à notícia do aceite ter sido dada ao fim de Junho, não foi possível incluir as revisões sugeridas pela revista na versão do artigo nesta tese. No entanto, as sugestões e críticas feitas pelos revisores estão disponibilizadas no Anexo I da tese. Trata-se de um estudo realizado no lago Dom Helvécio (Médio Rio Doce, MG, Brasil) cujo objetivo foi avaliar a contribuição de cada componente opticamente ativo na atenuação da radiação fotossinteticamente ativa na coluna d'água.
- Capítulo 2: manuscrito formatado para ser submetido no periódico *Limnologica* após considerações da banca. Este estudo também foi realizado no lago Dom Helvécio e o objetivo foi avaliar as diferenças espaciais e temporais na qualidade e quantidade da matéria orgânica dissolvida considerando mudanças recentes na pluviosidade regional.
- Capítulo 3: manuscrito aceito no periódico *Journal of Limnology* e se encontra disponível em: <http://dx.doi.org/10.4081/jlimnol.2016.1359> . Este trabalho foi realizado no lago Dom Helvécio e no reservatório da Pampulha (Belo Horizonte, MG, Brasil). O objetivo foi comparar os efeitos das diferentes fontes de matéria orgânica dissolvida na atenuação da radiação fotossinteticamente ativa e ultravioleta e na disponibilidade de nutrientes nos dois distintos sistemas tropicais.
- Capítulo 4: manuscrito formatado para ser submetido no periódico *Biogeochemistry* após considerações da banca, e originado a partir de um experimento de mesocosmos no lago Carioca (Médio Rio Doce, MG, Brasil), como parte do projeto Carbon Cycling in Lakes (COCLAKE). Este projeto foi desenvolvido em parceria com o pesquisador dinamarquês Peter Anton Staehr. O manuscrito foi escrito na Universidade de Aarhus (Roskilde, Dinamarca) durante o meu período de doutorado sanduíche (2015-2016) e contou com a colaboração de pesquisadores dinamarqueses e brasileiros. O objetivo foi avaliar sobre condições experimentais os efeitos do aumento

1 das concentrações de matéria orgânica alóctone e autóctone na absorção espectral da
2 matéria orgânica dissolvida colorida.

- 3 • Considerações finais: abrange a aplicabilidade, contextualização e importância dos
4 resultados gerados com esta tese e as perguntas que surgiram com este trabalho.

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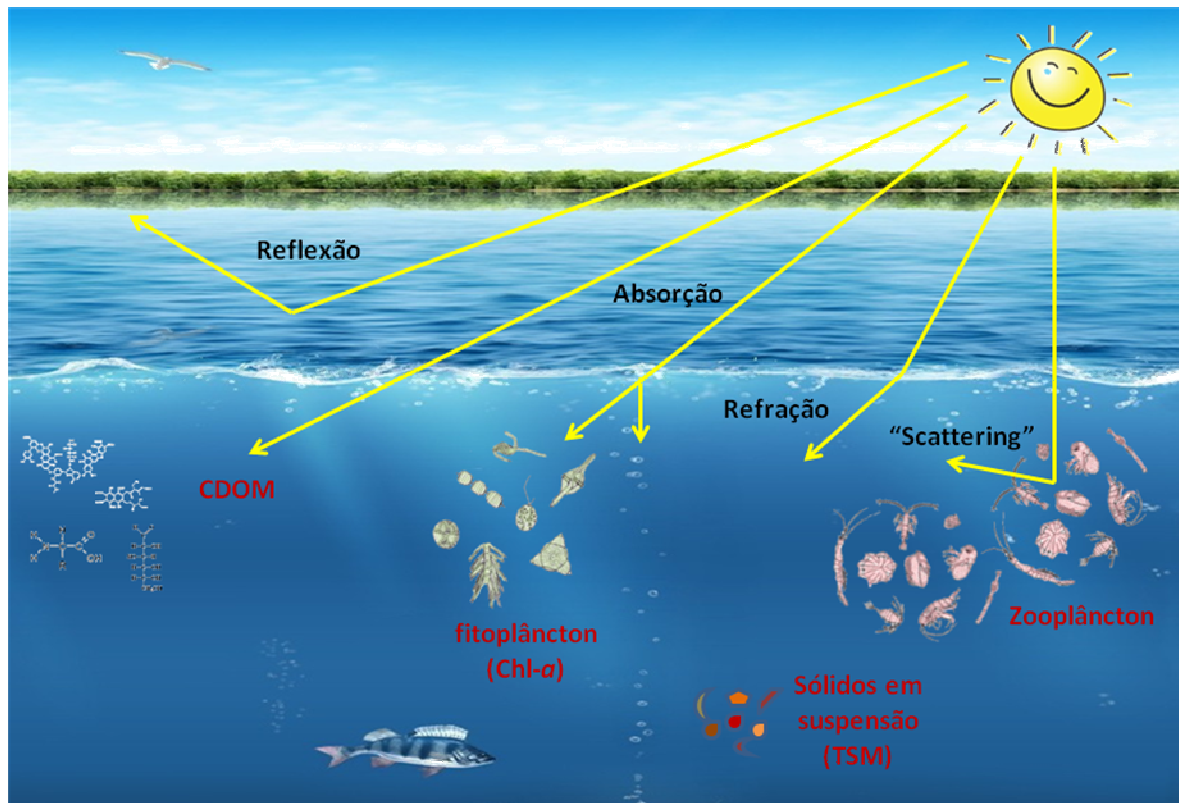
INTRODUÇÃO

A radiação solar e as propriedades ópticas dos ecossistemas aquáticos

O fluxo de energia nos sistemas aquáticos, assim como nos terrestres, depende da incidência de radiação solar, a qual influencia fortemente as condições químicas e biológicas dos ecossistemas afetando o metabolismo e funcionamento aquático (Parsons et al., 1984; Kirk, 1994; Helbling e Zagareze, 2003; Loiselle et al., 2009).

A radiação solar é subdividida em três faixas segundo a Comissão Internacional de Iluminação (CIE): ultravioleta (UV – comprimentos de onda entre 100 e 400 nm), visível ou radiação fotossinteticamente ativa (PAR – entre 400 e 700 nm) e infravermelho (maior que 700 nm). A radiação UV ainda é dividida em UV-C (100-280 nm), UV-B (280-315 nm) e UV-A (315-400 nm). UV-C é completamente absorvida pela camada de ozônio, chegando à superfície terrestre apenas UV-B e UV-A e PAR.

Os ecossistemas aquáticos possuem a capacidade de refletir entre 2 a 100% da radiação incidente na superfície da água, sendo esta quantidade dependente das condições da superfície (plana ou ondulada) e do ângulo de incidência dos raios solares. Por isto, em estudos de radiação subaquática o horário de amostragem é importante e deve ser limitado aos horários de menores porcentagens de reflexão, entre 10:00 h e 14:00 h. A reflexão somada à refração, absorção e dispersão da radiação solar são denominadas propriedades ópticas dos ambientes aquáticos (Kirk, 1994; Esteves, 1998). A água, alguns materiais particulados bióticos e abióticos em suspensão na coluna d'água (como plâncton e detritos), e a matéria orgânica dissolvida são responsáveis por absorver e/ou dispersar a luz solar, e são denominados componentes opticamente ativos (Kirk, 1994; Pfannkuche, 2002; Branco e Kremer, 2005; Kelble et al., 2005). A fração dissolvida somente causa absorção, enquanto as partículas orgânicas e inorgânicas causam predominantemente dispersão ("scattering"), exceto os produtores primários (fitoplâncton, macrófitas aquáticas) que contribuem em ambos os processos (Figura 1).



1

2 **Figura 1 - A radiação solar na superfície aquática e na coluna d'água.**

3 As propriedades ópticas são divididas em aparentes (AOP - apparent optical properties) e
 4 inerentes (IOP - inherent optical properties). AOP são aquelas que dependem da natureza
 5 angular do campo luminoso, como o coeficiente de atenuação difusa da radiação na coluna
 6 d'água (K_d) (obtida *in situ* através de radiômetros e disco de Secchi) e a reflectância para o
 7 sensoriamento remoto (Kirk, 1994; Morris et al., 1995; Effler et al., 2010). Em contrapartida,
 8 as IOP dependem somente das concentrações dos componentes opticamente ativos, e são
 9 representadas pelos coeficientes de absorção da radiação pela matéria particulada e dissolvida
 10 presentes na água (obtidos *in situ* com equipamentos modernos ou em laboratório através de
 11 amostras de água coletadas em conjunto com uma análise espectrofotométrica) (Morel, 1988,
 12 1991; Kirk, 1994; Huovinen et al., 2003; Zhang et al., 2007a). A atenuação da luz decai
 13 exponencialmente com a profundidade e é dependente dos coeficientes de absorção e
 14 dispersão, que por sua vez dependem das concentrações dos componentes opticamente ativos
 15 na coluna d'água (Obrador e Pretus, 2008). Dessa forma, a camada fótica (limite inferior é
 16 definido como a profundidade onde a intensidade luminosa corresponde a 1% da radiação que
 17 atinge a superfície) pode variar de poucos centímetros a dezenas de metros de profundidade
 18 dependendo da capacidade do meio em atenuar a radiação subaquática (Esteves, 1998).

1 A quantidade de energia luminosa e o tamanho da zona fótica disponível na coluna d'água
2 (principalmente a PAR) são fatores limitantes para a distribuição e abundância dos produtores
3 primários e a produtividade do sistema, que por sua vez afetam todo o funcionamento do
4 ecossistema (Parsons et al., 1984; McMahon et al., 1992; Gallegos e Moore, 2000; Zhang et
5 al., 2007b). Nos últimos anos, o interesse também sobre o efeito da radiação UV e seu
6 impacto subaquático tem aumentado devido a crescente incidência de raios ultravioleta por
7 causa da redução da camada de ozônio estratosférico e também pela modificação dos
8 ambientes aquáticos em função de poluição local e regional (Campbell et al., 2011).

9 *A matéria orgânica: origem, características e importância para os ecossistemas aquáticos*

10 A matéria orgânica (OM – organic matter) é constituída de uma fração particulada (POM –
11 particulate organic matter) e uma fração dissolvida (DOM – dissolved organic matter). POM é
12 representada pela biota aquática (por exemplo, bactérias, fitoplâncton, zooplâncton) e por
13 detritos orgânicos e inorgânicos subprodutos de decomposição e metabolismo aquático,
14 enquanto que a DOM por convenção representa os compostos orgânicos menores que 0,2-0,7
15 μm (Kalenak et al., 2013). A matéria orgânica dissolvida colorida (CDOM – colored
16 dissolved organic matter) é a fração de DOM capaz de absorver radiação solar nas faixas do
17 UV e PAR por possuir cromóforos em sua estrutura (Amon e Benner, 1996), o que influencia
18 as propriedades ópticas dos ecossistemas aquáticos.

19 DOM é o principal reservatório de carbono orgânico em ecossistemas aquáticos (Bertilsson e
20 Tranvik, 2000) e a sua dinâmica temporal e espacial tem recebido maior atenção nos últimos
21 anos devido ao seu importante papel em processos ecológicos e biogeoquímicos (Cole et al.,
22 2007; Tranvik et al., 2009). Também é atualmente reconhecido como o principal substrato
23 para a comunidade bacteriana e representa a base da cadeia trófica planctônica (Farjalla et al.,
24 2006). Através do fenômeno “microbial loop” o bacterioplâncton transforma DOM em POM,
25 a qual pode ser assimilada pelo protozooplâncton (Azam et al., 1983). Além disso, a
26 mineralização de DOM por bactérias heterotróficas proporciona a reutilização da forma
27 inorgânica pelos produtores para a produção primária (Azam et al., 1994).

28 Além da degradação microbiana (biodegradação), outra importante forma de degradação de
29 DOM é a fotodegradação, que devido à capacidade de CDOM em absorver radiação provoca
30 a oxidação parcial ou total da molécula, sendo a oxidação total a transformação de CDOM até
31 CO_2 . Sendo assim, a fotodegradação pode ser de enorme importância na mineralização do

1 CDOM, podendo ser comparada a processos como a respiração fitoplanctônica (Granéli et al.,
2 1996) e mineralização pelo bacterioplâncton (Amado et al., 2006).

3 A absorção da luz pelo CDOM ainda provoca uma série de consequências para os organismos
4 e processos aquáticos, dentre elas a competição pela luz visível com os produtores primários
5 (Kostoglidis et al., 2005) e proteção dos organismos aquáticos contra os raios nocivos UV
6 limitando sua penetração à superfície da coluna d'água (Granéli et al., 1998). Em
7 contrapartida provoca o fotobranqueamento da coluna d'água que consiste na perda de
8 coloração da água devido à fotodegradação do CDOM, o que expõe os organismos às
9 radiações nocivas (Del Vecchio e Blough, 2002). Sendo assim, as propriedades de absorção
10 luminosa de CDOM torna-o um importante regulador da produtividade ecossistêmica,
11 podendo favorecer a produção primária (através da proteção contra excesso de radiação PAR
12 e UV) ou inibir (devido à competição por luz), além de afetar a estrutura termal e
13 profundidade da camada de mistura da coluna d'água (Jones, 1988; Fee et al., 1996, Bezerra-
14 Neto et al., submetido).

15 Existem duas fontes de DOM em ecossistemas aquáticos continentais: fonte alóctone,
16 originada da degradação de vegetação e solo terrestre, onde DOM entra no sistema aquático
17 carregada pelas chuvas (Kieber et al., 2006; Miller et al., 2009); e fonte autóctone, produzida
18 dentro do sistema como resultado principalmente da degradação ou lise de células dos
19 produtores primários (Kritzberg et al., 2004; Guillemette e Del Giorgio, 2012). A composição
20 química e as características ópticas de DOM (qualidade de DOM) estão relacionadas com a
21 sua fonte, e possuem susceptibilidades distintas aos processos de degradação por micro-
22 organismos e fotodegradação (Wetzel et al., 1995; Bertilsson e Tranvik, 2000). A DOM
23 autóctone é formada por compostos simples de baixo peso molecular como carboidratos,
24 proteínas, lipídeos e aminoácidos (Wetzel, 2001), enquanto a DOM alóctone é constituída de
25 uma mistura de compostos aromáticos de alto peso molecular que formam as substâncias
26 húmicas (Thurman, 1985). As substâncias húmicas representam o principal componente de
27 DOM em sistemas aquáticos naturais e são constituídas de ácidos fúlvicos (compostos
28 hidrofílicos) e ácidos húmicos (compostos hidrofóbicos) (Piccolo, 2001).

29 O metabolismo bacteriano pode ser suportado por DOM alóctone e autóctone, mas existem
30 estudos que mostram que apenas uma pequena fração de DOM alóctone é incorporada como
31 biomassa pelas bactérias e transferida para a cadeia trófica (Cole et al., 2006; Roland et al.,
32 2010). Nos trópicos, as substâncias húmicas são importantes fontes de energia para a

1 comunidade microbiana, mas estudos demonstram que estão mais relacionadas com a
2 respiração do que com a produção bacteriana (Amado et al., 2006; Farjalla et al., 2009a). Em
3 contrapartida, as bactérias possuem a capacidade de assimilar rapidamente DOM autóctone de
4 origem algal (Chen & Wangersky 1996). Portanto, atualmente é aceito o conceito de que
5 DOM autóctone algal possui maior labilidade para a biodegradação (Farjalla et al., 2009b;
6 Fonte et al., 2013), enquanto DOM alóctone é mais susceptível para a fotodegradação e
7 absorve fortemente a radiação na faixa do UV (Amon e Benner, 1994; McKnight et al., 1994;
8 Benner, 2002; Helms et al., 2008). Devido à complexidade e alto peso molecular as
9 substâncias húmicas são recalcitrantes para a degradação microbiana, e por isso tendem a se
10 acumular em lagos podendo representar 80% de DOM total (Wetzel et al., 1995). No entanto,
11 após a fotodegradação parcial as substâncias húmicas podem se tornar foto-produtos
12 biodisponíveis para a comunidade microbiana aumentando a produção do bacterioplâncton
13 (Vähätalo et al., 2003). Nos trópicos, a fotodegradação e biodegradação são favorecidas
14 devido à elevada radiação solar e altas temperaturas durante todo o ano, embora poucos ainda
15 são os estudos nesses ecossistemas.

16 A concentração e qualidade de DOM em lagos podem ser heterogêneas verticalmente na
17 coluna d'água ou horizontalmente, em função da proximidade de fontes de carbono, como
18 tributários, margens com vegetação e outras características hidrológicas da bacia hidrográfica
19 (Wetzel, 1992; Tao, 1998; Porcal et al., 2004; Kostoglidis et al., 2005). Temporalmente as
20 concentrações de DOM também podem variar em função de aportes esporádicos ou sazonais,
21 sendo estes últimos controlados pelo regime de chuvas ou pulso hidrológico (Farjalla et al.,
22 2002, Suhett et al., 2006, Brandão et al., 2016). Embora a importância de DOM seja clara e
23 bem demonstrada, a distribuição de DOM no espaço e tempo ainda foi pouco estudada,
24 especialmente em lagos naturais tropicais e ambientes eutrofizados (Toming et al., 2009;
25 Zhang et al., 2011, Brandão et al., 2016).

26 Atualmente as metodologias mais acessíveis que permitem o estudo da composição e origem
27 de DOM são análises espectrofotométricas de CDOM, normalmente combinadas com estudos
28 de fluorescência (Helms et al., 2008). Através do espectro de absorção de CDOM podemos
29 extrair métricas que fornecem características sobre a qualidade das moléculas de carbono
30 como: aromaticidade (absorbância específica para UV - $SUVA_{254}$; Weishaar et al., 2003);
31 fonte, mudanças devido a fotodegradação e tamanho molecular ("spectral slope" - $S_{275-295}$;
32 Helms et al., 2008); mudanças devido a biodegradação ("spectral slope" - $S_{350-400}$; Helms et
33 al., 2008); relação entre fotodegradação e biodegradação ("slope ratio" - S_R ; Helms et al.,

1 2008); tamanho molecular estimado (“molecular weight = a_{250}/a_{365} ”; De Haan and De Boer,
2 1987; Peuravuori e Pihlaja, 1997). Tais métricas têm sido largamente utilizadas para analisar
3 as propriedades ópticas em ambientes aquáticos temperados (Zhang et al., 2007c, 2009;
4 Helms et al., 2008; Bracchini et al., 2010), mas poucos são os estudos utilizando essa
5 abordagem em sistemas tropicais, principalmente no Brasil (por exemplo Teixeira et al., 2011,
6 2013; Bittar et al., 2015).

7 Pretendo com esta tese contribuir com estudos de propriedades ópticas em ecossistemas
8 aquáticos tropicais. Além disso, estudos que avaliam particularmente as mudanças no
9 espectro de absorção de CDOM em função das fontes e transformações de DOM ainda são
10 escassos inclusive em ambientes temperados (Helms et al., 2013; Reader et al., 2015) e, até
11 onde pude verificar na literatura disponível, acredito ser um trabalho pioneiro nesta
12 abordagem (capítulo 4) em sistemas tropicais.

13 *Objetivos*

14 Esta tese teve como objetivo inicial o estudo espacial e temporal da distribuição e qualidade
15 de DOM no lago Dom Helvécio (Médio Rio Doce, MG, Brasil). Primeiramente avaliei qual a
16 contribuição de CDOM na atenuação da radiação PAR neste lago (capítulo 1). Neste capítulo
17 considerei quatro amostragens espaciais em 2013, sendo uma em cada estação do ano
18 contemplando os períodos de estratificação e mistura, assim como os períodos intermediários.
19 Além disso, incluí também amostragens mensais realizadas em um único ponto na zona
20 pelágica entre 2011 e 2012.

21 O entendimento da variação espacial e temporal de DOM no lago Dom Helvécio foi o
22 objetivo do capítulo 2. Para isso realizei seis amostragens entre 2012 e 2014, onde avaliei os
23 períodos de entrada do DOM alóctone com as chuvas.

24 No capítulo 3 o objetivo foi comparar dois sistemas aquáticos com diferentes fontes de DOM
25 quanto às características ópticas, qualidade de DOM e concentrações de nutrientes. Para isto,
26 utilizei duas amostragens realizadas no lago Dom Helvécio (seca e chuva em 2013) e incluí
27 duas amostragens realizadas no reservatório da Pampulha (Belo Horizonte, MG, Brasil) na
28 mesma época.

29 Por fim, o capítulo 4 teve como objetivo avaliar o efeito de DOM alóctone e autóctone no
30 espectro de absorção de CDOM. Para isto, realizei um experimento de mesocosmos no lago
31 Carioca (Médio Rio Doce, MG, Brasil), um lago próximo ao lago Dom Helvécio dentro dos

1 limites do Parque Estadual do Rio Doce. Foi manipulado neste experimento as concentrações
2 de matéria orgânica alóctone e nutrientes (o que estimula crescimento fitoplanctônico com
3 consequente aumento de DOM autóctone), assim como a disponibilidade de radiação solar.

4 Todas as amostragens foram realizadas em ecossistemas aquáticos em Minas Gerais. No
5 entanto, os capítulos e artigos foram escritos completamente ou iniciados na Universidade de
6 Aarhus, Dinamarca, durante meu período de doutorado sanduíche (entre abril de 2015 a
7 fevereiro de 2016). Assim, todos os trabalhos contaram com a participação do coorientador
8 Peter Staehr, e o capítulo 4 também com outros pesquisadores estrangeiros.

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CAPÍTULO 1

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Partitioning of the diffuse attenuation coefficient for photosynthetically available irradiance in a deep dendritic tropical lake

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Abstract

We studied the effects of particulate and dissolved optically active components on the attenuation of photosynthetic active radiation (PAR) in a tropical lake. The temporal and spatial distribution of tripton, Chl-*a* and $a_{CDOM}(440)$ and their relative contribution to the diffuse PAR attenuation coefficient (K_d) was investigated at 21 sites (dry and wet seasons and two intermediate periods) and at monthly interval at 1 pelagic site. Higher values of K_d were observed during the mixing period, characterized by a higher concentration of tripton and Chl-*a* compared to the stratified rainy season. In the spatial sampling PAR attenuation was dominated by tripton absorption/scattering (average relative contribution of 79%), followed by Chl-*a* (average 11.6%). In the monthly sampling tripton and Chl-*a* accounted for most of the K_d with relative contributions of 47.8% and 35.6%, respectively. Multiple linear regression analysis showed that Chl-*a* and tripton in combination explained 97% of the monthly variation in K_d ($p < 0.001$), but Chl-*a* had more influence (higher regression

1 coefficient). Thus, although most of light attenuation was due to tripton, seasonal variations in
2 phytoplankton abundance were responsible for most of the temporal fluctuations in *Kd*.

3 Key words: CDOM, chlorophyll-*a*, *Kd* partition, light attenuation, tripton.

4 **Resumo**

5 Nós estudamos os efeitos dos componentes opticamente ativos particulados e dissolvidos na
6 atenuação da radiação fotossinteticamente ativa (PAR) em um lago tropical. As distribuições
7 temporal e espacial do tripton, Chl-*a* e $a_{\text{CDOM}(440)}$ e suas contribuições relativas no
8 coeficiente de atenuação difusa PAR (*Kd*) foram investigadas em 21 pontos (estações seca e
9 chuvosa e dois períodos intermediários) e em intervalo mensal em 1 ponto pelágico. Maiores
10 valores de *Kd* foram observados durante período de mistura, caracterizado por maiores
11 concentrações de tripton e Chl-*a* comparados com o período chuvoso estratificado. Na
12 amostragem espacial, a atenuação PAR foi dominada pela absorção/dispersão por tripton
13 (média contribuição relativa 79%), seguido por Chl-*a* (média 11.6%). Na amostragem mensal
14 tripton e Chl-*a* contribuíram com a maioria do *Kd* com média das contribuições relativas de
15 47.8% e 35.6%, respectivamente. Análises de regressão linear múltipla mostraram que Chl-*a*
16 e tripton explicaram 97% da variação mensal do *Kd* ($p < 0.001$), mas Chl-*a* teve maior
17 influência (maior coeficiente de regressão). Contudo, embora a maior atenuação da luz
18 ocorreu devido ao tripton, variações sazonais na abundância do fitoplâncton foram
19 responsáveis pela maioria da flutuação temporal no *Kd*.

20 Palavras-chave: atenuação da luz, CDOM, clorofila-*a*, partição de *Kd*, tripton.

21

22

23

1 **Introduction**

2 Availability of photosynthetically active radiation (PAR; 400 to 700 nm) in the water column
3 provides the energy for primary producers and is a key determinant of the overall productivity
4 of aquatic ecosystems (Parsons et al. 1984, Kirk 1994). While it is well known that PAR is
5 attenuated exponentially with depth, little is known of the causes for temporal and spatial
6 variability in the vertical diffuse attenuation coefficient for down-welling PAR (K_d ; m⁻¹) in
7 tropical lakes.

8 Light attenuation in aquatic ecosystems is basically the sum of light absorbing and scattering
9 optically active components present in water. These components are phytoplankton pigments
10 (of which chlorophyll-*a* dominates), colored dissolved organic matter (CDOM), pure water
11 and tripton (defined in this type of studies as non pigmented particulate matter) (Kirk 1994,
12 Pfannkuche 2002, Branco and Kremer 2005, Kelble et al. 2005).

13 Studies aiming to determine the relative contribution of optically active components to K_d
14 have previously concentrated on shallow lakes (Somlyódy and Koncsos 1991, Blom et al.
15 1994, James et al. 1997, Van Duin et al. 2001, Pierson et al. 2003, Zhang et al. 2007a, Zhang et
16 al. 2007b, Balogh et al. 2009), estuaries and coastal lagoons (Phlips et al. 1995, Christian and
17 Sheng, 2003, Lund-Hansen 2004, Kelble et al. 2005, Yamaguchi et al. 2013), while little
18 information about partition of K_d is available on deep (Effler et al. 2010) and tropical lakes
19 (Loiselle et al. 2008). The partition of this optical property is based on the equation:

$$20 \quad K_d = K_W + K_{CDOM} + K_{TRI} + K_{Chl-a} \quad (\text{Equation 1})$$

21 where K_W is the partial attenuation coefficient by pure water, K_{CDOM} by CDOM, K_{TRI} by
22 tripton and K_{Chl-a} by Chl-*a* (Kirk 1994, Phlips et al. 1995).

1 The concentrations of optically active components and how these attenuate light in the water
2 column determine the amount of light available for photosynthesis. Therefore, the available
3 light restricts production, abundance, and distribution of pelagic and benthic primary
4 producers, with implications for higher trophic levels in the ecosystem (Parsons et al.1984,
5 McMahon et al. 1992, Gallegos and Moore 2000). Studies about *K_d* partition provide
6 important understanding of how each component is responsible for the attenuation of light
7 and this fact also allows predict the underwater light climate from the concentrations of these
8 components (Zhang et al. 2007).

9 Spatial variations in PAR light attenuation coefficients are expected in large and dendritic
10 lakes because of the differences in depth and proximity to the border, which leads to less
11 stability of the water column and increased input of allochthonous matter. The shallower
12 regions of the lakes are susceptible to wind driven sediment resuspension which increases the
13 concentration of suspended matter in the water column and reduces light penetration and thus
14 primary production (Somlyódy and Koncsos 1991). These variations also differ between
15 seasons due to higher input via rainfall (rain period) and due to mixing of water column (dry
16 period).

17 Lake Dom Helvécio (DH - Minas Gerais, Brazil) is a deep dendritic tropical lake located in a
18 preserved area surrounded by Atlantic Forest. Although there is a greater input during rainy
19 season, the water column stability in DH is large in this period (Brighenti et al. 2015) and the
20 allochthonous matter coming with runoff from rains remains retained below the thermocline
21 due to differences in water temperature (Reynolds 2009). Redistribution of particulate matter
22 into the upper mixed layers accordingly occurs during the dryer and colder season which
23 therefore is characterized by less transparent water. Following these observations, our
24 hypotheses are that: 1) the total suspended matter - TSM (Chl-*a* and tripton) present greater
25 relative contribution in the attenuation of PAR radiation at all the sampling stations and

1 seasons, as CDOM occurs at low concentrations in Lake DH and absorbs more in the range of
2 ultraviolet spectrum; 2) the spatial variability in PAR attenuation coefficient values among the
3 21 sites can be explained by the influence of the lake shore in the closest sites (direct input of
4 allochthonous organic matter and shallower water column compared to deeper sites); 3) the
5 PAR attenuation coefficient is higher during the dryer mixing period due to elevated levels of
6 suspended matter in the water column. In order to test these hypothesis, our objectives were to
7 document the spatial and temporal contribution of tripton, Chl-*a* and CDOM in total PAR
8 attenuation through *K_d* partition study at Lake DH and determine how variations in optically
9 attenuating conditions depend on the time and place of sampling.

10 **Materials and Methods**

11 *Study site*

12 Dom Helvécio lake is located inside the Rio Doce State Park – PERD Southeast Brazil
13 (S19°46.94', W42°35.48'). The lakes complex in the Middle Rio Doce is one of the most
14 important districts of lakes in Brazil, with approximately 130 lakes, 42 of them located inside
15 the PERD area, which is the largest remnant of the Atlantic Forest in Minas Gerais, totaling
16 36000 ha (Maia-Barbosa et al. 2010) (Fig. 1). This lake is one of the deepest natural lakes of
17 Brazil (maximum depth: 39.5 m), surrounded by Atlantic forest. It is oligotrophic (total
18 phosphorus: 3.0-22.0 µg L⁻¹, annual mean 11.2 µg L⁻¹; Chl-*a*: 0.5-11.0 µg L⁻¹, annual mean
19 4.5 µg L⁻¹), has a perimeter of 37.7 km, area equal to 5.27x10⁶ m² and volume equal to
20 5.94x10⁷ m³ (Bezerra-Neto and Pinto-Coelho 2008). Dom Helvécio is a warm monomictic
21 lake, with thermal stratification period beginning in September and lasting until April (rainy
22 period) and a mixed period from May to August (dry period) (Barbosa and Tundisi 1980).

23 *Field measurements*

1 The spatial sampling was carried out in January (Summer), April (transition-Autumn), August
2 (Winter) and December (transition-Spring) 2013 at 21 sampling stations in the lake (Fig. 1c).
3 Although in tropical environments the four seasons are not well defined and then are
4 commonly separated into dry and rainy periods in aquatic studies, we included in this study
5 the samplings during the transitions periods (autumn and spring) because they are important
6 intermediate periods between the rainy and dry seasons. Temporal samples were taken
7 monthly at a single station in the pelagic part of the lake (station 8, Fig. 1), from June 2011 to
8 November 2012. Six data of temporal sampling and one of the spatial sampling stations
9 (station 1 in January 2013) were not included in the analysis because the Chl-*a* measurements
10 were not analyzed in a timely manner and the values were of low accuracy.

11 Upon monthly sampling at the central deep region (station 8) of the lake, vertical profiles of
12 water column temperature (°C) were measured every meter from surface to 28 m using a
13 Hydrolab DS5 probe. The surface irradiance (E_0) was obtained by a solar radiation sensor
14 (WE300, Global Water®) mounted just above lake surface on a buoy located at this station.

15 For spatial analysis, vertical profiles of photosynthetically active radiation (PAR) (between
16 10:00 and 14:00) were performed using a radiometer (BIC Biospherical Instruments). The
17 profiles were made on the sun side of the boat avoiding shadows and measurements during
18 days with great clouds were avoided. The radiometer was attached to a battery and notebook
19 with the software LoggerLight™ (Biospherical Instruments) while collecting radiation data
20 PAR and depth during descent and ascent up of the probe. In the spatial sampling the
21 maximum depth of measurement of radiation was defined on the field when PAR reached 1
22 $\mu\text{Einsteins m}^2 \text{ s}$. For monthly analysis, light profiles were measured at depths 0.1, 1.0, 2.0 and
23 3.0 m using a model UA-002 light sensors (HOBO®).

1 Water samples (0.5 m from the surface) were collected using a van Dorn bottle (5L) and
2 filtered immediately for Chl-*a* (0.47µm Millipore filter) and TSM (AP040 Millipore filter),
3 and the filters were frozen until analysis. Water samples were also filtered for analysis of
4 dissolved organic carbon (DOC) and colored dissolved organic matter (CDOM) (0.22 µm
5 Millipore filter) and stored in amber glass bottles (pre-washed with distilled water and
6 hydrochloric acid 10%) at 4°C in the dark.

7 *Laboratory analysis and calculation of the optical properties*

8 The diffuse PAR attenuation coefficient (*K_d*) was calculated as the linear slope between depth
9 and the natural logarithm of the measured downwelling PAR radiation. From *K_d* we
10 calculated the depth of the euphotic zone ($Z_{eu} = 4.6 / K_d$; m) where 1% of the radiation
11 incident on the surface is reached (Kirk, 1994).

12 The measurement of concentration of Chl-*a* was obtained by acetone extraction (90%)
13 measured in a spectrophotometer (UV-VIS Shimadzu) at 665 and 750 nm and calculated
14 according to APHA (1998). Total suspended matter (TSM; mg L⁻¹) were determined by the
15 gravimetric method, considering the difference between the dry weight of AP40 Millipore
16 filters (105°C for 2 hours) before and after the filtration of water samples (APHA 1998). The
17 tripton represents the non-living fraction of the suspension material, but in this study it
18 represents the non-pigmented part of TSM (non-phytoplankton particulate matter). This was
19 obtained subtracting the dry weight of the phytoplankton from the TSM. The ratio between
20 concentration of Chl-*a* corrected by pheophytin and the dry weight of phytoplankton was
21 assumed 1:100 (Phlips et al. 1995). Thus, the concentration of the Chl-*a* corrected by
22 pheophytin (mg L⁻¹) was multiplied by 100 to obtain the dry weight of phytoplankton and
23 then subtracted from TSM (mg L⁻¹) to obtain the concentration of tripton (mg L⁻¹) (Christian
24 and Sheng 2003).

1 The DOC concentration (mg L^{-1}) was obtained by catalytic oxidation method of high
2 temperature using TOC Analyzer (Shimadzu TOC – 5000).

3 The spectral absorption of CDOM was measured at 440 nm ($a_{\text{CDOM}(440)}$) in a
4 spectrophotometer (UV-VIS Shimadzu) using 5 cm quartz cuvette and Milli-Q water as a
5 reference. The absorption spectra of each sample was performed in triplicate and determined
6 using the Spectrum Pack Program software (Shimadzu Inc.). The absorption coefficients were
7 derived from the absorption spectra (A) at 440 nm (m^{-1}) according to the equation 2 (Kirk,
8 1994):

$$9 \quad a_{\text{CDOM}(440)} = 2.303A(440) l^{-1} \quad (\text{equation 2})$$

10 Where l is the optical path of the cuvette (in meters). Absorption coefficients were corrected
11 for backscattering by subtracting the value of the coefficient at 700 nm from the absorbance at
12 440 nm. The spectrophotometric absorption coefficient at 440 nm was used as an index of
13 CDOM concentration (Williamson *et al.* 1996).

14 We used the software Lake Analyzer (Read *et al.* 2011) to determine the depth of the upper
15 mixed layer (Z_{mix}) and the water column stability (Schmidt Stability-St) from monthly
16 temperature profiles. Mean available light in the upper mixed layer (E_{mean} ; $\text{mol photons m}^{-2} \text{ d}^{-1}$)
17 was estimated from surface light, light attenuation and mixing depth as described in Staehr
18 and Sand-Jensen (2007). Accumulated rainfall recorded at every three hours was obtained
19 from a meteorological station located less than 2 km from the lake
20 (<http://sinda.crn2.inpe.br/PCD/>).

21 Unfiltered water samples were taken for analysis of total nitrogen (TOC Analyzer, Shimadzu
22 TOC – 5000) and total phosphorus (according to Mackereth *et al.* 1978).

23 *Partitioning of the diffuse PAR attenuation coefficient*

1 The partition of K_d was performed for the spatial (integrating the spatial data from four
2 seasons sampling in 2013) and temporal data (monthly sampling at site 8 in 2011 and 2012)
3 separately, but using the same equations.

4 The attenuation of the PAR radiation can be partitioned according to the following equation 3
5 (Yamaguchi et al. 2013):

$$6 \quad K_d = K_W + K_{CDOM} + K_{TSM} \quad (\text{equation 3})$$

7 where K_W , K_{CDOM} and K_{TSM} are the partial attenuation coefficient due to respectively pure
8 water, $a_{CDOM}(440)$ and TSM in the PAR range (400 to 700 nm).

9 The light attenuation by the pure water (K_W) was assumed to be constant at 0.027 m^{-1} (Smith
10 and Baker 1978). The PAR attenuation by CDOM (K_{CDOM}) was estimated according to the
11 equation 4 (Pfannkuche 2002, Lund-Hansen 2004, Zhang et al. 2007a):

$$12 \quad K_{CDOM} = 0.221 * a_{CDOM}(440) \quad (\text{equation 4})$$

13 Where, $a_{CDOM}(440)$ is the value of the absorption coefficient measured at 440 nm and 0.221 is
14 the specific absorption coefficient of CDOM at 440 nm.

15 The light attenuation of TSM can be subdivided according to the equation 5:

$$16 \quad K_{TSM} = K_{Chl-a} + K_{TRI} \quad (\text{equation 5})$$

17 Where K_{Chl-a} and K_{TRI} are the partial attenuation coefficients of phytoplankton (as measured
18 by Chl- a) and tripton, respectively. To calculate the K_{TSM} we subtracted the sum ($K_W +$
19 K_{CDOM}) from K_d (Yamaguchi et al. 2013).

20 A simple linear regression analysis was performed between the K_{TSM} (m^{-1}) and the Chl- a
21 concentration ($\mu\text{g L}^{-1}$) to estimate the specific attenuation coefficient of the Chl- a , according
22 to equation 6 (Yamaguchi et al. 2013):

1 $K_{TSM} = b * [Chl-a] + c$ (equation 6)

2 Where the slope b is the specific attenuation coefficient of Chl- a (K^*_{Chl-a}) and c is the
3 intercept. The Chl- a absorption coefficient (K_{Chl-a} ; m^{-1}) was obtained by multiplying K^*_{Chl-a}
4 with [Chl- a]. The K_{TRI} (light absorption by non-pigmented particulate matter) was finally
5 obtained according to equation 5 by subtracting K_{Chl-a} from K_{TSM} (Yamaguchi et al. 2013).

6 Following the model in equation 1 is possible to calculate the relative contributions (%) of
7 each component in the total Kd (100%).

8 *Statistical Analysis*

9 Kruskal-Wallis (One Way Analysis of Variance) and pair wise test (Tukey Test) were
10 performed to assess seasonal differences in Kd values. Simple and multiple linear regressions
11 were made between concentrations of optically active components and partial and total
12 coefficients of PAR attenuation. To demonstrate the spatial distribution of Kd and the optical
13 components maps were made using a Kriging interpolation technique with Surfer 10 software.

14 **Results**

15 *Overall patterns in lake physics and water chemistry*

16 Periods with full mixing occurred between June and August in all three years. The water
17 surface temperature ranged from 21 to 31 °C with lowest temperatures during the mixing
18 period. The photic zone (Z_{eu}) ranged from 3.9 to 14.3 m, reaching greater depths during the
19 stratified rainy periods. Greater stability of the water column (Schmidt) was accordingly
20 observed in the rainy season (mean Schmidt stability = 386.4 J m^{-2}). The deep water
21 temperature remained lower than the surface temperature during the entire rain period and
22 was stable throughout the year (annual average 23.5°C \pm 1°C standard deviation). The depth

1 of the mixing layer was greater in dry and colder period reached a maximum of 27.9 m, while
2 during the warmer rainy season Z_{mix} reached a maximum of 13.4 m (Table 1 and Fig. 2).
3 DOC occurred in higher concentrations during the onset of the rains (December, spatial
4 average 6.5 mg L^{-1}) and mixing in winter (spatial average 6.5 and monthly average 5.9 mg L^{-1}).
5 The DOC concentration in the deeper hypolimnetic waters was higher during thermal
6 stratification (4.9 in November and 5.1 mg L^{-1} in December 2012) than in the upper mixed
7 surface waters (2.1 mg L^{-1} in November and 1.5 mg L^{-1} in December 2012) and elevated DOC
8 levels were accordingly stored in the hypolimnion until the onset of mixing in June. Total
9 nitrogen (TN) occurred in higher concentrations during the mixed dry period (spatial average
10 357.6 and monthly average $446.6 \text{ } \mu\text{g L}^{-1}$) compared to conditions in the upper mixed layer
11 during stratification (averages between 276.8 and $351.7 \text{ } \mu\text{g L}^{-1}$). The total phosphorus (TP)
12 occurred in higher concentrations during winter, but only for the spatial sampling (average
13 $12.8 \text{ } \mu\text{g L}^{-1}$), and no difference was observed between dry and rainy in the monthly sample
14 (Table 1 and Fig. 2a-h).

15 *Spatial variations in optical properties*

16 The coefficients of variation were used to estimate the variability of Kd and optically active
17 components at each site. The spatial distribution of these coefficients of variation and mean
18 values for each parameter in the four sampling periods are shown in Fig. 3 and 4. The largest
19 variation observed for Kd occurred at sites 7 (CV = 0.38), 11 (CV = 0.43) and 15 (CV =
20 0.35), while the lowest occurred at sites 3 (CV = 0.10), 18 (CV = 0.14) and 21 (CV = 0.13).
21 Chl-a presented coefficients of variation between 0.20 (site 12) and 0.83 (site 9), tripton
22 presented between 0.08 (site 6) and 0.80 (site 1) and TSM ranged from 0.05 (site 7) to 0.56
23 (site 21). The highest variation occurred to $a_{\text{CDOM}}(440)$ ranging from 0.17 (site 21) to 2.00
24 (sites 5 and 7). There was no significant correlation ($p > 0.05$) among the coefficients of

1 variation of all parameters measured or between the coefficients and the depth of the sites or
2 proximity to shore.

3 The spatial variation of the Kd was also different for each sampling period, ranging from 15
4 to 18% (measured by the coefficient of variation - CV = standard deviation / average) among
5 sites in the spring (December - CV = 0.15) and summer (January - CV = 0.18), and less than
6 8% in autumn and winter (April - CV = 0.08; August - CV = 0.06). The spatial variation for
7 Chl-*a* was higher in summer (56%) and less than 30% in other sampling periods (April - CV
8 = 0.26; August - CV = 0.30; December - CV = 0.29) , while the variation of tripton was
9 higher in winter (45%) and close at other periods (January - CV = 0.36; April - CV = 0.31;
10 December - CV = 0.30). $a_{CDOM}(440)$ presented larger variation than other optically active
11 components with a minimum of 64% in summer and between 85 and 89% at other periods
12 (April - CV = 0.85; August - CV = 0.89; December - CV = 0.88).

13 We found a significant variation in Kd between sampling seasons (Kruskal-Wallis One Way
14 Analysis of Variance, $p < 0.001$), except between January (summer) and December 2013
15 (Tukey Test, $p > 0.05$). The Kd ranged from 0.32 m^{-1} (station P11 during the summer) to 0.91
16 m^{-1} (station P11 during the winter) during the entire period (average 0.60 m^{-1}). Higher values
17 of Kd (average 0.78 m^{-1}) were observed during the dry winter period, characterized by a fully
18 mixed water column, while lower values (average 0.46 m^{-1}) were observed during the warm,
19 thermally stratified rainy season (January and December) (Table 1 and Fig. 3). The
20 coefficients of determination (R^2) for the calculation of Kd from light vs depth were always
21 higher than 0.9 which indicates a vertically homogeneous water column in relation to the
22 optically active components (Zhang et al. 2007), even during the stratified period.

23 The Chl-*a* concentrations varied 20 fold between 0.5 (station P15 in the summer) and $11.0 \mu\text{g}$
24 L^{-1} (station 8 in the intermediate period in April) (average $4.5 \mu\text{g L}^{-1}$). In comparison the TSM

1 concentrations varied 8 fold and ranged from 0.4 (station P21 in the summer) to 2.8 mg.L⁻¹
2 (station P21 in winter) (average 1.6 mg L⁻¹). Higher Chl-*a* and TSM concentrations were
3 observed in April and winter. The tripton concentration ranged from 0.1 (station P8 in the
4 winter) to 2.4 mg L⁻¹ (station P21 in winter) (average 1.1 mg L⁻¹). Although tripton
5 concentrations reached similar levels as TSM, the spatial variation in tripton was much higher
6 (26 fold), with the highest tripton concentrations also observed during winter where water
7 column was fully mixed. The a_{CDOM}(440) varied from 0 to 0.4 m⁻¹, average 0.1 m⁻¹. Higher
8 a_{CDOM}(440) concentrations were observed in the summer rainy period (Table 1).

9 Using the additive model (equation 1) tripton was found to contribute most to the attenuation
10 of visible light all year round and at all sampling stations during spatial sampling (minimum
11 50%, maximum 91% and average 79%). The relative contribution of phytoplankton pigments
12 estimated from Chl-*a* showed a minimum of 2%, maximum 30.2% and average 11.6% for
13 both seasonal and spatial variations. The a_{CDOM}(440) varied from a minimum contribution of
14 0% to a maximum of 15.6%, and average of 4.8%. The pure water contribution ranged
15 between 2.9% and 8.4%, with an average of 4.7% (Fig. 5). With increasing tripton
16 concentration and its relative contribution to light attenuation, *Kd* also increased. In
17 comparison the contributions of Chl-*a* and a_{CDOM}(440) to PAR attenuation were lower and
18 tended to decrease with higher *Kd* values (Fig. 6). Since the absorption of pure water is
19 constant its relative contribution to *Kd* decreased with increasing attenuation by the other
20 optical components. The specific attenuation coefficients for each optically active component
21 are shown in Table 2.

22 Considering the four sampling events as replicates we did not observe significant relations
23 among the spatial distribution of the concentration of optically active components (Pearson's
24 correlations between Chl-*a* and tripton (R = -0.22, p = 0.33); Chl-*a* and a_{CDOM}(440) (R = -
25 0.10, p = 0.65); tripton and a_{CDOM}(440) (R = -0.16, p = 0.47)) (Fig. 6). The spatial distribution

1 of Kd was significantly correlated only with Chl- a ($R = 0.31$, $p < 0.01$) and we did not
2 observe significant correlations between Kd and tripton ($R = 0.13$, $p = 0.24$) or $a_{CDOM(440)}$ (R
3 $= -0.04$, $p = 0.67$).

4 *Monthly trends in optical properties*

5 From our monthly sampling we found that Kd ranged from a minimum of 0.4 m^{-1} (November
6 2011) to a maximum of 1.3 m^{-1} (July 2012) around an average of 0.8 m^{-1} . Highest values were
7 as previously described observed in the dry period with full mixing of the water column
8 (average 0.9 m^{-1}), while in the warm stratified rain period had a lower average of 0.7 m^{-1}
9 (Table 1 and Fig. 2-h).

10 The Chl- a concentrations ranged 16 fold (min December 0.6 ; max August $8.9 \mu\text{gL}^{-1}$) with an
11 average of $4.5 \mu\text{gL}^{-1}$. Higher TSM concentrations occurred during dry periods and the
12 concentrations ranged only 3.5 fold from 0.9 (April 2012) to 3.2 (June 2011) (average of 1.9
13 mg L^{-1}). Higher tripton concentrations were observed in dry periods (June and July 2011 and
14 August 2012) and in two months of rain period (January and November 2012), and ranged 8
15 fold (average 1.5 mg L^{-1}). The $a_{CDOM(440)}$ absorption ranged from 0.1 (November 2012) to
16 0.6 (July 2011) (average 0.3 m^{-1}). Overall higher concentration of Chl- a , TSM, tripton and
17 $a_{CDOM(440)}$ occurred during the dry season of 2011 and 2012. The $a_{CDOM(440)}$ absorption
18 ranged from 0.1 (November 2012) to 0.6 (July 2011) (average 0.3 m^{-1}). There was a
19 noticeable fall in the concentration of DOC and $a_{CDOM(440)}$ from June 2011 to December
20 2012 (Table 1 and Fig. 2-e,f,g).

21 For the monthly and annual variations observed at the station P8 tripton was most
22 predominant in June, July, November and December 2011, and January and December 2012.
23 The contribution of tripton attenuation varied from a minimum of 18.3% (May 2012) to
24 maximum of 77.2% (December 2011) and an overall average of 47.8 % throughout the

1 sampling period. In October 2011 and March to August 2012, Chl-*a* was the primary
2 contributor to *K_d*. Throughout the period the Chl-*a* showed a minimum contribution of 6.8%,
3 maximum of 65.3% and an average of 35.6%. In June 2012, Chl-*a* and tripton contributed
4 almost equally with 45.6% and 45.5%, respectively. We observed that the Chl-*a* contribution
5 increased in the dry mixing periods. The $a_{\text{CDOM}(440)}$ contribution to *K_d* ranged between
6 3.4% and 22% with an average of 12%. Pure water contributed with only 4.6% on average
7 (Table 2 and Fig. 7).

8 Using a multiple regression analysis we found that K_{Chl} and K_{TRIP} together explained 97% of
9 *K_d* ($F(2,9) = 139.832$, $p < 0.001$, $r = 0.98$, $R^2 = 0.97$). The regression coefficient for Chl-*a*
10 (0.96) was higher than for tripton (0.92), which means that while total PAR attenuation
11 depended on a combination of optical conditions, the single most important component
12 driving temporal variation in *K_d* was Chl-*a* (Fig. 8).

13 From the analysis of correlations between optical components and physical and chemical
14 conditions characterizing changes at the central station in the lake we found significant
15 correlations ($p < 0.05$) between: 1) *K_d* and Chl-*a* ($R = 0.66$) and rainfall ($R = -0.58$); 2) TSM
16 and tripton ($R = 0.92$), Z_{mix} ($R = 0.71$) and Schmidt Stability ($R = -0.61$); 3) Schmidt Stability
17 and E_0 ($R = 0.74$) and E_{mean} ($R = 0.71$); 4) E_0 and Z_{mix} ($R = 0.61$) and DOC ($R = -0.49$); 5)
18 tripton and Z_{mix} ($R = 0.62$); 6) Z_{eu} and rainfall ($R = 0.62$); 7) $a_{\text{CDOM}(440)}$ and DOC ($R =$
19 0.60); and 8) Chl-*a* and rainfall ($R = -0.59$). All correlations of these parameters are shown in
20 Table 3. These correlations indicate that the seasonal changes in optical components affecting
21 PAR attenuation are strongly linked to meteorological conditions as these affect water column
22 stability and mixing conditions. The combination of these couplings are finally shown as
23 responses in mean light available and the depth of the photic zone.

24 Discussion

1 *Spatial sampling at 21 stations*

2 Considering the coefficients of variation, the optically active components and K_d varied
3 among the 21 sampled sites, but these spatial variability were not significant related among
4 them or explained by the sites depth or proximity with the lake shore as expected. Some deep
5 stations (> 10 m) accordingly had variation in K_d values similar to stations located near the
6 shore suggesting that other parameters such as complex internal movement of water driven by
7 wind may control the spatial heterogeneity in PAR attenuation (Somlyódy and Koncsos
8 1991). However, although there wasn't relation between the coefficients of variation of K_d
9 and the coefficients of variation of the other parameters, we found a significant and positive
10 correlation between K_d values and Chl-*a* concentrations analyzing all sites and sampling
11 periods together. In this way, although tripton had contributed more with the total PAR
12 attenuation in all sites and seasons, the Chl-*a* concentration was the only parameter correlated
13 with the variation of K_d .

14 In addition to the spatial variability among sites, there were distinct spatial variability for Chl-
15 *a*, tripton and $a_{CDOM(440)}$ also among sampling periods. During the dry and colder winter
16 mixing period elevated levels of particulate matter (phytoplankton and tripton) caused overall
17 higher PAR attenuation. However, the spatial variability of K_d in this period was lower (6%)
18 than that observed in the summer (18%) and spring (15%), although the spatial variability of
19 Chl-*a* and tripton in winter was 30% and 45%, respectively. It has previously been
20 demonstrated that mixing of the water column causes resuspension of particulate matter
21 increasing turbidity and tripton concentrations (Blom et al. 1994, James et al. 1997, 2004,
22 Philips et al. 1995, Zhang et al. 2007), which consequently would explain the higher values of
23 K_d and the highest average contribution of tripton (83.9%) during the mixed dry season.
24 Higher lake transparency (lower values of K_d) occurred in summer and spring, when
25 concentrations of particulate matter were lower, but the $a_{CDOM(440)}$ higher and its

1 contribution in K_d became more apparent in the sites (Fig. 5a,d). The higher spatial variability
2 of K_d (15-18%) was observed in periods of more transparent waters (spring and summer),
3 which coincided with the highest spatial variability of Chl- a during the summer (56%).
4 Despite the increase in Chl- a and $a_{CDOM(440)}$ contribution to K_d during summer, tripton was
5 largely responsible for the attenuation of visible light when comparing all sites and sampling
6 periods. Moreover, tripton contribution to the attenuation increased with increasing K_d value
7 (see Fig. 6). The relationship between reduced Chl- a and $a_{CDOM(440)}$ contributions and
8 increasing tripton contribution with the increase of K_d was also demonstrated by Zhang et al.
9 (2007). Other studies have similarly found a strong positive relationship between tripton and
10 K_d and a great contribution of this component to PAR attenuation in lakes (Phlips et al. 1995,
11 Christian and Sheng 2003, Yang et al. 2005, Zhang et al. 2007, Balogh et al. 2009) and
12 estuaries and oceans (McMahon et al. 1992, Lin et al. 2009, Yamaguchi et al. 2013) (See
13 Table 4). In this way, the main component responsible for the PAR attenuation was tripton,
14 followed by Chl- a , which is in agreement with our first hypothesis since most of the
15 variability in K_d was due to the particulate matter.

16 The low contribution of $a_{CDOM(440)}$ to K_d (maximum 15.6%) was expected since the
17 $a_{CDOM(440)}$ occurred at low concentrations in this lake and its absorption decays with
18 increasing wavelength (Green and Blough, 1994, Kitidis et al. 2006), implying that the overall
19 contribution of $a_{CDOM(440)}$ absorption is higher in the ultraviolet part of the spectra (200-400
20 nm).

21 *Monthly sampling at the deep central station*

22 In the monthly sampling at the deep central station 8 (between 2011 and 2012), the
23 contribution of Chl- a for the total light attenuation was more noticeable and higher than the
24 tripton contribution in some periods (October-2011 and from March to August-2012). The

1 majority of studies, predominantly in coastal waters, showed that the relative contribution of
2 tripton to the PAR attenuation was always higher than that of Chl-*a*. This has been
3 documented in Aarhus Bay, Kattegat (Lund-Hansen, 2004), Florida Bay (Phlips et al. 1995,
4 Kelble et al. 2005), Tampa Bay and Charlotte Harbor (McPherson and Miller 1987,
5 McPherson and Miller 1994) and in the Yellow Sea (Lin et al. 2009). While tripton is often
6 dominate light attenuation, Zhang et al. (2007b) showed that increases in Chl-*a* during
7 summer algal blooms makes light absorption by pigments a more dominant optical
8 component than tripton in some regions of Taihu Lake (China). In hypereutrophic lakes
9 Balogh et al. (2009) also observed higher Chl-*a* contribution to the PAR attenuation and Yang
10 et al. (2005) in some regions of the Lake Donghu (China). Effler et al. (2010) in contrast,
11 found that $a_{CDOM(440)}$ was the primarily factor responsible for PAR attenuation in Lake
12 Superior (USA). In our study, Chl-*a* and tripton together explained 97% (R^2 , multiple linear
13 regression) of the temporal variation of K_d but the Chl-*a* was the most responsible. Similar
14 results were found by Yamaguchi et al. (2013) where although tripton have been the
15 predominant component responsible for the K_d in percentage of attenuation, Chl-*a* was the
16 main component that influenced the K_d monthly variation in Harima Sound ($R^2 = 75\%$)
17 (Eastern Seto Inland Sea, Japan). Obrador and Pretus (2008) also demonstrated that the main
18 driver for seasonal changes in K_d was phytoplankton (Chl-*a*) despite that DOC had a higher
19 relative contribution. Reinforcing this conclusion, we found a smaller temporal variation in
20 tripton concentration (coefficient of variation: $CV = 0.51.$) than in Chl-*a* ($CV = 0.65$), which
21 suggests that although tripton contributed significantly to light attenuation, variations in Chl-*a*
22 was the main driver for seasonal fluctuations in K_d values. This finding allows us to conclude
23 that the main factor that attenuates light is not necessarily the same one that causes temporal
24 variation of the light attenuation (Obrador and Pretus 2008) (see Table 4).

1 From our monthly sampling we found that higher irradiance and rainfall values in summer
2 were positively related to the mean light available to primary production (E_{mean}) and Schmidt
3 stability while negatively related to the mixed layer depth (Z_{mix}). In this period, the DOC and
4 $a_{\text{CDOM}}(440)$ occurred in low concentrations in the photic zone suggesting high photobleaching
5 caused by the high incident radiation. It is known that with rain allochthonous carbon enters
6 the deeper layers of the lake and remains in the hypolimnion until mixing period, when it
7 becomes distributed in the water column (Reynolds 2009). Corroborating this finding we
8 observed higher DOC concentration at the bottom of the lake than at the subsurface in the
9 rainy period (November and December 2012). With a lower incidence of solar radiation and
10 cooling of surface waters during winter the depth of the mixed layer increased, causing a
11 more homogeneous distribution of nutrients and particulate matter in the water column
12 (Barbosa and Tundisi 1980, Petrucio and Barbosa 2004, Brighenti et al. 2015). This would
13 explain the observed increase in Chl-*a* concentrations in the full mixing period in 2012 and its
14 higher relative contribution in PAR attenuation (maximum 65% in May 2012). Brighenti et al.
15 (2015) investigating the patterns of primary production in Dom Helvécio lake during 2011 to
16 2012 demonstrated that during thermally stratified period, with higher surface light and lower
17 light attenuation, coincided with a higher occurrence of photoinhibition of phytoplankton
18 productivity. Also these authors found that the availability of nutrients and the highest
19 phytoplankton biomass coincided with periods of lower photoinhibition (mixing periods).
20 Thus, the increase of Chl-*a* combined with the increase of tripton concentrations in mixing
21 period resulted in higher values of K_d at this time.

22 *Conclusions*

23 The present study demonstrated that is important to link temporal and spatial sampling in
24 order to understand the role of the optical components influencing PAR attenuation, because
25 the main component that attenuates light is not necessarily the same component responsible

1 for the temporal variation of K_d . Optimally this should be combined with a high-frequency
2 sampling of physical conditions characterizing water column mixing conditions. This
3 combination allows better certainty of the importance of each optical component for the total
4 attenuation of light and a better mechanistic understanding of the physical and chemical
5 conditions driving temporal changes. Particularly, we found that there was a significant
6 temporal and spatial variability in water transparency, and although most of PAR attenuation
7 occurred due to tripton concentrations, the variability of K_d was mainly associated to variation
8 in the concentration of Chl-*a*. However, we did not find support for our expectation of higher
9 variability in optical condition at near shore sites, suggesting that new studies on the
10 variability in K_d must consider the hydrodynamic conditions in the distribution of optical
11 components within the lake.

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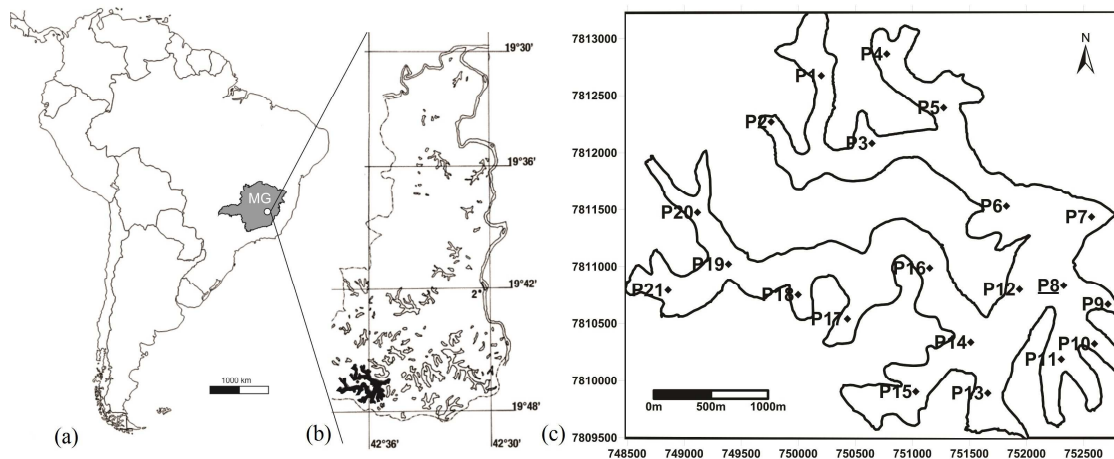
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4 **Figures and Tables**

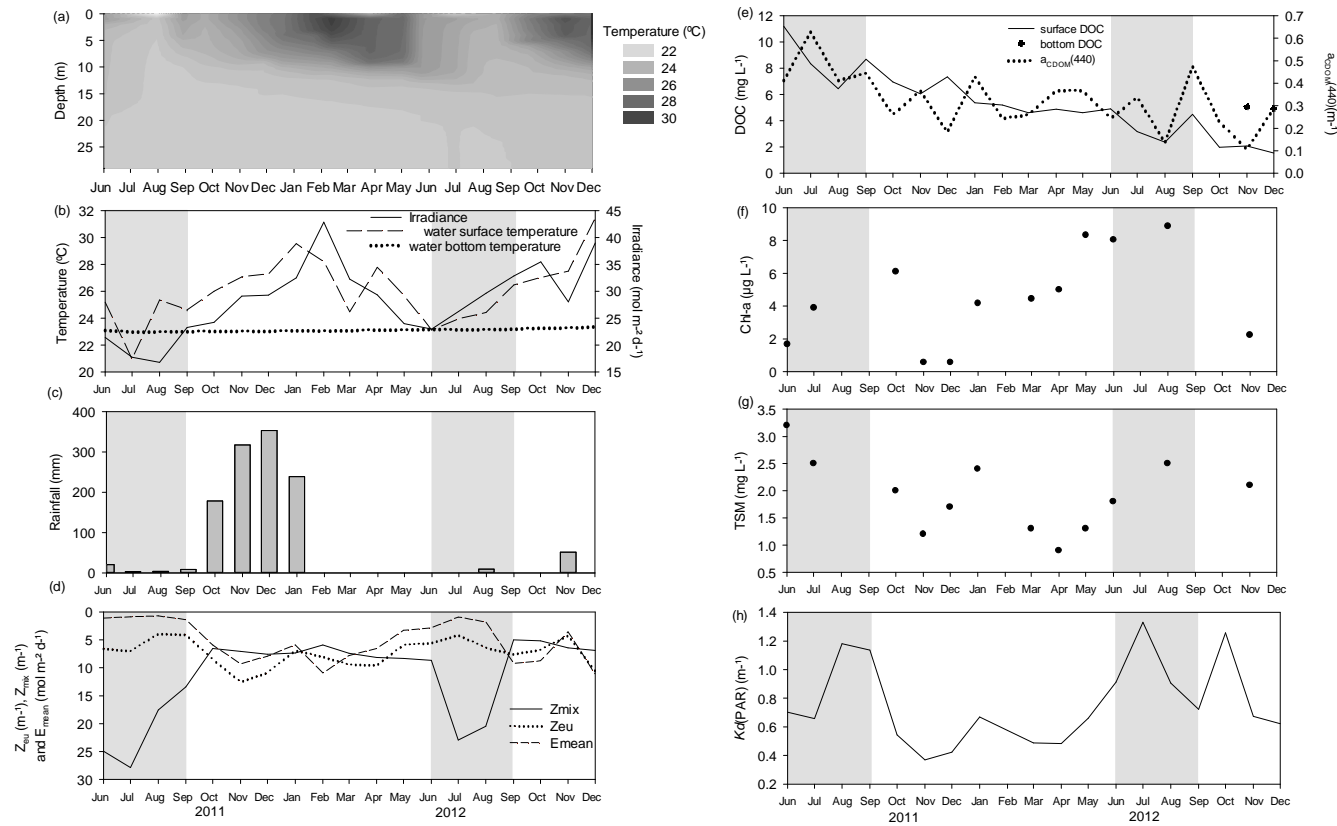


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6 Figure 1 - Location of Dom Helvécio lake in Minas Gerais (a), inside of the Rio Doce State
7 Park (b), Middle Rio Doce and the spatial sampling stations (P1-P21) (c). The station P8 (c;
8 underlined) also represents the location of monthly sampling. The numbers on the axes
9 correspond to geographic coordinates.

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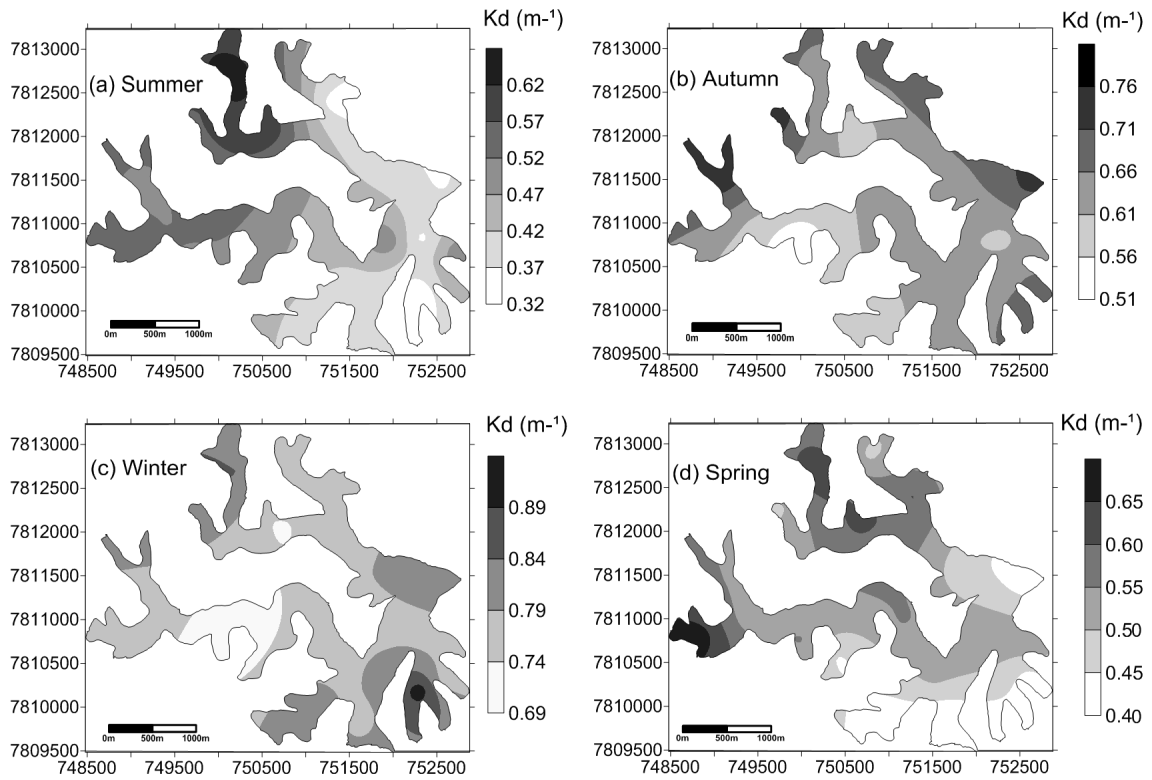
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2 Figure 2 – Monthly variation of the (a) water temperature depth profile, (b) surface irradiance, (c) rainfall, (d) mixing layer depth (Z_{mix}), euphotic
 3 zone depth (Z_{eu}), mean light climate (E_{mean}), (e) dissolved organic carbon, colored dissolved organic matter, (f) chlorophyll-*a*, (g) total suspended
 4 matter and (h) PAR attenuation coefficient ($K_d(PAR)$) from June 2011 to December 2012.



1

2 Figure 3 - Spatial variation of the attenuation coefficient of photosynthetically active
 3 radiation (m^{-1}) in the summer (January 2013 - a), autumn (April 2013 - b), winter
 4 (August 2013 - c) and spring (December 2013 - d). Notice that the scales are not
 5 similar. The numbers on the axes correspond to geographic coordinates.

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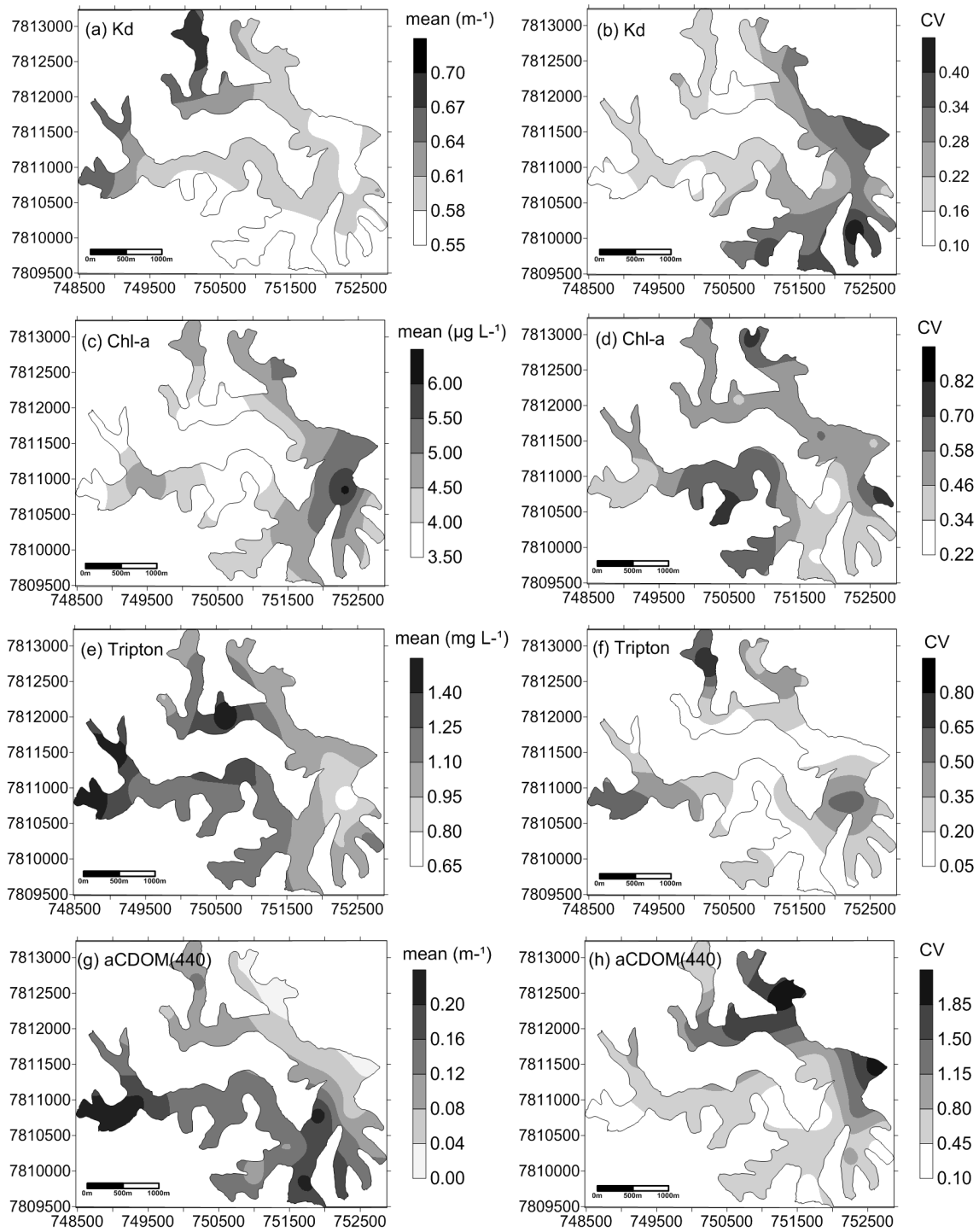
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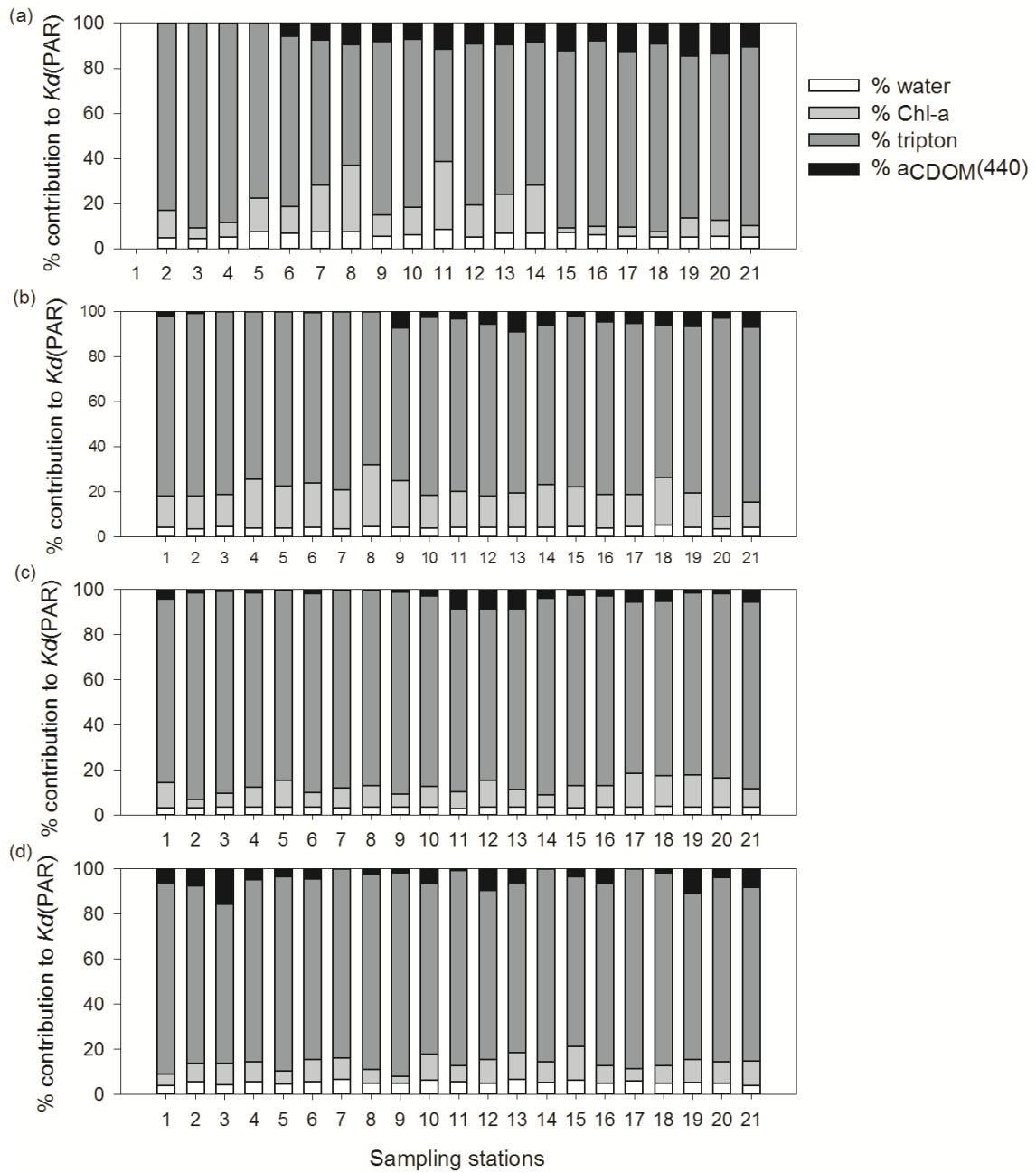


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2 Figure 4 – Mean values (left panel) and coefficients of variation (CV; right panel) of
 3 each sampling site during the four seasons for *Kd* (a,b), *Chl-a* ($\mu g L^{-1}$) (c,d), tripton (mg
 4 L^{-1}) (e,f) and *aCDOM(440)* (m^{-1}) (g,h). The numbers on the axes correspond to geographic
 5 coordinates.

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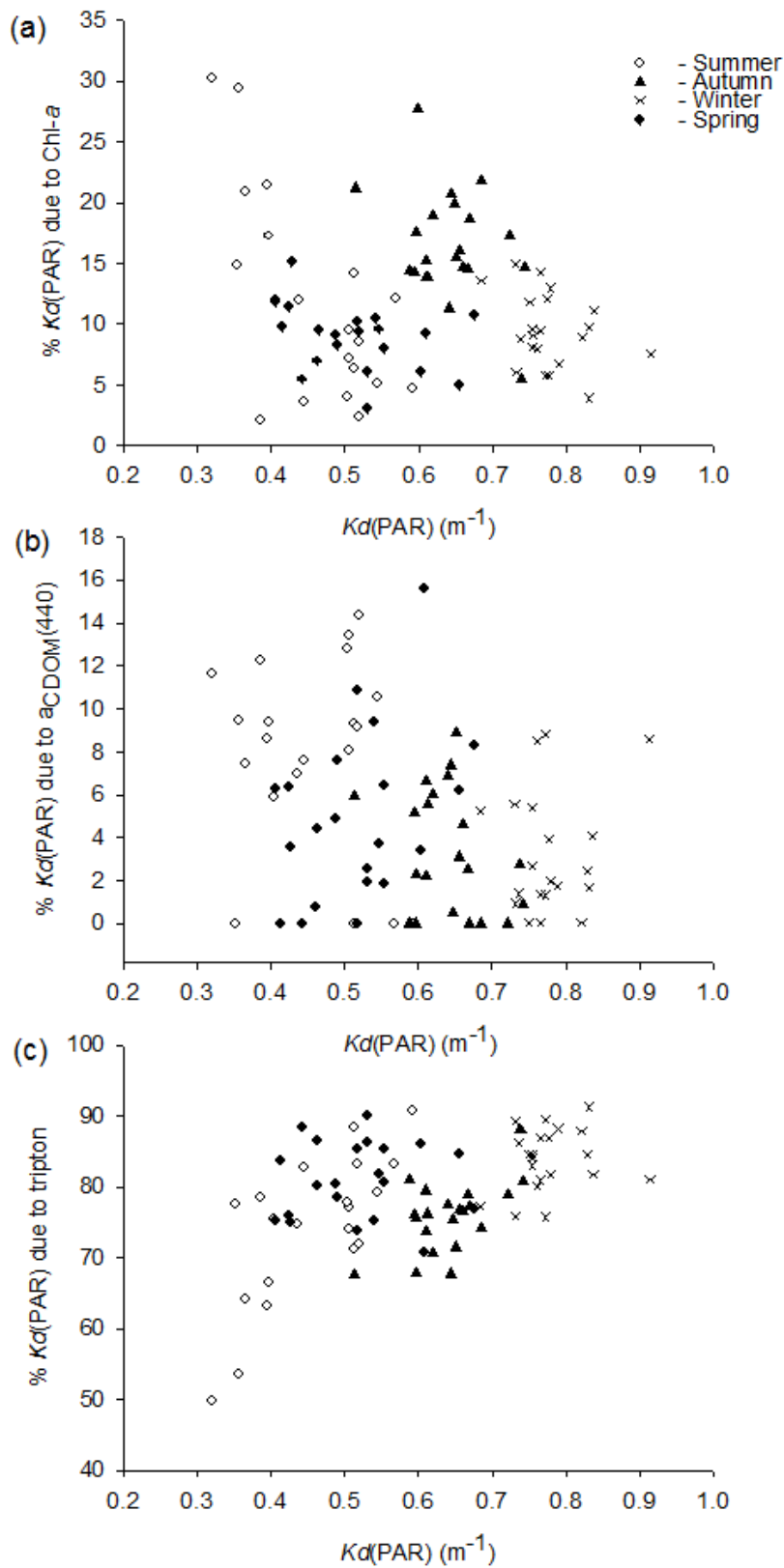
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3 Figure 5 - The contributions of pure water, phytoplankton, tripton and colored dissolved
4 organic matter to $K_d(PAR)$ during (a) summer, (b) autumn, (c) winter and (d) spring.

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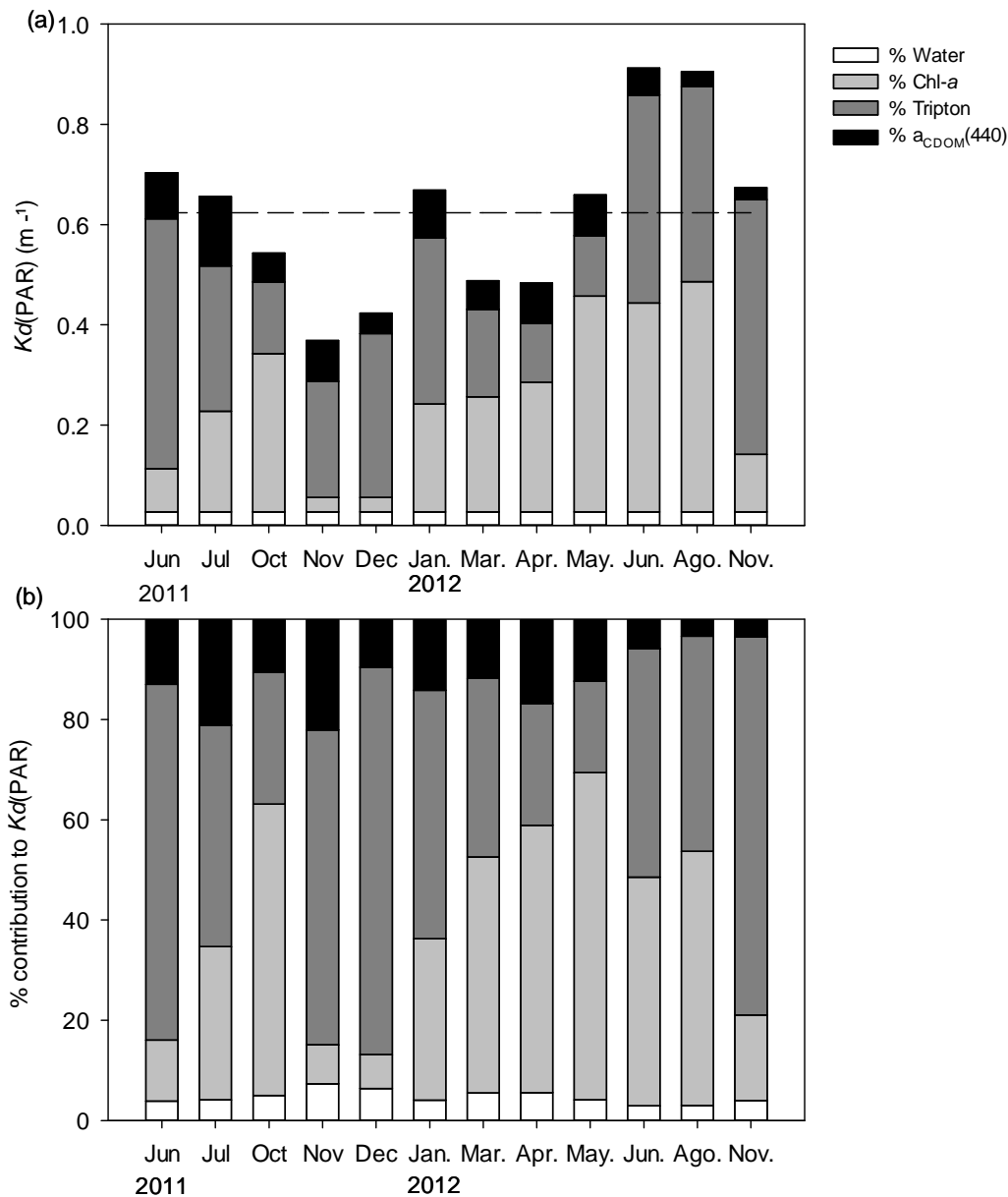
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2 Figure 6 - The relative contributions of Chl-*a* (a), $a_{CDOM}(440)$ (b) and tripton (c) to

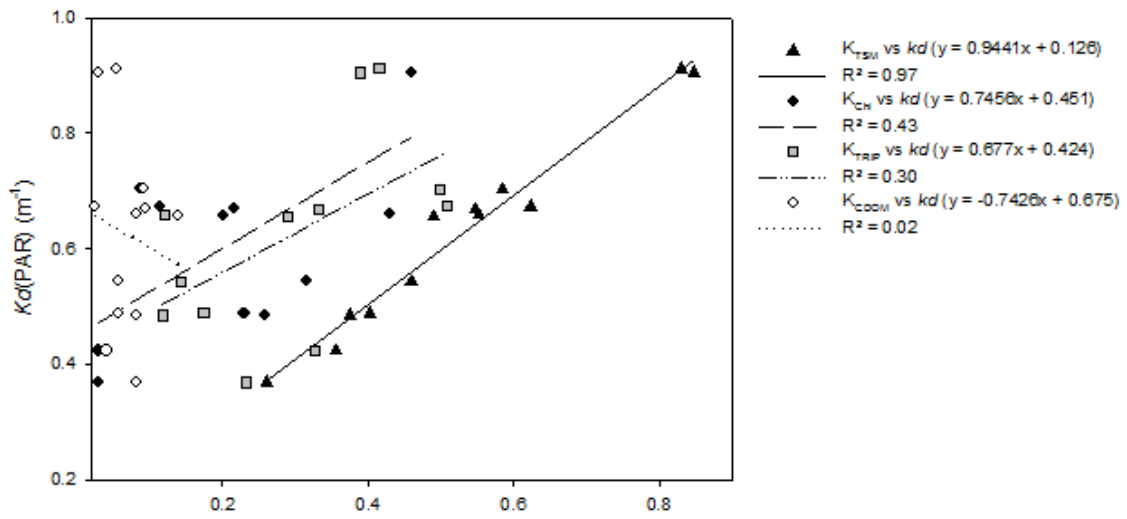
3 $Kd(PAR)$ during the spatial measurements.



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2 Figure 7– (a) Temporal variation of the PAR attenuation coefficient in the water
 3 column..White, light grey, dark grey and black bars show the partial attenuation
 4 coefficients due to pure water (K_w), phytoplankton ($K_{\text{chl-a}}$), tripton (K_{trip}) and
 5 $a_{\text{CDOM}}(440)$ (K_{CDOM}), respectively. The dashed line indicates the average of $K_d(\text{PAR})$
 6 through the study period (0.62 m⁻¹). (b) Percentage contributions of the K_w , $K_{\text{chl-a}}$, K_{trip}
 7 and K_{CDOM} to $K_d(\text{PAR})$ during the same period.

8



1 Specific attenuation coefficients due to TSM, Chl-a, tripton and $a_{CDOM}(440)$ (m^{-1})

2 Figure 8 – Simple linear regressions between $Kd(PAR)$ and partial attenuation
 3 coefficients of visible radiation due to TSM (triangles), tripton (crosses), Chl-a (black
 4 circles) and $a_{CDOM}(440)$ (white circles) during monthly sampling.

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1 Table 1- Summary of parameters measured during temporal and spatial analyzes in the
 2 lake between 2011 and 2013.

		Spatial sampling				Temporal sampling	
		Summer (Jan. 2013)	Autumn (Apr. 2013)	Winter (Aug. 2013)	Spring (Dec. 2013)	Dry season	Rainy season
$K_d(PAR)$ (m^{-1})	Mean	0.46	0.64	0.78	0.52	0.91	0.66
	Min-Max	0.32-0.59	0.51-0.74	0.68-0.91	0.41-0.68	0.66-1.33	0.37-1.26
	std dev	0.08	0.05	0.05	0.08	0.26	0.26
Tripton ($mg L^{-1}$)	Mean	1.03	1.09	1.19	1.14	1.65	1.33
	Min-Max	0.18-1.65	0.68-1.80	0.09-2.37	0.57-2.19	0.46-3.03	0.40-1.88
	std dev	0.37	0.34	0.53	0.34	0.99	0.57
Chl-a ($\mu g L^{-1}$)	Mean	3.31	6.98	4.81	2.94	6.15	3.28
	Min-Max	0.47-6.90	2.68-10.91	2.10-7.22	1.13-4.80	1.69-8.91	0.59-6.11
	std dev	1.86	1.82	1.47	0.87	3.20	2.20
$a_{CDOM(440)}$ (m^{-1})	Mean	0.16	0.10	0.11	0.12	0.36	0.28
	Min-Max	0.00-0.30	0.00-0.30	0.00-0.40	0.00-0.40	0.10-0.60	0.10-0.40
	std dev	0.10	0.08	0.10	0.11	0.19	0.11
TSM ($mg L^{-1}$)	Mean	1.36	1.79	1.67	1.43	2.26	1.66
	Min-Max	0.36-2.13	1.28-2.28	0.57-2.77	0.91-2.40	1.29-3.20	0.90-2.42
	std dev	0.39	0.24	0.55	0.35	0.73	0.54
DOC ($mg L^{-1}$)	Mean	4.30	3.09	6.46	6.47	5.87	5.24
	Min-Max	3.31-5.05	2.49-3.46	5.25-8.32	4.10-8.47	2.38-11.21	1.92-8.73
	std dev	0.44	0.29	0.98	0.89	3.08	2.04
Mixing layer (m)	Mean	-	-	-	-	17.73	7.25
	Min-Max	-	-	-	-	8.40-27.90	5.00-13.40
	std dev	-	-	-	-	7.61	2.16
Euphotic zone (m)	Mean	10.37	7.22	5.94	9.08	5.65	8.22
	Min-Max	7.80-14.30	6.20-8.90	5.00-6.70	6.80-11.30	3.90-7.00	4.10-12.50
	std dev	1.91	0.62	0.36	1.31	1.12	2.55
Surface Temperature ($^{\circ}C$)	Mean	30.00	28.73	25.40	31.00	24.27	27.41
	Min-Max	-	-	-	-	20.90-25.70	24.50-31.40
	std dev	-	-	-	-	1.58	1.89
Water column depth (m)	Mean	14.03	-	13.04	-	28.50	28.50
	Min-Max	4.50-30.00	-	2.90-30.00	-	27.00-30.00	27.00-30.00
	std dev	8.52	-	8.58	-	0.84	0.84
Total nitrogen ($\mu g L^{-1}$)	Mean	276.81	278.53	357.60	125.14	446.61	351.71
	Min-Max	215.49-424.10	242.62-342.10	12.73-519.71	3.68-431.47	292.21-688.20	229.27-695.29
	std dev	41.03	27.05	146.42	158.91	127.29	128.15
Total phosphorus ($\mu g L^{-1}$)	Mean	3.78	-	12.75	8.27	8.06	8.22
	Min-Max	2.90-5.61	-	5.69-21.77	4.80-11.90	5.46-10.74	4.60-12.82
	std dev	0.97	-	4.10	1.83	2.09	2.42
Schmidt stability ($J m^2$)	Mean	-	-	-	-	124.16	386.43
	Min-Max	-	-	-	-	31.90-282.90	142.30-621.90
	std dev	-	-	-	-	82.72	167.89

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- 1 Table 2- The specific attenuation coefficients of Chl-a, tripton, CDOM, pure water and
 2 TSM to the K_d and their relative contribution (%) to the total PAR attenuation.

		Specific attenuation coefficients (m ⁻¹)					Contribution to PAR attenuation (%)				
Monthly sampling	$K_d(PAR)$	K_{Chl-a}	K_{TRI}	K_{CDOM}	K_W	K_{TSM}	K_{Chl-a}	K_{TRI}	K_{CDOM}	K_W	
June-11	0.70	0.09	0.50	0.09	0.03	0.58	12	71	13	4	
July-11	0.66	0.20	0.29	0.14	0.03	0.49	31	44	21	4	
October-11	0.54	0.32	0.14	0.06	0.03	0.46	58	26	11	5	
November-11	0.37	0.03	0.23	0.08	0.03	0.26	8	63	22	7	
December-11	0.42	0.03	0.33	0.04	0.03	0.36	7	77	10	6	
January-12	0.67	0.22	0.33	0.10	0.03	0.55	32	50	14	4	
March-12	0.49	0.23	0.17	0.06	0.03	0.40	47	36	12	6	
April-12	0.48	0.26	0.12	0.08	0.03	0.38	53	24	17	6	
May-12	0.66	0.43	0.12	0.08	0.03	0.55	65	18	12	4	
June-12	0.91	0.42	0.42	0.05	0.03	0.83	46	45	6	3	
August-12	0.91	0.46	0.39	0.03	0.03	0.85	51	43	3	3	
November-12	0.67	0.11	0.51	0.02	0.03	0.62	17	75	4	4	
Spatial sampling	$K_d(PAR)$	K_{Chl-a}	K_{TRI}	K_{CDOM}	K_W	K_{TSM}	K_{Chl-a}	K_{TRI}	K_{CDOM}	K_W	
January-2013	Mean	0.46	0.05	0.35	0.04	0.03	0.40	12	74	8	6
	min	0.32	0.01	0.16	0.00	0.03	0.26	2	50	0	5
	max	0.59	0.10	0.54	0.07	0.03	0.57	30	91	14	8
	SD	0.08	0.03	0.10	0.02	0.00	0.08	8	11	5	1
April-2013	Mean	0.64	0.11	0.49	0.02	0.03	0.59	17	76	3	4
	min	0.51	0.04	0.35	0.00	0.03	0.46	5	68	0	4
	max	0.74	0.17	0.65	0.06	0.03	0.71	28	88	9	5
	SD	0.05	0.03	0.07	0.02	0.00	0.06	4	5	3	0
August-2013	Mean	0.78	0.07	0.65	0.03	0.03	0.72	9	84	3	3
	min	0.68	0.03	0.53	0.00	0.03	0.62	4	76	0	3
	max	0.91	0.11	0.76	0.08	0.03	0.81	15	91	9	4
	SD	0.05	0.02	0.06	0.02	0.00	0.05	3	4	3	0
December-2013	Mean	0.52	0.04	0.42	0.03	0.03	0.46	9	81	5	5
	min	0.41	0.02	0.31	0.00	0.03	0.35	3	71	0	4
	max	0.68	0.07	0.56	0.10	0.03	0.59	15	90	16	7
	SD	0.08	0.01	0.07	0.02	0.00	0.07	3	5	4	1

3

- 4 Table 3 - Pearson correlations of the parameters obtained in the analysis of temporal
 5 changes at station P8 in 2011 and 2012. Significant correlations are highlighted in bold.
 6 (St = Schmidt stability; E0 = irradiance; Z_{eu} = euphotic zone depth).

	TSM (mg L ⁻¹)	Chl-a (µg L ⁻¹)	Tripton (mg L ⁻¹)	aCDOM(440) (m ⁻¹)	St (J m ²)	E0 (mol photons m ⁻² d ⁻¹)	Emean (mol photons m ⁻² d ⁻¹)	Ze _u (m)	Zmix (m)	DOC (mg L ⁻¹)	Rainfall (mm)
$k_d(PAR)$ (m ⁻¹)	0.5550	*0.656	0.2420	-0.1390	-0.4790	-0.3210	**0.813	***0.828	0.41	-0.238	*0.579
TSM		-0.067	***0.922	0.153	*0.614	-0.405	*0.697	-0.517	**0.705	0.409	-0.122
Chl-a			-0.449	-0.13	-0.173	-0.121	-0.411	-0.485	0.0221	-0.451	*0.586
Tripton				0.188	-0.482	-0.316	-0.464	-0.275	*0.623	0.542	0.119
a _{CDOM} (440)					-0.302	-0.493	-0.204	0.1	0.486	*0.604	-0.0948
St						***0.737	***0.712	0.404	***0.714	-0.331	0.118
E0							***0.822	0.389	**0.614	*0.494	0.107
Emean								***0.718	***0.784	-0.252	0.373
Ze _u									-0.386	0.138	**0.624

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* $p < 0.05$, ** $p < 0.01$, *** $p < 0.001$

8

1 Table 4- Studies published since 2003 about the relative contribution of optically active
 2 components to the attenuation of PAR radiation in a range of aquatic systems.

Environment	Chlorophyll-a %	Tripton %	CDOM %	Water %	Type of Sampling	Reference
coastal lagoon, Florida	10.0-26.0% mean 10.0-26.0%	59-78% mean 61-59%	5-25% mean 25-11%		spatial in 6 sampling events, 2 different methods to partition	Christian, D. & Y. P. Sheng 2003
Aarhus Bay, Kattegat, North Sea-Baltic	mean 32%	mean 42%	mean 17%	mean 9%	time series study at one station	Lund-Hansen 2004
Florida Bay	mean 1-3%	mean 88-89%	mean 1-2%	mean 7-9%	spatial and monthly over 2 years	Kelble et al. 2005
subtropical Lake Donghu, China	49-65%	46-66%	18-24%	1-2%	spatial and monthly over 1 year	Yang et al. 2005
subtropical shallow Lake Taihu, China	2.8-34.6% mean 9.7%	58.1-95.9% mean 82.6%	0.8-22.8 mean 6.8%	0.2-3.1% mean 0.9%	spatial in 1 sampling event	Zhang et al. 2007
subtropical shallow Lake Taihu, China	summer: 8-70% mean 26%; winter: 1-11% mean 5%	summer: 21-65% mean 48%; winter: 38-91% mean 80%	summer: 7-37% mean 16%; winter: 3-50% mean 11%	summer: 2-25% mean 10%; winter: 2-8% mean 4%	spatial in summer and winter	Zhang et al. 2007b
Mediterranean coastal lagoon, Abufera des Grau	mean 44%	mean 6%	mean 47%	mean 3%	spatial and monthly over almost 4 years	Obrador & Pretus, 2008
tropical Lake Victoria, Uganda and Kenya	<50%					Loiselle et al. 2008
30 shallow lakes and ponds, Hungary	56-70% in hypertrophic waters	55-88% in most water bodies	72-97% in highly coloured waters	-	comparison among lakes	Balogh et al. 2009
Sturgeon River, Keweenaw Bay and Lake Superior, USA	maximum 13%	7-13%	mean > 75%	maximum 3%	spatial in 2 sampling events	Effler et al. 2010
Harima Sound, Seto Inland Sea, Japan	mean 27%	mean 45%	9-18% mean 13%	8-21% mean 15%	monthly over 1 year at one point	Yamaguchi et al. 2013
Deep tropical lake, Brazil	spatial: 2-30% mean 12%; monthly: 7-65% mean 36%	spatial: 50-91% mean 79%; monthly: 18-77% mean 48%	spatial: 0-16% mean 5%; monthly: 3-22% mean 12.0%	spatial: 2.9-8.4% mean 4.7%; monthly: 3-7.3% mean 4.6%	spatial in 4 seasons and monthly	this present study

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CAPÍTULO 2

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Temporal and spatial analysis of DOM quality and quantity in a dendritic tropical lake

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Abstract

Dissolved organic matter (DOM) is an important driver of biogeochemical and ecological processes in aquatic environments. In order to verify if there were spatial and temporal differences in DOM quantity and quality we analyzed at 21 sites during 6 campaigns the variability of DOM in a deep tropical natural lake (Lake Dom Helvécio, MG, Brazil). Surface water samples were taken to analyze DOC, Chl-a, suspended solids, nutrients and CDOM spectra ($SUVA_{254}$, $S_{275-295}$, $a_{CDOM440}$ and $a_{CDOM254}$). Depth of sites and rainfall were also used in the analysis. We found that seasonally the differences in DOM were higher than spatially, and there was no clear spatial pattern for DOC. Rainfall correlated negatively with DOC and $a_{CDOM440}$, and positively $SUVA_{254}$ and Chl-a. Aromatic carbon input increased with the increase of rainfall and the strong negative relation between $SUVA_{254}$ and DOC indicates that with the degradation of DOC the molecules became more aromatics, suggesting higher degradation in the PAR range. Negative correlations were observed between depth and $a_{CDOM254}$, probably because aromatic carbon remains restricted to the hypolimnion in deep sites. This study

1 showed that the change in rainfall pattern in the region in the last years has affected the
2 seasonal patterns of input of DOM and nutrients in the lake, with possible impacts on
3 the seasonality of phytoplankton and consequently the ecosystem functioning.

4 Key words: CDOM, DOC, SUVA, spectral slope, organic matter.

5 **1 Introduction**

6 The dissolved organic matter (DOM) is the largest fraction of organic carbon in aquatic
7 ecosystems and easily exceeds the amount of carbon contained in particulate form
8 (POM) (Hedges, 1992; Battin et al., 2008). Therefore, DOM is an important driver of
9 biogeochemical and ecological processes in aquatic environments (Guo et al., 2014)

10 Although the quantity of DOM is important for the carbon cycle within the ecosystem,
11 its role in biogeochemical cycles depends on its quality (source and composition)
12 (Mueller et al., 2012). In turn, the DOM composition depends on its source, which can
13 be allochthonous (terrestrial origin from the degradation of plant and soil) and
14 autochthonous (production within the water column mainly from the excretion,
15 decomposition and lysis of macrophytes and algae) (Thurman, 1985; Wetzel, 2001).
16 The allochthonous DOM is in general more susceptible to photodegradation and absorbs
17 solar radiation mainly in the UV range, because it is composed by aromatic compounds
18 of higher molecular weights (Amon and Benner, 1994; McKnight et al., 1994; Benner,
19 2002; Helms et al., 2008). Autochthonous DOM can be further subdivided into two
20 types with different characteristics: DOM from phytoplankton mainly consists of simple
21 molecules (carbohydrates, proteins, amino acids) of low molecular weight and it is more
22 labile for microbial community (Farjalla et al., 2009; Fonte et al., 2013); and DOM
23 originating from macrophytes can be as aromatic and recalcitrant as DOM from
24 terrestrial vegetation (Catalán et al., 2013). Nowadays, there are useful, rapid, sensitive
25 and non-destructive tools to examine DOM quality, as absorbance and fluorescence
26 spectroscopy (Weishaar et al., 2003; McKnight et al., 2001; Helms et al., 2008), but
27 studies connecting DOM quantity and quality in tropical environments, especially in
28 tropical lakes, are still scarce (Teixeira et al., 2011, 2013; Bittar et al., 2015; Brandão et
29 al., 2016).

30 The quality and quantity of DOM in aquatic systems is dynamic and dependent on some
31 characteristics such as regional climate, hydrologic regime, thermal stratification

1 pattern, depth, environment morphology, land cover, and also dependent on seasonal
2 events such as periods of algal bloom, and storms (Mattsson et al., 2005; Maie et al.,
3 2006; Jaffé et al., 2008).

4 The input of allochthonous DOM coming from terrestrial plants and soils in freshwater
5 systems is of great importance to the global carbon cycle and nutrients source as it
6 represents a linkage between terrestrial and aquatic ecosystems (Jaffé et al., 2008). In
7 addition, allochthonous DOM inputs influences the whole ecosystem functioning
8 because it interferes in the quality and quantity of carbon. The increase of aromatic
9 compounds changes the optical characteristics of water column as the transparency to
10 visible and ultraviolet radiation affecting the light available to productivity of the
11 system (Kieber et al., 2006; Miller et al., 2009 ; Brandão et al., 2016). In this way,
12 allochthonous DOM also affects the microbial food webs because besides it is more
13 recalcitrant to bacteria (Amon and Benner, 1994), it also influences the phytoplankton
14 dynamic and consequently the production of autochthonous DOM which is more labile
15 for microbial degradation (Lønborg et al., 2010). In addition, DOM quality also affects
16 several chemical reactions (Fulton et al., 2004).

17 With the current pattern of climate change in recent decades, rainfall and temperature
18 has been changed dramatically, with consequences for the inputs, production and
19 degradation of DOM in aquatic ecosystems (Tian et al., 2013). Some regions of the
20 world are facing reduction in annual rainfall, unseasonal rains and extremely warm
21 summers, such as the Middle Rio Doce (MG, Brazil) where the studied Lake Dom
22 Helvécio is located. These changes are expected to have affected allochthonous inputs
23 of DOM. This will likely alter the concentration and quality of carbon and nutrients
24 causing impacts on physical (thermal stratification patterns, temperature, light
25 availability) and biogeochemical processes (metabolism, ecosystem productivity) of
26 lakes in the region. However, studies on the dynamics of DOM in tropical aquatic
27 ecosystems are still scarce (Teixeira et al., 2011, 2013; Bittar et al., 2015; Brandão et
28 al., 2016) and to our knowledge none deals with effect of climate change. In this
29 context, the aim of this study was investigate if there were differences in DOM quantity
30 and quality in space and time in Lake Dom Helvécio through analysis of the optical
31 properties of DOM, especially considering the changes in the rainfall observed for the
32 last years in the region.

1 **2 Material and Methods**

2 *2.1 Study site*

3 Lake Dom Helvécio is located inside a protected area, the Rio Doce State Park – PERD
4 Southeast Brazil (S19°46.94', W42°35.48'). The lakes complex in the Middle Rio Doce
5 is one of the most important districts of lakes in Brazil, with approximately 130 lakes,
6 42 of them located inside the PERD area, which is the largest remnant of the Atlantic
7 Forest in Minas Gerais, totaling 36000 ha (Maia-Barbosa et al., 2010) (Fig. 1a-b). This
8 lake is one of the deepest natural lakes of Brazil (maximum depth: 39.5 m), surrounded
9 by Atlantic forest. It is oligotrophic (total phosphorus: 3.0-22.0 $\mu\text{g L}^{-1}$, annual mean
10 11.2 $\mu\text{g L}^{-1}$; Chl-*a*: 0.5-11.0 $\mu\text{g L}^{-1}$, annual mean 4.5 $\mu\text{g L}^{-1}$; Brighenti, 2014), has a
11 perimeter of 37.7 km, area equal to $5.27 \times 10^6 \text{ m}^2$ and volume equal to $5.94 \times 10^7 \text{ m}^3$
12 (Bezerra-Neto and Pinto-Coelho, 2008). Dom Helvécio is a warm monomictic lake,
13 with thermal stratification period beginning in September and lasting until April (rainy
14 period) and a mixed period from June to August (dry period) (Barbosa and Tundisi,
15 1980; Brighenti, 2014). Since 2012 the region of the Middle Rio Doce has had its
16 rainfall patterns changed, occurring unseasonal rains and lower volume of rainfall (mm)
17 during the year. The sampling months had unusual volume of rainfall and between
18 November of 2012 and April of 2013 the total volume of rainfall was 354 mm (in the
19 sampling months were: November - 80.2, January - 0 and April - 103 mm), which was
20 lower than that observed for the last years (Figure 1d; source: Climatological Station of
21 Timóteo, MG, Brazil). The rains that usually begin in September did not happen in 2013
22 and it remained until January of 2014, which classified December of 2013 (sampling
23 month) as an unusual dry period instead of expected rainy period. In contrast, during the
24 dry period in 2014 there were more rains than in the last years (Figure 1d).

25 *2.2 Field measurements*

26 The 6 samplings were carried out in November of 2012, January, April (stratified
27 periods), August (mixing period) and December (stratified period) of 2013 and in July
28 (mixing period) of 2014. We collected samples at 21 sites covering all the lake (Fig.
29 1c). At the central deep region (station 8) of the lake, vertical profiles of water column
30 temperature ($^{\circ}\text{C}$) were measured every meter from surface to 28 m using a Hydrolab
31 DS5 probe.

1 Water samples (0.5 m from the surface) were collected and filtered immediately for
2 Chl-*a* (0.47µm Millipore filter) and TSM (AP040 Millipore filter), and the filters were
3 frozen until analysis. Water samples were also filtered for analysis of dissolved organic
4 carbon (DOC) and colored dissolved organic matter (CDOM) (0.22 µm Millipore filter)
5 and stored in amber glass bottles (pre-washed with distilled water and hydrochloric acid
6 10%) at 4°C in the dark.

7 *2.3 Laboratory analysis and calculation of the optical properties*

8 The measurement of concentration of Chl-*a* (µg L⁻¹) was obtained by acetone extraction
9 (90%) measured in a spectrophotometer (UV-VIS Shimadzu) at 665 and 750 nm and
10 calculated according to APHA (1998). Total suspended matter (TSM; mg L⁻¹) were
11 determined by the gravimetric method, considering the difference between the dry
12 weight of AP40 Millipore filters (105°C for 2 hours) before and after the filtration of
13 water samples (APHA, 1998).

14 The DOC concentration (mg L⁻¹) was obtained by catalytic oxidation method of high
15 temperature using TOC Analyzer (Shimadzu TOC – 5000).

16 The spectral absorption of CDOM was measured between 250 and 700 nm (1 nm of
17 interval) in a spectrophotometer (UV-VIS Shimadzu) using 5 cm quartz cuvette and
18 Milli-Q water as a reference. The absorption spectra of each sample was performed in
19 triplicate and determined using the Spectrum Pack Program software (Shimadzu Inc.).
20 The absorption coefficients were derived from absorbance measurements (A) at λ (m⁻¹)
21 according to the equation $a_{CDOM}(\lambda) = 2.303A(\lambda) l^{-1}$, where $a_{CDOM}(\lambda)$ is the absorption
22 coefficient at λ and *l* is the optical path of the cuvette (in meters). Absorption
23 coefficients were corrected for backscattering by subtracting the value of the coefficient
24 at 700 nm from each absorption coefficient. The spectrophotometric absorption
25 coefficient at 440 nm was used as an index of PAR-absorbing (Williamson et al., 1996)
26 and at 254 nm as UV-absorbing of CDOM. The spectral slope *S* (nm⁻¹) was calculated
27 for wavelengths between 275-295 nm and 350-400nm (Helms et al., 2008) and it was
28 estimated using linear adjustment technique between log values of the absorption
29 coefficients and the specific wavelength range. The specific UV absorbance (SUVA₂₅₄)
30 (m² mg⁻¹ C) was calculated dividing the value of the absorption coefficient at 254 nm

1 by the concentration of DOC (mg L^{-1}), and the increase of SUVA indicates increase of
2 compounds of higher aromaticity (Weishaar et al., 2003).

3 Accumulated rainfall recorded at every three hours was obtained from a meteorological
4 station (Timóteo, MG) located near to the lake (<http://www.inmet.gov.br>).

5 Unfiltered water samples were taken for analysis of total nitrogen (TOC Analyzer,
6 Shimadzu TOC – 5000) and total phosphorus (according to Mackereth et al., 1978).

7 *2.4 Statistical analysis*

8 Differences in DOM quality and quantity between seasons were assessed by Kruskal-
9 Wallis One Way (Analysis of Variance on Ranks) and followed by Pairwise Multiple
10 Comparison (Dunn's Method) analysis (significant $p < 0.05$).

11 Pearson correlations and regression analysis were made to examine the relationships
12 between all the parameters measured (significant $p < 0.05$).

13 To demonstrate the spatial distribution of DOM average values from six temporal
14 samplings of $S_{275-295}$, $SUVA_{254}$, and DOC were used to predict values for unsampled
15 locations in a raster. The interpolation was made using ArcGIS's 10.3 – Geostatistical
16 Analyst extension. Different surfaces were created using different interpolation methods
17 (*i.e.* IDW and Kriging) or by using the same method with various parameters. These
18 surfaces were systematically compared with each other, eliminating the worst of the two
19 being compared, until only the best surface remained. The criteria for eliminating the
20 surfaces was the standardized mean nearest to zero, the smallest root-mean-squared
21 prediction error, the average standard error nearest the root-mean-squared prediction
22 error, and the standardized root-mean-squared prediction error nearest to 1. Therefore,
23 our model has simple kriging as its interpolation method, with Geostatistical Analyst's
24 model optimization, standard neighbourhood, and four sectors.

25 A principal component analysis (PCA) was generated with correlation matrix
26 (parameters with different units) using the parameters of quality and quantity of DOM
27 (DOC, $a_{CDOM440}$, $a_{CDOM254}$, $SUVA_{254}$ and $S_{275-295}$), Chl-a, depth, sites, seasons and
28 rainfall.

29 **3 Results**

1 *3.1 Variation in the parameters measured*

2 The lake was thermally stratified during November 2012 and April 2013 and not
3 stratified (mixing periods) in August 2013 and July 2014. The water surface
4 temperature ranged from 21 to 31 °C with lowest temperatures during the mixing
5 periods.

6 Considering all sites and sampling periods, the $S_{275-295}$ values ranged from 0.019 (site 4
7 in July 2014) to 0.036 nm^{-1} (site 15 in January 2013), and there is no clear variation
8 over time and seasons. $SUVA_{254}$ ranged from 0.880 (site 3 in August 2013) to 4.824 m^2
9 $\text{mg}^{-1} \text{C}$ (site 4 in July 2014), while DOC concentrations ranged between 2.5 (site 3 in
10 April 2013) and 8.5 mg L^{-1} (site 7 in December 2013).

11 The highest concentration of Chl-a occurred in the site 8 (11.0 $\mu\text{g L}^{-1}$ during April
12 2013) and the lowest in the site 15 (0.5 $\mu\text{g L}^{-1}$ during November 2012). Higher averages
13 of Chl-a occurred during mixing periods (August 2013 – 4.8 $\mu\text{g L}^{-1}$; July 2014 – 3.6 μg
14 L^{-1}) and during thermal stratification in April 2013 (7.0 $\mu\text{g L}^{-1}$) (Table 1).

15 TN and TP did not have the same pattern. Higher averages of TN concentrations were
16 observed during mixing periods (August 2013 – 0.4 mg L^{-1} ; July 2014 – 0.7 mg L^{-1}).
17 There was no seasonality for TP concentrations, and the highest averages occurred in
18 the stratified November of 2012 (17.2 $\mu\text{g L}^{-1}$) and during the mixing period in August
19 of 2013 (12.7 $\mu\text{g L}^{-1}$) (Table 1).

20 *3.2 Spatial and temporal variation of DOM*

21 Analyzing the average considering the 6 sampling, the west portion of the lake (sites 19,
22 20 and 21) had DOM more aromatic (higher $SUVA_{254}$) and compounds with higher
23 molecular size (lower $S_{275-295}$) (Figure 2a-b). DOC concentrations did not have clear
24 spatial differences (Figure 2c).

25 $S_{275-295}$ presented variation among sites about 0.002-0.003 nm^{-1} (Table 1, standard
26 variation). $SUVA_{254}$ had higher standard variation among sites in November 2012 and
27 July 2014 (Table 1), while for DOC it was observed in November 2012, August and
28 December of 2013.

1 Although there were differences between the sampled sites, differences among seasons
2 were higher than among sites for quantity and quality of DOM (Figure 3 and Table 1).
3 There were differences among almost all the seasons to SUVA and DOC, but it was not
4 the same to $S_{275-295}$ (Table 2). $S_{275-295}$ changed over time significantly, but differences
5 were observed only between the months January-April 2013 and July 2014 (Figure 4a
6 and Table 2). DOC and $SUVA_{254}$ were negative correlated ($R = -0.80$; $R^2 = 0.64$, $p <$
7 0.000), which indicates a strong connection between quality and quantity of DOM
8 (Figure 5). We did not find defined seasonal pattern to DOM quality and quantity
9 related to thermal stratification and mixing processes.

10 The PCA results showed that 4 principal components had eigenvalue higher than 1 and
11 explained 76% of variance (PC1 – 29%; PC2 – 24%; PC3 – 12%; PC4 – 11%) of the
12 parameters measured (DOC, $SUVA_{254}$, $S_{275-295}$, $a_{CDOM254}$, $a_{CDOM440}$, Chl-a, depth,
13 seasons, sites and rainfall). With the plot of PC1 versus PC2 (Figure 6) and with the
14 Pearson's correlations (Table 3) we see the opposite relation between DOC and
15 $SUVA_{254}$, a positive effect of rainfall in $SUVA_{254}$ and Chl-a and negative in DOC
16 concentration and $a_{CDOM440}$, and an effect of the increase of depth decreasing $a_{CDOM254}$.
17 We also observed that the different sampling periods (seasons) had an effect in $a_{CDOM440}$
18 and going to the direction from site 1 to site 21 (going from north to west portion of the
19 lake) the $a_{CDOM254}$ increased (Figure 6).

20 **4 Discussion**

21 There were spatial and temporal differences in the quantity (DOC) and quality of DOM
22 ($SUVA_{254}$ and $S_{275-295}$) in Lake Dom Helvécio. Spatially the differences in DOM were
23 lower than seasonally, and it was not identified a clear spatial pattern related to the
24 variables investigated in this study (Chl-a, nutrients, TSM, depth). Only considering the
25 seasonality is that we identified a relationship between DOM and Chl-a with rainfall
26 and between $a_{CDOM254}$ and depth, while no direct relations were found with water
27 column stratification and mixing periods.

28 Higher concentrations of Chl-a were observed during periods with full water column
29 mixing (August 2013 and July 2014) but also during the stratified April of 2013.
30 Phytoplankton community is known to be limited by nutrients (especially by phosphate)
31 in the epilimnion during stratified periods in Dom Helvécio (Barros et al., 2006;

1 Brighenti et al., 2015). It happens because the input of terrestrial OM and nutrients with
2 the rains remains in the hypolimnion due to differences in the temperature in the layers
3 of water column (Reynolds, 2009) and they become distributed and available in the
4 epilimnion only during mixing water between May and August as already demonstrated
5 by some authors (Barros et al., 2006; Brighenti et al., 2015; Brandão et al., 2016). The
6 highest Chl-a concentration observed in April of 2013 (stratified period) might have
7 happened due to higher concentrations of TP at the same time and also as a consequence
8 of smaller volume of rains since November of 2012.

9 The seasonal differences found for DOC, $a_{CDOM440}$ and $SUVA_{254}$ were related to
10 cumulative rainfall for each sampling month (Table 3). This suggests that inputs of
11 aromatic terrestrial DOM from the basin are coupled to rainfall events (Reynolds,
12 2009), increasing $SUVA_{254}$ values (Brandão et al., 2016). Although $SUVA_{254}$ has DOC
13 concentrations in its formula ($a_{CDOM254}/DOC$) it is not guaranteed a strong negative
14 correlation between them (Jaffé et al., 2008), but in this study they were strongly
15 correlated ($R = -0.80$; $R^2 = 0.64$; $p < 0.000$). The relationship implies that there was a
16 significant relation between quantity and quality of DOM (Jaffé et al., 2008). It means
17 that with the degradation of DOC the molecules become more aromatics, in other words
18 the compounds responsible for the absorbance in the PAR range (labile compounds) are
19 more degraded than those responsible for the absorbance in the UV spectrum (aromatic
20 compounds). It is known that photodegradation provides loss in PAR and UV-
21 absorbing, but it has more impact mainly in the ultraviolet range (Markager and
22 Vincent, 2000). In contrast, the loss by biodegradation is generally higher in PAR-
23 absorbing due to preference of bacterial communities for more labile and bioavailable
24 compounds (Asmala et al., 2014) and this results suggest that this is the main process of
25 DOM degradation in this lake. It corroborates with the negative relation between
26 $a_{CDOM440}$ and rainfall (Table 3), which means a loss of absorption in PAR range with the
27 increase of rains e $SUVA_{254}$.

28 Although the seasonal differences for DOC, $SUVA_{254}$ and $S_{275-295}$ were statistically
29 significant (Table 2, $p < 0.05$), there was only a smooth reduction of $S_{275-295}$ temporally
30 (Figure 4a). The decline of $S_{275-295}$ may indicate an increase in molecular size (Helms et
31 al., 2008), which corroborated with the increase of aromatic compounds viewed through
32 the increase of $SUVA_{254}$ with DOC degradation.

1 The results obtained through the PCA indicated that the variation of $a_{\text{CDOM}254}$ was more
2 related to the sampling sites (spatial variation) and of $a_{\text{CDOM}440}$ to the seasonality
3 (temporal variation) (negative Pearson correlations, Table 3). In deeper sites there may
4 be a limitation in the depth of the mixed layer, and the aromatic carbon coming into the
5 lake with the rains remain restricted in the hypolimnion and do not reach the surface
6 layers even with the partial mixing of water column. This fact could be an explanation
7 for the increase in depth of the sampling sites causing a decline in $a_{\text{CDOM}254}$. This is
8 consistent with more aromatic carbon compounds (higher SUVA_{254}) with higher
9 molecular size (lower $S_{275-295}$) observed in west region of the lake (points 19, 20 and 21;
10 Figure 2a-b) in the sites of lowest depths (between 4.7 and 6.5 m). These points are
11 close to the lakeshore and certainly accumulate much of the terrestrial carbon during the
12 input by rains.

13 This study showed that the change in rainfall pattern in the region has affected the
14 seasonal patterns of input of DOM and nutrients in the lake, reflecting in the seasonality
15 of phytoplankton and consequently affecting other trophic levels and the metabolism of
16 the ecosystem. In addition, the absence of spatial pattern to DOC proves the importance
17 of including the quality of DOM in carbon dynamics studies on aquatic ecosystems in
18 order to understand the lake functioning.

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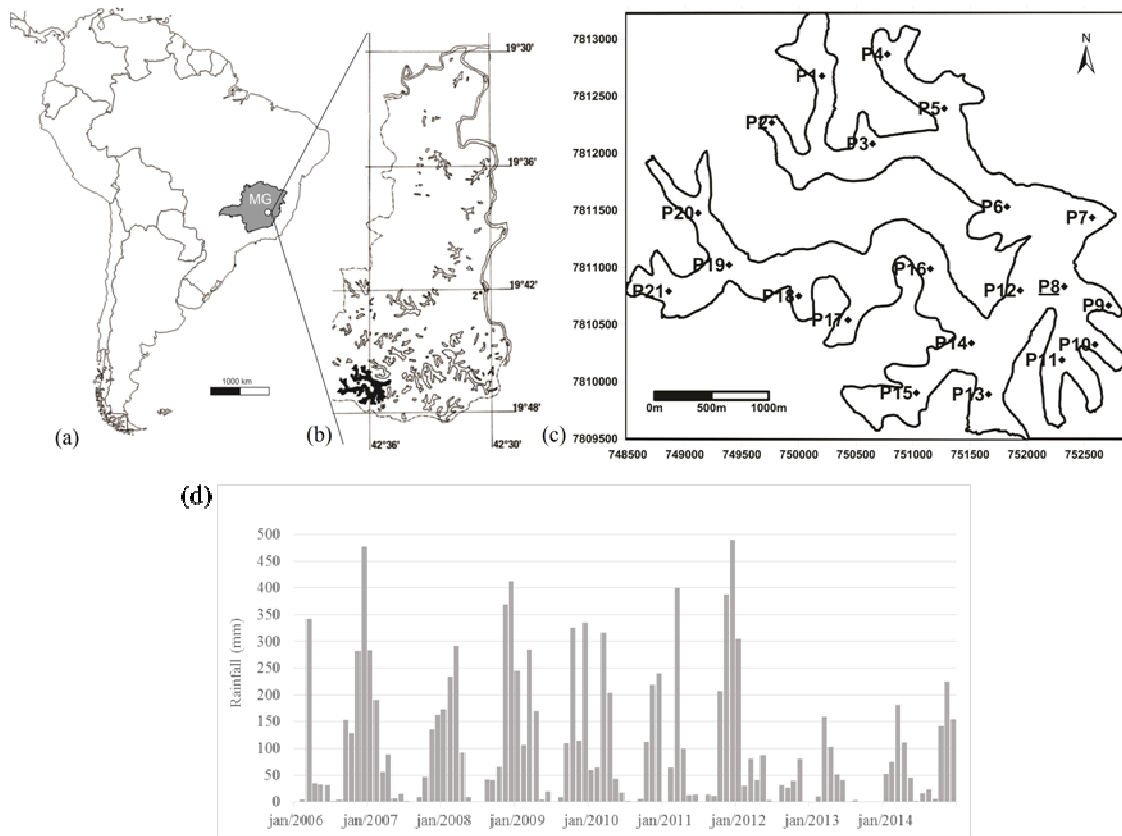
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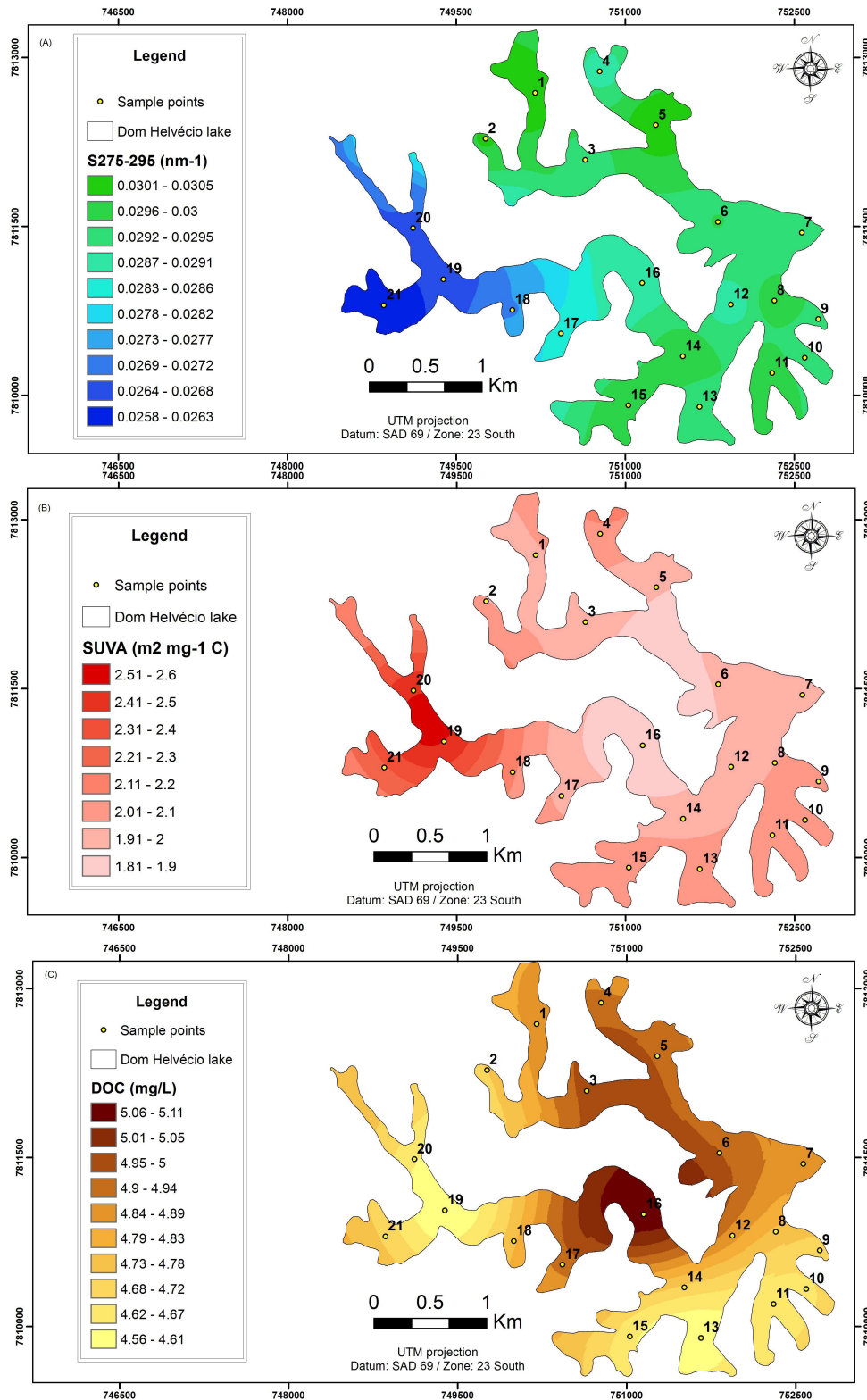
1 **Figures and Tables**



2

3 Figure 1 - Location of Dom Helvécio lake in Minas Gerais (a), inside of the Rio Doce
4 State Park (b), Middle Rio Doce and the spatial sampling stations (P1-P21) (c). The
5 numbers on the axes correspond to geographic coordinates. Rainfall from January 2006
6 to December 2014 (d).

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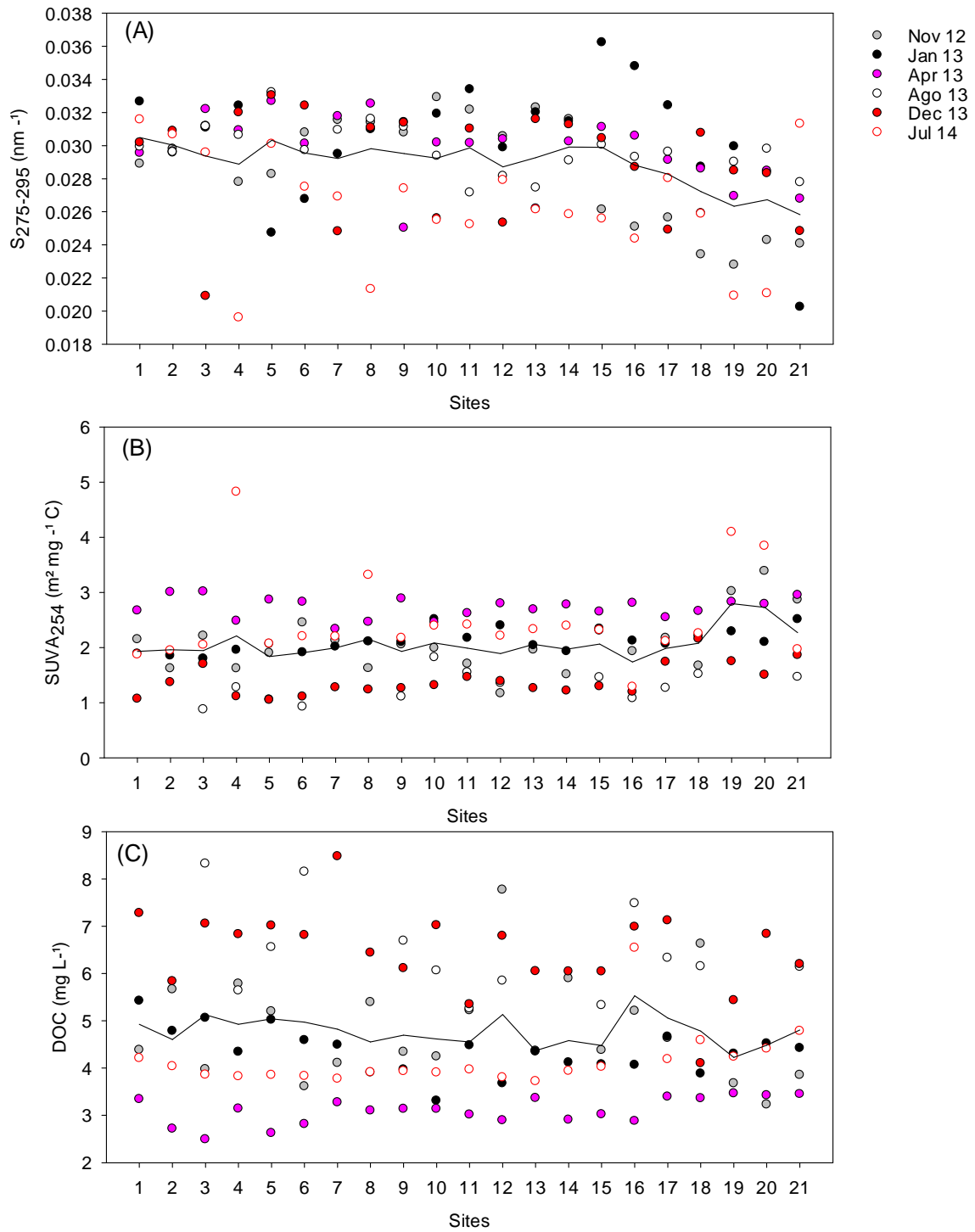


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2 Figure 2 – Spatial maps of the average values from 6 temporal sampling of S₂₇₅₋₂₉₅ (A),
 3 SUVA₂₅₄ (B) and DOC (C).

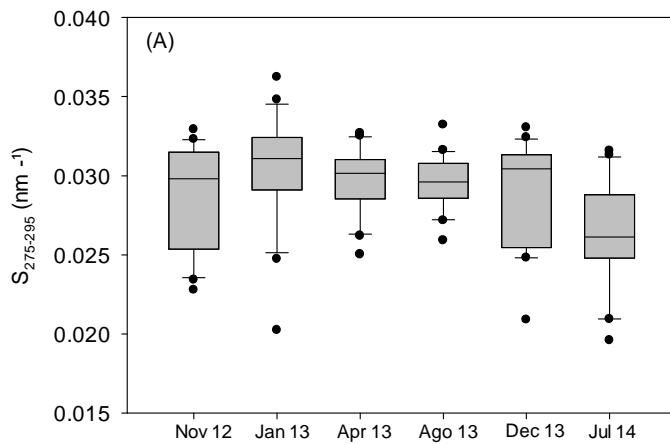
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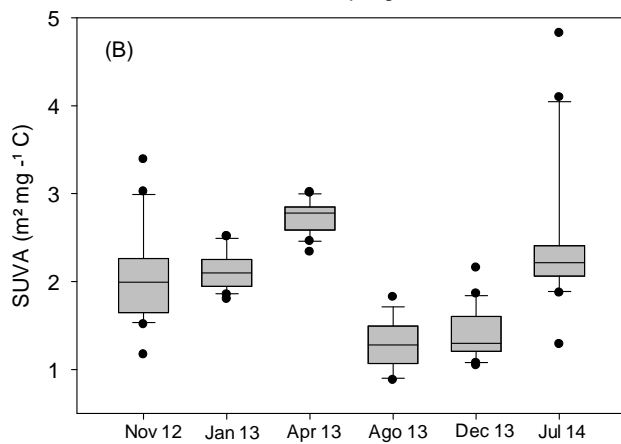


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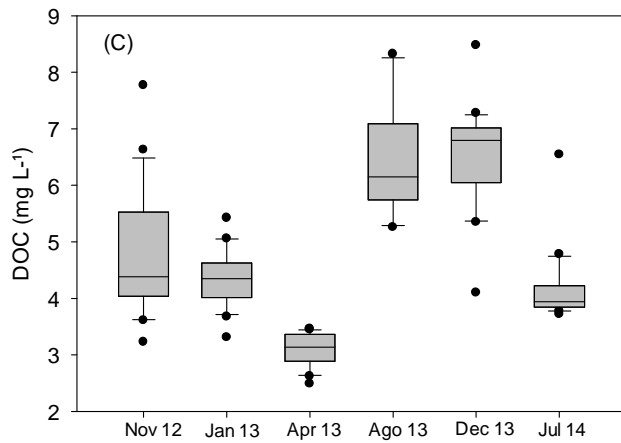
2 Figure 3 – Distribution and range of $S_{275-295}$ (A), $SUVA_{254}$ (B) and DOC (C) for all
 3 samples from studied sites. The black line represents the average of each site in the 6
 4 sampling periods.



Sampling



Sampling



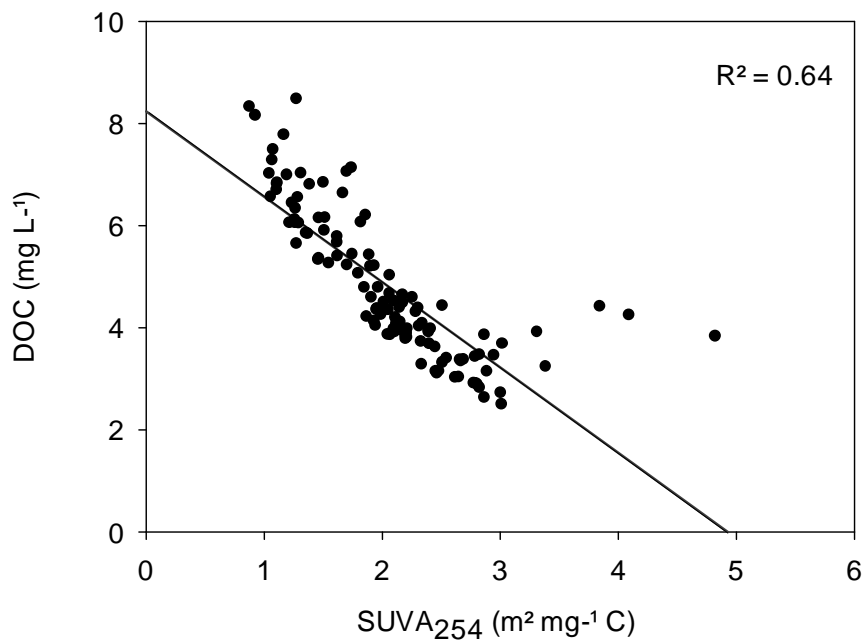
Sampling

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2 Figure 4 – Temporal variations (center line-median, outer grey-5th/95th percentiles,
 3 bars- 95% confidence level, black circles- outliers) of $S_{275-295}$ (A), $SUVA_{254}$ (B) and
 4 DOC (C) at 21 sites.

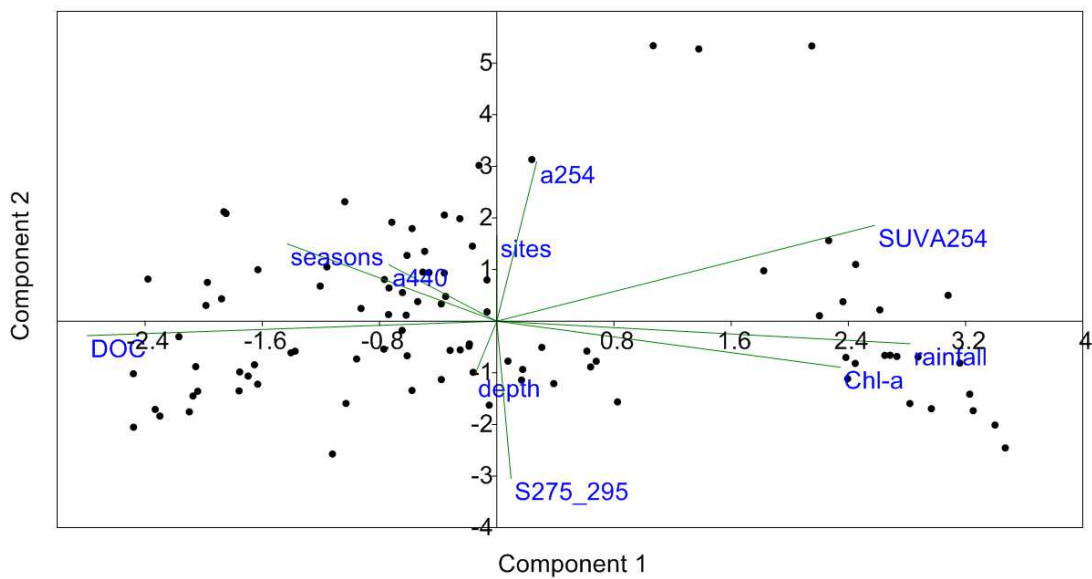
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2 Figure 5 – Relation between DOC and SUVA₂₅₄ (Linear regression: R = -0.80, R² =
3 0.64, p < 0.000).



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5 Figure 6 – Results of principal component analysis plotting component 1 (29% of
6 explanation) versus component 2 (24%).

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1 Table 1 - Summary (mean and standard variation-SD) of parameters measured during
 2 temporal and spatial analyzes in the six sampling campaigns.

		DOC (mg L ⁻¹)	SUVA254 (m ² mg ⁻¹ C)	S275-295 (nm ⁻¹)	a _{CDOM440} (m ⁻¹)	a _{CDOM254} (m ⁻¹)	Chl-a (µg L ⁻¹)	TSM (mg L ⁻¹)	TN (mg L ⁻¹)	TP (µg L ⁻¹)	Depth (m)
Nov/12	mean	4.8	2.070	0.029	0.034	9.540	3.3	2.2	0.3	17.2	14.0
	SD	1.1	0.530	0.003	0.065	0.895	1.8	0.5	0.1	3.2	8.5
Jan/13	mean	4.4	2.115	0.030	0.152	9.148	3.3	1.4	0.3	6.0	14.0
	SD	0.5	0.204	0.003	0.105	0.805	1.8	0.4	0.0	3.4	8.5
Apr/13	mean	3.1	2.722	0.030	0.097	8.396	7.0	1.8	0.3	10.3	14.0
	SD	0.3	0.188	0.002	0.082	0.787	1.8	0.2	0.0	8.4	8.5
Aug/13	mean	6.5	1.291	0.030	0.115	7.979	4.8	1.7	0.4	12.7	14.0
	SD	1.0	0.274	0.002	0.102	0.913	1.5	0.6	0.1	4.1	8.5
Dec/13	mean	6.5	1.398	0.029	0.121	8.914	2.9	1.4	0.1	8.3	14.0
	SD	0.9	0.291	0.003	0.106	1.611	0.9	0.3	0.2	1.8	8.5
Jul/14	mean	4.2	2.490	0.026	0.269	10.216	3.6	1.2	0.7	3.5	14.0
	SD	0.6	0.835	0.003	0.164	3.293	1.0	0.3	0.2	1.1	8.5

3

4 Table 2 – Summary results of Kruskal-Wallis One Way (Analysis of Variance on
 5 Ranks) and Pairwise Multiple Comparison (Dunn’s Method) analyzes.

Differences among seasons by Dunn's Method (p<0.05)							Kruskal-Wallis results (n = 21)		
S275-295							S275-295 (p=0.003)		
	Nov 12	Jan 13	Apr 13	Aug 13	Dec 13	Jul 14	Median	25%	75%
Nov 12	-	no	no	no	no	no	0.0298	0.0254	0.0315
Jan 13		-	no	no	no	yes	0.0311	0.0291	0.0324
Apr 13			-	no	no	yes	0.0302	0.0285	0.031
Aug 13				-	no	no	0.0296	0.0286	0.0308
Dec 13					-	no	0.0304	0.0255	0.0313
Jul 14						-	0.0261	0.0248	0.0288
SUVA254							SUVA254 (p<0.001)		
	Nov 12	Jan 13	Apr 13	Aug 13	Dec 13	Jul 14	Median	25%	75%
Nov 12	-	no	yes	yes	yes	no	1.992	1.648	2.261
Jan 13		-	yes	yes	yes	no	2.1	1.944	2.251
Apr 13			-	yes	yes	no	2.779	2.585	2.848
Aug 13				-	no	yes	1.279	1.069	1.493
Dec 13					-	yes	1.296	1.208	1.604
Jul 14						-	2.214	2.06	2.406
DOC							DOC (p<0.001)		
	Nov 12	Jan 13	Apr 13	Aug 13	Dec 13	Jul 14	Median	25%	75%
Nov 12	-	no	yes	no	yes	no	4.383	4.04	5.531
Jan 13		-	yes	yes	yes	no	4.351	4.017	4.627
Apr 13			-	yes	yes	yes	3.135	2.888	3.364
Aug 13				-	no	yes	6.152	5.745	7.09
Dec 13					-	yes	6.797	6.045	7.018
Jul 14						-	3.941	3.844	4.224

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1 Table 3 - Pearson correlations of the parameters obtained in the temporal and spatial
 2 analysis (n = 126). Only significant correlations are shown.

	SUVA254 (m ² mg ⁻¹ C)	Chl-a (µg L ⁻¹)	Depth (m)	Rainfall (mm)
DOC (mg L ⁻¹)	-0.80***	-0.43***		-0.53***
SUVA254 (m ² mg ⁻¹ C)				0.46***
a _{CDOM440} (m ⁻¹)		-0.19*		-0.29**
a _{CDOM254} (m ⁻¹)			-0.218*	
Chl-a (µg L ⁻¹)				0.65***

3 * p < 0.05 ** p < 0.01 ***p < 0.001

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CAPÍTULO 3

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Seasonal changes in optical properties of two contrasting tropical freshwater systems

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Abstract

We investigated how allochthonous and autochthonous sources of dissolved organic matter (DOM) affected the optical conditions and chemical characteristics of two contrasting tropical freshwater systems (Dom Helvécio-DH and Pampulha reservoir) in a dry and rainy period in 2013. We analyzed PAR (photosynthetically active radiation) and UV (ultraviolet) attenuation coefficients, nutrients, chlorophyll-a (Chl-a), dissolved organic matter (DOC) and spectral characteristics of CDOM (colored dissolved organic matter). Significant differences in CDOM sources and quantity were observed, with a dominantly terrestrial input in DH during the rainy period. The eutrophic Pampulha had several fold higher levels of DOC, Chl-a, TN and TP, with organic matter of mostly originating from phytoplankton in both seasons. Differences in source and quantity had strong implications on water transparency, DOC concentrations, CDOM quality and its susceptibility to photo- and biodegradation. DH was several fold clearer in both the UV and PAR spectrum. In DH transparency to both UV and PAR radiation was highest during the summer, suggesting elevated photo- and biodegradation during stratification.

1 Pampulha was most transparent in the dry period even during period of algal bloom. In
2 both systems we observed seasonal variations in concentrations of nutrients and Chl-a,
3 and in DH differences were also found in DOC concentrations as well as the specific
4 UV absorbance (SUVA₂₅₄) and molecular size (M). Our results documents that different
5 sources of DOM and seasonal inputs reflect in the seasonality of apparent and inherent
6 optical properties and nutrients availability with implications for water quality and
7 aquatic community.

8 Keywords: CDOM, DOM, optical properties, photodegradation, carbon source

9

10 1. INTRODUCTION

11 The dissolved form of organic matter (DOM) constitutes about 90% of the total pool of
12 organic matter (Tranvik and Wachenfeldt, 2009), and understanding of the production
13 and transformation of DOM is of key importance in studies of carbon cycling in lakes
14 (Thurman, 1985; Bertilson and Tranvik, 2000). DOM represents a complex of
15 compounds and derives from main sources: allochthonous material produced from
16 degradation of humic substances coming from runoff of terrestrial origin (Kieber *et al.*,
17 2006; Miller *et al.*, 2009); and autochthonous material, produced by autotrophic and
18 heterotrophic in-situ activities, but mainly by the bacterial degradation of phytoplankton
19 and macrophytes (Kritzberg *et al.*, 2004; Hanamachi *et al.*, 2008; Henderson *et al.*,
20 2008; Guillenette and Del Giorgio, 2012). The two different sources of DOM can be
21 distinguished from differences in their optical and chemical characteristics. The DOM
22 from autochthonous source is furthermore more biologically labile than the DOM from
23 allochthonous origin, which absorbs more solar radiation, has higher molecular weight

1 due to the more complex aromatic molecules in its composition (McKnight *et al.*, 1994;
2 Benner, 2002).

3 CDOM is the chromophoric or colored part of the DOM and also an optically active
4 component which plays an important role in freshwater systems. CDOM absorbs solar
5 radiation reducing exponentially the penetration of light in the water column and the
6 absorption increases from long to short wavelengths being higher in the ultraviolet (UV)
7 range (Twardowski *et al.*, 2004). The CDOM absorption is due to photo-oxidation
8 (known as photodegradation) of the molecules that can originate photo products or can
9 be completely mineralized to CO₂, affecting the amount of CO₂ emission to the
10 atmosphere (Arrigo *et al.*, 2011). This phenomenon restricts the available light for
11 production, abundance and distribution of pelagic and benthic primary producers, with
12 consequences for higher trophic levels in the ecosystem (McMahon *et al.* 1992,
13 Gallegos and Moore 2000). Moreover, it significantly reduces harmful effects of UV
14 radiation on aquatic organisms (Zhang *et al.*, 2011). Several studies in temperate aquatic
15 systems have demonstrated the role of CDOM in the light attenuation (Christian and
16 Sheng, 2003; Balogh *et al.*, 2009; Yamaguchi *et al.*, 2013), but the effect of CDOM in
17 tropical systems still requires effort (Obrador and Pretus, 2008) especially in lakes
18 (Brandão *et al.*, in review). DOM is furthermore an energy source for the bacterial
19 production, which is responsible for the regeneration of nutrients in the water (Lønborg
20 *et al.*, 2010). It also operates in the physical, chemical and biological processes of
21 transport and availability of nutrients and the development of blooms of phytoplankton
22 in eutrophic lakes (Qualls and Richardson, 2003; Coble, 2007; Stedmon *et al.*, 2007)
23 and it is an important component to estimates and effective management of water
24 quality and eutrophication control (Zhang *et al.*, 2011b). Understanding the
25 environmental conditions and physical, chemical and biological processes involved in

1 the production and mineralization of DOM, is therefore also of great importance to
2 assess the role of lakes in carbon cycling at regional and global scale (Bertilsson and
3 Tranvik, 2000; Johannessen *et al.*, 2007). Although the importance of distribution and
4 cycling of DOM is well known, the origins, transport and transformations are not well
5 understood (Zhang *et al.*, 2009), especially in tropical environments.

6 The composition and concentration of DOM in aquatic ecosystems depends mainly on
7 the spatial and seasonal variability of its sources and the availability of nutrients and
8 radiation, which also affects the rates of photo- and microbial degradation
9 (biodegradation) of these compounds (Bracchini *et al.*, 2009; Zhang *et al.*, 2011b). The
10 spatial variation is given by the heterogeneity of the environment, as the presence or
11 absence of macrophytes, different depths and the proximity to the borders which
12 increases the impact by the input of allochthonous matter and nutrients by runoff
13 (Wetzel, 1992; Tao, 1998; Obrador and Pretus, 2013). The seasonal variation derives
14 from changes in DOM sources, such as higher input of allochthonous organic matter in
15 the lake during the rainy season and the highest incidence of radiation in spring and
16 summer, which can increase photodegradation rates of CDOM and still favor the lability
17 of the molecules through the breakage part, facilitating microbial degradation
18 (Bertilsson and Tranvik, 1998; Neale *et al.*, 2007; Fasching and Battin, 2012; Catalán *et*
19 *al.*, 2013). Furthermore, CDOM degradation is also affected by the availability of
20 nutrients throughout the year, a limiting factor for microbial degradation. In eutrophic
21 lakes, degradation of phytoplankton is an important source of CDOM and regeneration
22 of nutrients for bacteria and also the phytoplankton community, especially after
23 seasonal blooms of these organisms (Bracchini *et al.*, 2009; Zhang *et al.*, 2009; Zhang
24 *et al.*, 2013b).

1 The CDOM spectrophotometry and fluorescence spectroscopy are important tools in
2 studies of composition and origin of organic matter (Helms *et al.*, 2008). There are
3 currently some metrics extracted from the CDOM absorbance spectrum that provide
4 information about the chemical characteristics and quality of its composition. The
5 increase of the specific UV absorbance ($SUVA_{254}$; $\text{mg C L}^{-1} \text{ m}^{-1}$) indicates an increase of
6 compounds of higher aromaticity (Weishaar *et al.*, 2003). The spectral slope $S_{275-295}$
7 (nm^{-1}) indicates different sources of CDOM and changes related to photodegradation
8 and molecular size, while $S_{350-400}$ (nm^{-1}) indicates changes in CDOM related to
9 biodegradation (Helms *et al.*, 2008). S_R is a ratio between both spectral slopes (S_{275-}
10 $295:S_{350-400}$) and indicates the relationship between photodegradation and biodegradation
11 of CDOM (Helms *et al.*, 2008). The estimated apparent molecular size (M) is obtained
12 by the ratio of two absorption coefficients ($a_{250}:a_{365}$), based on the fact that declining
13 ratio indicates increasing of molecular size (De Haan, 1987; Peuravuori and Pihlaja,
14 1997). In recent decades researchers have used these spectral metrics to investigate the
15 optical properties in temperate aquatic systems (Zhang *et al.*, 2007, 2007b, 2009; Helms
16 *et al.*, 2008; Bracchini *et al.*, 2010), but studies using this approach in tropical
17 environments are still scarce and little is known to Brazilian aquatic systems (Teixeira
18 *et al.*, 2011, 2013; Bittar *et al.*, 2015).

19 The aim of the study was to investigate the effects of CDOM sources on the light
20 environment and nutrient availability in the two contrasting tropical freshwater systems.
21 Specifically we focus on DOM properties obtained from the CDOM spectral absorbance
22 (absorbance coefficient (a_{350}), spectral slopes ($S_{275-295}$, $S_{350-400}$), slope ratio (S_R),
23 molecular size (M)).

24 Dom Helvécio-DH (Middle Rio Doce, MG, Brazil) is a warm monomictic, deep (max
25 39.5 m), dendritic, oligotrophic and natural lake located in a natural Atlantic forest. The

1 input of DOM and nutrients to DH occurs via runoff during the rainy season, when the
2 environment is thermally stratified (Petruccio *et al.*, 2005). However, most of the DOM
3 remains below the thermocline due to temperature differences (Reynolds, 2009) and
4 becomes dispersed into the entire water column during the mixing of water in the dry
5 season, increasing the DOC and CDOM concentrations in surface waters (Brighenti *et*
6 *al.*, 2015; Brandão *et al.*, in review). In contrast, Pampulha (Belo Horizonte, MG,
7 Brazil) is a shallower (max 17 m) and less dendritic eutrophic reservoir with reoccurring
8 phytoplankton blooms throughout the year, especially in the summers during thermal
9 stratification. As in other eutrophic lakes, concentrations and quality of the DOM are
10 strongly influenced by the high phytoplankton abundance (Tôrres *et al.*, 1998; Bracchini
11 *et al.*, 2009; Zhang *et al.*, 2009; Zhang *et al.*, 2013b). Considering that the main source
12 of dissolved organic matter in the eutrophic Pampulha reservoir comes from
13 degradation of phytoplankton, especially after an algal bloom, while the main source of
14 DOM in DH Lake comes from allochthonous material from the forest in rainy season,
15 we hypothesized that: a) The different sources of DOM (allochthonous in DH and
16 autochthonous in Pampulha) provide different optical qualities of CDOM, which alters
17 the under-water light climate and the availability of nutrients; b) The lakes have
18 seasonal differences in their inherent and apparent optical properties. The
19 autochthonous organic matter in the Pampulha reservoir is expected to increase with the
20 increase of phytoplankton abundance during higher availability of nutrients, which will
21 reduce the attenuation of visible and ultraviolet radiation by algae and CDOM
22 absorption and also light dispersion by the phytoplankton community. In opposition to
23 this we expect that DH Lake will show higher water transparency during the rainy
24 season because of intense photobleaching in the epilimnion, and because most of the

1 organic matter coming from the catchment remains in the hypolimnion until the mixing
2 in winter.

3 2. METHODS

4 2.1 Study area

5 Lake Dom Helvécio (DH) is located inside the Rio Doce State Park - PERD (Middle
6 Rio Doce, MG, Brazil). The lake complex in the Middle Rio Doce is one of the most
7 important districts of lakes in Brazil, with approximately 130 lakes, 42 of them located
8 inside the PERD area, which is the largest remnant of the Atlantic Forest in Minas
9 Gerais, totaling 36000 ha (Maia-Barbosa *et al.*, 2010) (Fig. 1a). This lake is one of the
10 deepest natural lakes in Brazil (maximum depth: 39.5 m), surrounded by Atlantic forest.
11 It is oligotrophic (total phosphorus: 3.0-22.0 $\mu\text{g L}^{-1}$, annual mean 11.2 $\mu\text{g L}^{-1}$;
12 Chlorophyll-a - Chl-a: 0.5-11.0 $\mu\text{g L}^{-1}$, annual mean 4.5 $\mu\text{g L}^{-1}$), has a perimeter of 37.7
13 km, area equal to $5.27 \times 10^6 \text{ m}^2$ and volume equal to $5.94 \times 10^7 \text{ m}^3$ (Bezerra-Neto and
14 Pinto-Coelho, 2008). DH is a warm monomictic lake, with a stable thermal stratification
15 period beginning in September and lasting until May (rainy period) and a mixed period
16 from June to August (dry period) (Barbosa and Tundisi, 1980; Henry and Barbosa,
17 1989; Briguenti *et al.*, 2015).

18 The Pampulha reservoir was inaugurated in 1938 and rebuilt in 1957 following a dam
19 rupture, and is together with its architectural set an important tourist area in the
20 metropolitan region of Belo Horizonte, Minas Gerais, Brazil. This reservoir is eutrophic
21 (total phosphorus: 135-819 $\mu\text{g L}^{-1}$, annual mean 6500 $\mu\text{g L}^{-1}$; Chl-a: 10-315 $\mu\text{g L}^{-1}$,
22 annual mean 75 $\mu\text{g L}^{-1}$), has an original perimeter of 21 km, maximum depth of 17 m,
23 and receives domestic and industrial sewage and other pollutants that contribute to poor
24 water quality and exacerbate eutrophication (Champs, 1992; Novais, 1992, Pinto-

1 Coelho, 1998, Resck *et al.*, 2007) (Fig. 1b). Pampulha remains thermally stratified
2 between September and April (rainy period) and the mixing water begins in May until
3 August (dry period) (Figueredo and Giani, 2001; Pinto-Coelho *et al.*, 2003).

4 *2.2 Field measurements and laboratory analysis*

5 The sampling was carried out in January (rainy summer) and August (dry winter) 2013
6 in 21 sampling stations in DH Lake and in March (rainy autumn) and July (dry winter,
7 during a bloom of cyanobacteria) 2013 in 16 sampling stations in Pampulha reservoir
8 (Fig. 1a-b).

9 For characterization of the systems vertical profiles of water column temperature (°C) at
10 the central point were measured every meter from surface to the bottom using a probe
11 Hydrolab DS5 (Hach Inc.).

12 Vertical profiles of photosynthetically active radiation (PAR) and ultraviolet radiation
13 (UV) (between 10:00 and 14:00) were performed using a radiometer BIC (Biospherical
14 Instruments) in each site sampling. The profiles were made on the sun side of the boat
15 avoiding shadows and measurements during days with great clouds were avoided. The
16 radiometer was attached to a battery and notebook with the software LoggerLight™
17 (Biospherical Instruments) while collecting radiation data PAR and UV (305, 340 nm)
18 and depth during descent and ascent up of the probe.

19 Water samples (0.5 m from the surface) were collected using a van Dorn bottle (5L).
20 These samples were filtered immediately after sampling for Chl-a (0.47µm filter) and
21 TSM (AP040 filter) and the filters were frozen until analysis. Water samples were also
22 filtered for analysis of dissolved organic carbon (DOC) and colored dissolved organic
23 matter (CDOM) (0.22 µm Millipore filter) and stored in amber glass bottles (pre-
24 washed with distilled water and hydrochloric acid 10%) at 4°C in the dark. The

1 chlorophyll-a concentration corrected by pheophytin (Chl-a) was obtained by acetone
2 extraction (90%) measured in a spectrophotometer (UV-VIS Shimadzu) at 665 and 750
3 nm and calculated according to APHA (1998). The DOC concentration (mg L^{-1}) was
4 obtained by catalytic oxidation method of high temperature using TOC Analyzer
5 (Shimadzu TOC – 5000). Total suspended solids (TSM) were determined by the
6 gravimetric method, considering the difference between the dry weight of AP40
7 Millipore filters (105 °C for 2 hours) before and after the filtration of water samples
8 (APHA, 1998).

9 Unfiltered water samples were taken for analysis of total nitrogen - TN (TOC Analyzer,
10 Shimadzu TOC – 5000) and total phosphorus - TP (according to Mackereth *et al.*,
11 1978).

12 *2.3 Calculation of the optical properties*

13 The diffuse PAR and UV attenuation coefficients ($K_{d(\text{PAR})}$ and $K_{d(\text{UV})}$) were calculated as
14 the linear slope between depth and the natural logarithm of the measured radiation.

15 Absorption spectra were obtained between 250 nm and 700 nm at 1 nm intervals in a
16 spectrophotometer (UV-VIS Shimadzu) using 5 cm quartz cuvette and Milli-Q water as
17 a reference. The absorption spectra of each sample was performed in triplicate and
18 determined using the Spectrum Pack Program software (Shimadzu Inc.). The absorption
19 coefficients ($a(\lambda)$) were derived from absorbance measurements ($A(\lambda)$) according to the
20 equation $a(\lambda)\text{m}^{-1} = 2,303A(\lambda)L^{-1}$, where L is the optical path of the cuvette (in meters).
21 Absorption coefficients were corrected for backscattering by subtracting the value of the
22 coefficient at 700 nm. The spectrophotometric absorption coefficient at 350 nm (m^{-1})
23 was used as an index of CDOM concentration (Zhang *et al.*, 2011b).

1 The spectral slope S (nm^{-1}) was calculated for wavelengths between 275-295 nm and
2 350-400nm (Helms *et al.*, 2008) and it was estimated using linear adjustment technique
3 between log values of the absorption coefficients and the specific wavelength range.
4 The slope ratio (S_R) was obtained by the equation $S_R = S_{275-295\text{nm}} / S_{350-400\text{nm}}$, where S is
5 the spectral slope. The estimated apparent molecular size (M) was obtained by the ratio
6 of the absorption coefficients at 250 and 365 nm (a_{250} / a_{365}) (De Haan, 1987;
7 Peuravuori and Pihlaja, 1997). The specific UV absorbance (SUVA_{254}) ($\text{mg C L}^{-1} \text{m}^{-1}$)
8 was calculated dividing the value of the absorption coefficient at 254 nm by the
9 concentration of DOC (mg L^{-1}) (Weishaar *et al.*, 2003).

10 *2.4 Statistical Analysis*

11 Differences in parameters between seasons were assessed by independent sample t-test
12 (for parametric data) and Mann-Whitney (for non parametric data) (significant $p <$
13 0.05), and differences between freshwater systems were assessed with Two-way
14 Analysis of Variance (significant $p < 0.05$). Pearson correlations were made to examine
15 the relationships between all the parameters measured (significant $p < 0.05$).

16 3. RESULTS

17 *3.1 Differences between the two contrasting freshwater systems*

18 The hyper eutrophic Pampulha reservoir had significantly higher concentrations of
19 nutrients (21 times higher for TN and 28 for TP), Chl-a (5.8 times) and DOC (1.4 times)
20 compared to the oligotrophic DH Lake. As Pampulha also had higher CDOM_{350}
21 absorbance (3.3 fold higher), both PAR and UV attenuation was significantly higher
22 than in DH. Indices of CDOM quality or composition furthermore indicated DOM of
23 different origin as Pampulha had DOC of higher specific UV absorbance (SUVA_{254}),
24 smaller spectral slopes ($S_{275-295}$, $S_{350-400}$), lower spectral slope ratio (S_R) and higher

1 molecular size of CDOM (lower M) than DH (Tabs. 1 and 2, Figs. 2a-d and 3a-d).
2 CDOM absorption was generally higher over the entire spectrum in Pampulha,
3 especially in the UV range (Fig. 4a-b).

4 *3.2 Seasonality in optical properties*

5 Concentration of nutrients, Chl-a and DOC were significantly higher in the fully mixed
6 dry season in DH (Fig. 2a-d). The PAR, UV-A and UV-B attenuation were also higher
7 during dry period (Fig. 5a). In this lake, only $SUVA_{254}$ and M values (Fig. 3a) were
8 higher in rainy period. In Pampulha we found that concentrations of TP (Fig. 2b), PAR
9 and UV attenuation (Fig. 5b) were significantly higher during the rainy season. Only
10 Chl-a and TN concentrations were higher in dry period (Fig. 2d) in this reservoir
11 (Tabs.1 and 2).

12 Analyzing the CDOM absorption spectrum at 21 points in DH and at 16 in Pampulha,
13 during a rainy and a dry period we noticed that the spatial variation in each environment
14 was smaller during the rainy period compared with the dry period. For DH, the mean
15 absorption spectrum (black line in Fig. 4a) was higher in the dry period for absorption at
16 wavelengths shorter than 350 nm. For Pampulha the mean absorption (black line in Fig.
17 4b) was equal in both seasons.

18 Significant and positive Pearson correlation was observed to DH between $S_{350-400}$ and
19 nutrients concentrations (TN; $r = 0.52$, $p < 0.015$; TP; $r = 0.82$, $p < 0.000$) only for rainy
20 period data, which indicates that biodegradation might be limited by nutrients in this
21 time. In the Pampulha reservoir, we found significant Pearson correlations between
22 $CDOM_{350}$ and Chl-a in the dry period ($r = 0.88$, $p < 0.000$), when there was an algal
23 bloom. The $CDOM_{350}$ was also positively correlated with the coefficients of attenuation
24 of PAR only in the dry period ($K_{d(PAR)}$; $r = 0.80$, $p < 0.000$) and with UV radiation in

1 both periods (dry period: $K_{d(305)}$; $r = 0.81$, $p < 0.000$; $K_{d(340)}$; $r = 0.90$, $p < 0.000$; rainy
2 period: $K_{d(305)}$; $r = 0.52$, $p = 0.037$; $K_{d(340)}$; $r = 0.76$, $p < 0.000$). In addition, the TSM
3 was positively correlated with Chl-a only in the dry period ($r = 0.91$, $p < 0.000$).

4 4. DISCUSSION

5 4.1 Sources of DOM and underwater light climate

6 The large difference between the quantity and quality of DOM in the two studied
7 freshwater systems indicated a terrestrial input of organic matter in DH Lake occurring
8 mostly during the rainy period in the summer, compared to a strong autochthonous
9 phytoplankton contribution in Pampulha reservoir related to an algal bloom especially
10 evident in the dryer winter period.

11 The differences in the $S_{275-295}$ values between the systems suggest that DOM originated
12 from different sources (Helms *et al.*, 2008). The increase of aromatic DOC (higher
13 SUVA) during rainy period in DH strongly suggests an allochthonous source of DOM
14 from the forested catchment around DH primarily entering the lake during this time. In
15 Pampulha, increases in CDOM absorbance was positively correlated with increasing of
16 Chl-a during an algal bloom in dry winter indicating an autochthonous input of DOM.
17 Other studies have linked DOM source from phytoplankton degradation considering the
18 positive correlations between CDOM absorption and Chl-a (Kahru and Mitchell, 2001;
19 Rochelle-Newall and Fisher, 2002; Zhang *et al.*, 2013b).

20 The DOM source affects some characteristics of lakes, as their transparency to UV and
21 PAR radiation, and recently some studies have compared the differences in optical
22 quality between DOM from autochthonous and allochthonous sources (Spencer *et al.*,
23 2009; Bracchini *et al.*, 2010; Shank *et al.*, 2010; Catalán *et al.*, 2013; Zhang *et al.*,
24 2013). The PAR and UV attenuation in Pampulha were much higher than in DH. In

1 Pampulha we observed that the predominantly autochthonous DOM was coupled to the
2 attenuation of PAR (positive correlation between $CDOM_{350}$ and PAR attenuation
3 coefficients only in dry period) and ultraviolet radiation (positive correlations in dry and
4 rainy period). A similar relation was not found for the allochthonous DOM in DH, and
5 it was already observed for this lake that PAR and UV attenuation were strongly
6 dependent on light absorption by particulate matter (tripton and phytoplankton) while
7 CDOM only contributed with less than 22% in total attenuation during monthly and
8 spatial sampling in 2013 (Brandão *et al.*, in review). However, when the CDOM comes
9 from phytoplankton degradation, as in eutrophic environments like Pampulha, it may
10 greatly increase its contribution to the attenuation of radiation and CDOM absorbance
11 ($CDOM_{350}$ higher in Pampulha) during blooms (Zhang *et al.*, 2007). The different DOM
12 sources in these systems also alter the importance of DOM degradation processes, as
13 allochthonous terrestrial carbon is known to be more humic consisting of heavier and
14 more aromatic molecules being typically more photodegraded. In contrast, the
15 autochthonous microbial/algal derived DOM consisted as expected by simpler
16 molecules of low molecular weight, less affected by photodegradation but more
17 susceptible to biodegradation (McKnight *et al.*, 1994; Benner, 2002; Catalán *et al.*,
18 2013, Zhang *et al.*, 2013b).

19 *4.2 Seasonality of optical properties between the contrasting systems*

20 Seasonal variations were observed for nutrients and Chl-a concentration and UV and
21 PAR attenuation for both systems and also seasonal differences only to DH Lake in
22 relation to the DOC concentration and the quality of CDOM ($SUVA_{254}$ and M).
23 Seasonal changes in the quality and quantity of DOM in DH suggest linkages with
24 rainfall events and water column mixing. In contrast, the reduced seasonality in
25 Pampulha seems attributed to reoccurring algal blooms in this reservoir, which causes

1 several events of autochthonous production of DOM throughout the year, although most
2 prevailing during the well mixed dryer winter period.

3 As we expected, the water column in DH was most transparent (both PAR and UV
4 radiation) during the rainy season when the water column was stratified and with DOM
5 components of a more aromatic structure (higher SUVA) and more susceptible to
6 photobleaching (Weishaar *et al.*, 2003; Helms *et al.*, 2008; Catalán *et al.*, 2013; Zhang
7 *et al.*, 2013). Positive correlations between $S_{350-400}$ and nutrients (TP and TN) only
8 during the rainy season suggest that microbial degradation in the epilimnion might be
9 limited by nutrients during thermal stratification. Nutrient limitation of primary
10 production during the rainy season in DH Lake was observed by Brighenti *et al.* (2015)
11 in a two year study of lake metabolism. In agreement with this Chl-a concentrations
12 were lower in the rainy period in DH. Furthermore, nutrient limitation of phytoplankton
13 growth raises light transparency and is expected to explain elevated levels of
14 photoinhibition during summer (Brighenti *et al.*, 2015). While our results showed that
15 the increase of aromatic DOM originates from surface runoff during rainy period, DOC
16 and nutrients concentrations were lower in the epilimnion in this period, indicating that
17 much of the newly introduced DOM remained in the hypolimnion until later mixing of
18 the water column in the dry and cooler winter period (Reynolds, 2009), when their
19 concentrations increased in the water surface (Barbosa and Tundisi, 1980; Petrucio and
20 Barbosa, 2004). Corroborating with this, we observed DOM characterized by higher
21 molecular size compounds (lower M) during the mixed winter period (Peuravuori and
22 Pihlaja, 1997; Helms *et al.*, 2008). Importantly, most of the sampling in DH Lake was
23 made in areas where total depth exceeded 10 meters thus allowing a stable hypolimnion
24 to develop over longer period of anoxia during thermal stratification from September to

1 May. Such conditions seem favorable for the changes in DOM quality observed after
2 full mixing of the water column.

3 Contrary to what we expected, the Pampulha reservoir was most transparent (both UV
4 and PAR radiation) in the dry mixing winter although it did experience a bloom of
5 cyanobacteria. The bloom in this period can be explained by the higher concentration of
6 nitrogen and the lower water level (approximately 1 meter), and also by the high
7 incidence of solar radiation throughout the year. We expected that an increase of
8 phytoplankton would increase the light attenuation coefficients (K_d), especially the
9 $K_{d(PAR)}$, but the opposite occurred in Pampulha. Although at high concentrations, the
10 autochthonous carbon from phytoplankton degradation tends to be less chromophoric
11 and attenuates radiation less, especially in the UV range (McKnight *et al.*, 1994;
12 Benner, 2002). The concentration of TSM during the dry-mixing period was however,
13 lower than in the rainy season, during which TSM was uncorrelated with Chl-a and
14 accordingly a greater amount of non-algal materials (tripton) occurred in the TSM. In
15 agreement with other studies it therefore seems that tripton must have accounted for
16 most of the PAR attenuation during the rainy stratified summer (Phlips *et al.*, 1995;
17 Christian and Sheng, 2003; Zhang *et al.*, 2007b; Brandão *et al.*, in review).

18 5. CONCLUSIONS

19 As the presence of chromophoric carbon in the water protects the aquatic organisms
20 from harmful ultraviolet radiation and from the excessive incidence of solar radiation
21 (Zhang *et al.*, 2007), our results suggest that the photoinhibition of phytoplankton
22 primary production observed in DH during summer by Brighenti *et al.* (2015) is directly
23 linked with the seasonal pattern of photobleaching of CDOM and nutrients availability,
24 where both dependent of the sources and quality of DOM. Thereby, we demonstrated

1 that the main source of DOM in DH come from the terrestrial input during the rainy
2 period, reinforcing the importance of preserving the original forest surrounding the
3 lakes to the global carbon cycle. In Pampulha reservoir we observed that degradation of
4 phytoplankton is an important source of DOM and most likely also to the regeneration
5 of nutrients, which has already been demonstrated for other eutrophic systems (Zhang *et*
6 *al.*, 2009; Zhang *et al.*, 2013; Zhang *et al.*, 2013b). We also found that autochthonous
7 production of CDOM can increase the PAR and UV attenuation, as observed in the
8 positive correlations between CDOM and attenuation coefficients only in Pampulha and
9 during an algal bloom.

10 In conclusion, our results documents that tropical freshwater systems can have a clear
11 seasonality in carbon quality and concentration related to trophic status and seasonal
12 differences in DOM inputs, degradation and distribution in the water column. Our
13 finding supports recent findings of a strong seasonality in primary production and
14 organic matter decomposition in tropical lakes and has implications on our
15 understanding of PAR and UV attenuation and water quality in such systems.

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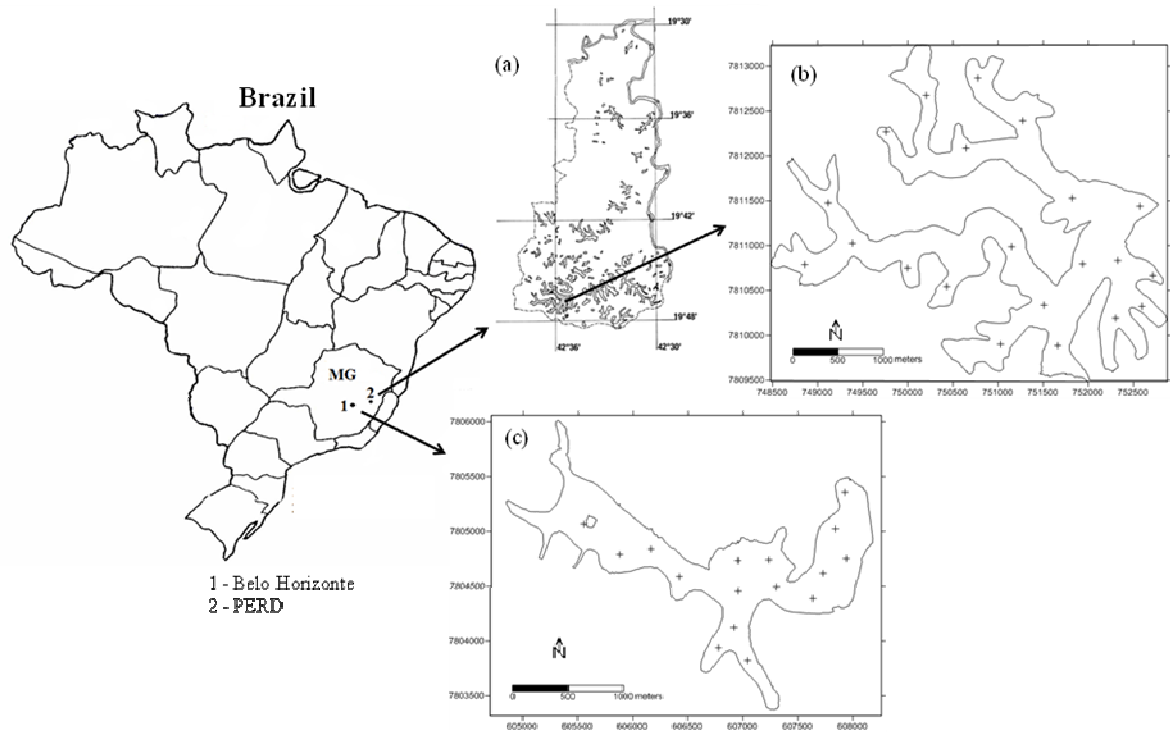
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1 **Figures and Tables**

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4 **Fig. 1** Location at (a) Rio Doce State Park (PERD) and (b) Dom Helvécio lake, and (c)
5 Pampulha reservoir in Belo Horizonte, Minas Gerais (MG), Brazil.

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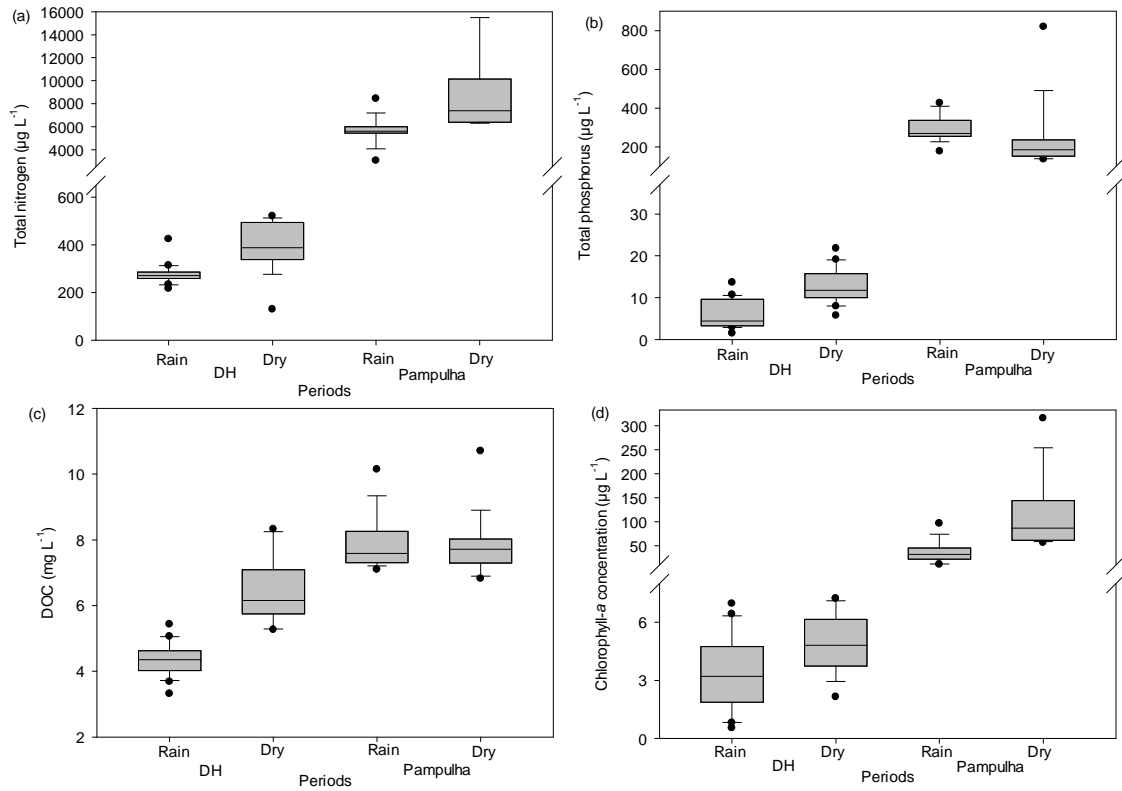
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2 **Fig. 2** Seasonal variations (center line-median, outer grey-5th/95th percentiles, bars-
 3 95% confidence level, black circles- outlier) in the concentrations of (a) total nitrogen,
 4 (b) total phosphorus, (c) dissolved organic carbon and (d) chlorophyll-*a* at 21 sites in
 5 Dom Helvécio lake and 16 in Pampulha reservoir collected in a dry and rainy period in
 6 2013.

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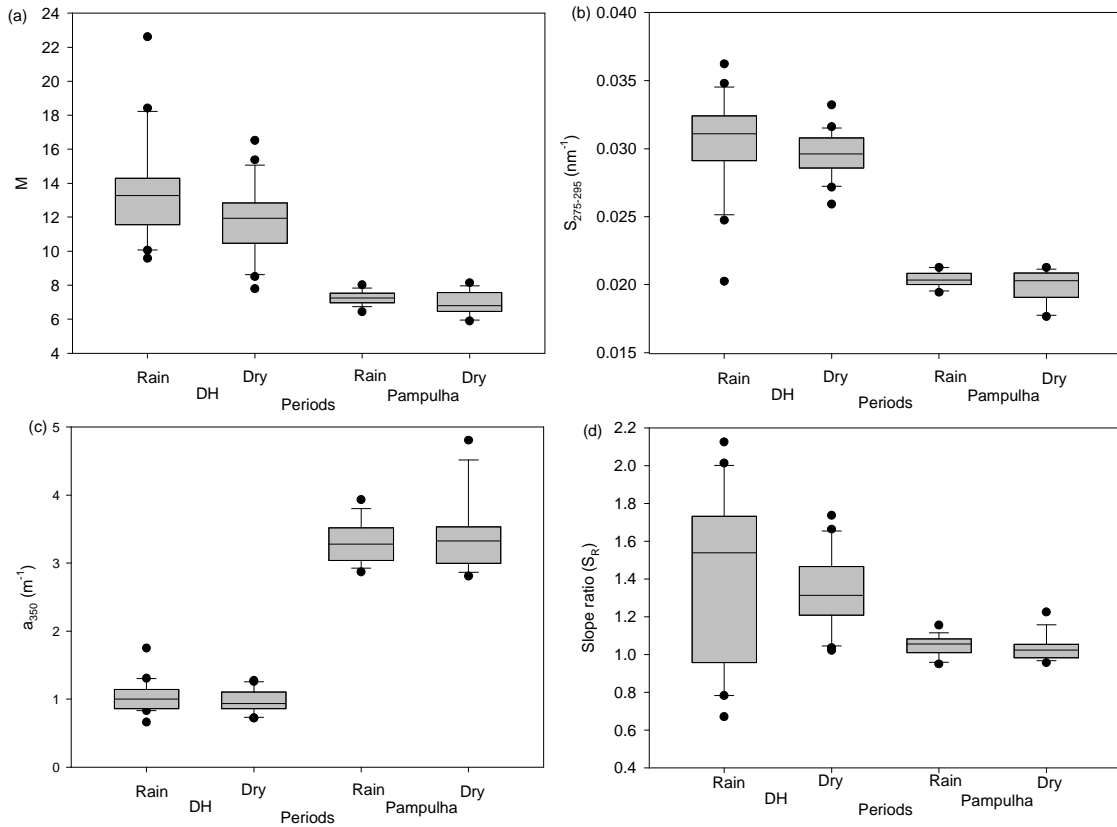
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2 **Fig. 3** Seasonal variations (center line-median, outer grey-5th/95th percentiles, bars-
 3 95% confidence level, black circles- outlier) in (a) M, (b) spectral slope $S_{275-295}$, (c)
 4 absorbance of colored dissolved organic matter at 350 nm (a_{350}) and (d) slope ratio S_R at
 5 21 sites in Dom Helvécio lake and 16 in Pampulha reservoir collected in a dry and rainy
 6 period in 2013.

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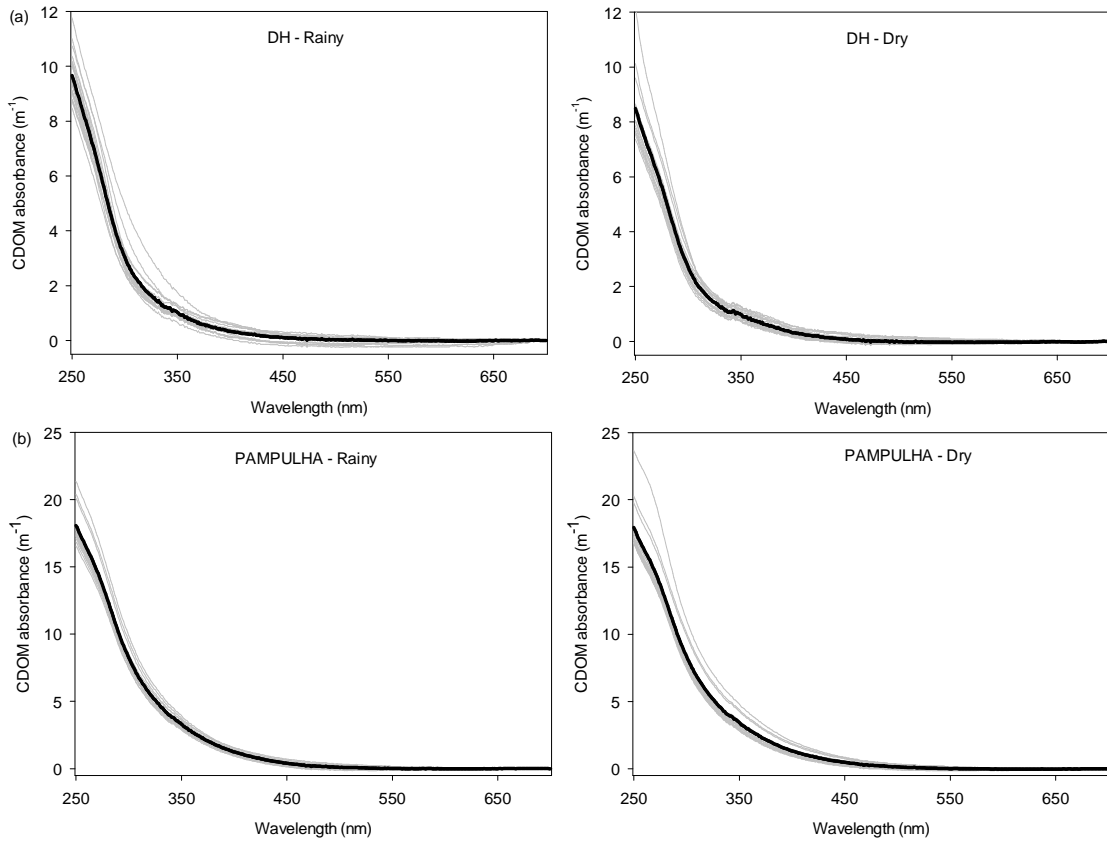
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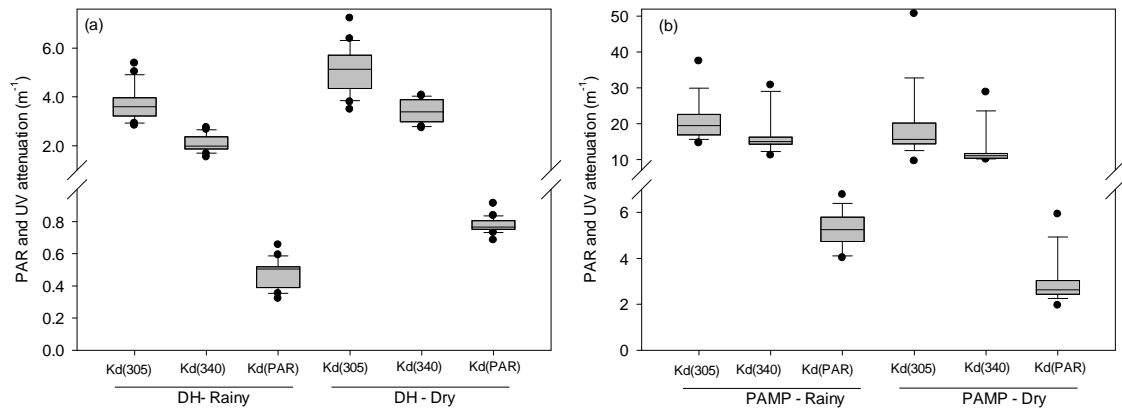
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2 **Fig. 4** – Spectral absorbance curves of CDOM (a) at 21 sampling sites (gray lines) in
 3 Dom Helvécio and the mean curve (black line) and (b) at 16 sampling sites (gray lines)
 4 in Pampulha and the mean curve (black line).

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2 **Fig. 5** - Seasonal variations (center line-median, outer grey-5th/95th percentiles, bars-
 3 95% confidence level, black circles- outlier) in the UV(305), UV (340) and PAR
 4 radiation at (a) 21 sites in Dom Helvécio lake and (b) at 16 in Pampulha reservoir
 5 collected in a dry and rainy period in 2013.

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1 Table 1 - Summary of parameters measured during spatial-seasonal analyzes in Dom
 2 Helvécio lake and Pampulha reservoir, MG, Brazil. (n = number of sampling stations)

		Dom Helvécio lake		Pampulha reservoir	
		Rainy (n=21)	Dry (n=21)	Rainy (n=16)	Dry (n=16)
Surface Temperature (°C)	Min-Max	30.0	25.4	26.0 - 28.3	22.5 - 23.7
	Mean ± SD	30 ± 0	25.4 ± 0	27.4 ± 0.8	23.3 ± 0.5
Water column depth (m)	Min-Max	4.5 - 30.0	2.9 - 30.0	1.0 - 10.0	0.6 - 9.5
	Mean ± SD				
Total suspended matter (mg L ⁻¹)	Min-Max	0.4 - 2.1	0.6 - 2.8	16.7 - 31.2	3.2 - 24.2
	Mean ± SD	1.4 ± 0.4	1.7 ± 0.6	24.7 ± 3.6	7.2 ± 5.6
Total nitrogen (µg L ⁻¹)	Min-Max	215.5 - 424.1	129.0 - 519.7	3000.0 - 8400.0	6300.0 - 15500.0
	Mean ± SD	276.8 ± 41.0	393.2 ± 98.9	5680.0 ± 1100.0	8615.0 ± 3300.0
Total phosphorus (µg L ⁻¹)	Min-Max	1.4-13.6	5.7-21.8	176.3 - 425.9	135.3 - 819.2
	Mean ± SD	6.0 ± 3.4	12.7 ± 4.1	292.3 ± 67.7	232.8 ± 165.5
Chlorophyll a (µg L ⁻¹)	Min-Max	0.5 - 6.9	2.1 - 7.2	10.7 - 96.2	56.1 - 315.4
	Mean ± SD	3.3 ± 1.9	4.8 ± 1.5	35.6 ± 21.4	111.6 ± 71.8
DOC (mg L ⁻¹)	Min-Max	3.3 - 5.4	5.3 - 8.3	7.1 - 10.1	6.8 - 10.7
	Mean ± SD	4.4 ± 0.5	6.5 ± 1.0	7.9 ± 0.8	7.8 ± 0.9
a(350) (m ⁻¹)	Min-Max	0.7 - 1.8	0.7 - 1.3	2.9 - 3.9	2.8 - 4.8
	Mean ± SD	1.0 ± 0.2	1.0 ± 0.2	3.3 ± 0.3	3.4 ± 0.6
Kd(PAR) (m ⁻¹)	Min-Max	0.3 - 0.7	0.7 - 0.9	4.0 - 6.8	1.9 - 5.9
	Mean ± SD	0.5 ± 0.1	0.8 ± 0	5.2 ± 0.8	3.0 ± 1.0
Kd(UV-A 340) (m ⁻¹)	Min-Max	1.5 - 2.7	2.7 - 4.1	11.1 - 30.7	10.0 - 28.8
	Mean ± SD	2.1 ± 0.3	3.4 ± 0.5	16.4 ± 5.3	13.1 ± 5.3
Kd(UV-B 305) (m ⁻¹)	Min-Max	2.8 - 5.4	3.5 - 7.2	14.6 - 37.5	9.5 - 50.7
	Mean ± SD	3.7 ± 0.7	5.1 - 0.9	20.7 ± 5.6	18.6 ± 9.4
S ₂₇₅₋₂₉₅ (nm ⁻¹)	Min-Max	0.02 - 0.04	0.03 - 0.03	0.02 - 0.02	0.02 - 0.02
	Mean ± SD	0.03 ± 0	0.03 ± 0	0.02 ± 0	0.02 ± 0
S ₃₅₀₋₄₀₀ (nm ⁻¹)	Min-Max	0.01 - 0.05	0.02 - 0.03	0.02 - 0.02	0.02 - 0.02
	Mean ± SD	0.02 ± 0.01	0.02 ± 0	0.02 ± 0	0.02 ± 0
S _R	Min-Max	0.7 - 2.1	1.0 - 1.7	1.0 - 1.2	1.0 - 1.2
	Mean ± SD	1.4 ± 0.4	1.3 ± 0.2	1.0 ± 0.1	1.0 ± 0.1
SUVA (mg C L ⁻¹ m ⁻¹)	Min-Max	1.8 - 2.5	0.9 - 1.8	2.0 - 2.4	1.6 - 3.0
	Mean ± SD	2.1 ± 0.2	1.3 ± 0.3	2.2 ± 0.1	2.2 ± 0.3
Molecular size (M)	Min-Max	9.6 - 22.6	7.8 - 16.5	6.4 - 8.0	5.9 - 8.1
	Mean ± SD	13.7 ± 3.1	11.8 ± 2.1	7.3 ± 0.4	7.0 ± 0.7

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1 Table 2 – Results of the two-way analysis of variance comparing the seasons (rainy and
 2 dry) and the freshwater systems (DH = Dom Helvécio; PA = Pampulha) and results of
 3 the *t*-test and Mann-Whitney comparing the seasons within each system.

	Two-way Analysis of Variance (F value) df = 1			t-test / Mann-Whitney	
	Systems (DH x PA)	Seasons (Rainy x Dry)	Seasons and systems interactions	DH (Rainy x Dry)	PA (Rainy x Dry)
Total nitrogen ($\mu\text{g.L}^{-1}$)	425.45***	21.33***	18.20***	*** 2	** 2
Total phosphorus ($\mu\text{g.L}^{-1}$)	169.67***	1.84	2.90	*** 1	*** 2
Chlorophyll a ($\mu\text{g L}^{-1}$)	10.68**	5.25*	5.03*	** 1	*** 2
DOC (mg. L^{-1})	155.63***	25.76***	32.83***	*** 2	ns 2
$a(350)$ (m^{-1})	887.87***	0.26	1.32	ns 2	ns 2
$Kd_{(\text{PAR})}$ (m^{-1})	651.92***	48.44***	85.07***	*** 1	*** 2
$Kd_{(\text{UV-A } 340)}$ (m^{-1})	217.67***	1.53	8.26**	*** 2	** 2
$Kd_{(\text{UV-B } 305)}$ (m^{-1})	163.16***	0.08	2.16	*** 1	* 2
$S_{275-295}$ (nm^{-1})	383.27***	1.85	0.11	ns 2	ns 1
$S_{350-400}$ (nm^{-1})	9.86**	0.21	0.07	ns 2	ns 1
S_R	28.58***	0.63	0.35	ns 1	ns 1
SUVA ($\text{mg C L}^{-1} \text{m}^{-1}$)	4.26***	2.55***	2.89***	*** 1	ns 2
Molecular size (M)	140.91***	5.54*	2.85	** 2	ns 1

* $p < 0.05$ ** $p < 0.01$ *** $p < 0.001$ ns = not significant

1 = parametric data and *t*-test was used

2 = non-parametric data and Mann-Whitney was used

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CAPÍTULO 4

Formatado para ser submetido em: *Biogeochemistry*

Distinctive effects of allochthonous and autochthonous organic matter on colored dissolved organic matter spectra

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Abstract

Despite the increasing understanding about differences in carbon cycling between temperate and tropical freshwater systems, our knowledge on the importance of organic matter (OM) pools on light absorption properties in tropical lakes is very scarce. We performed a mesocosm factorial experiment in a natural tropical lake (MG, Brazil) to evaluate the effects of increased concentrations of allochthonous and autochthonous OM on the light absorption of CDOM. We manipulated inorganic nutrients (to stimulate phytoplankton growth and OM autochthonous production), allochthonous OM and shading. DOC, CDOM, Chl-a, nutrients and total suspended solids (TSM) were measured every 3 days to evaluate how they responded to these treatments. CDOM quality was obtained by spectral indexes ($S_{250-450}$, $S_{275-295}$, $S_{350-450}$, slope ratio- S_R and $SUVA_{254}$). The effect of carbon sources on the CDOM spectra was investigated through

1 principal component analysis (PCA) and a redundancy analysis (RDA). As expected,
2 phytoplankton biomass (Chl-a as a proxy) was stimulated by addition of nutrients and
3 the CDOM quality related to different sources was different between experimental units
4 with addition of nutrients and with allochthonous OM. Spectral indexes ($S_{250-450}$ and S_R)
5 were only affected by the addition of allochthonous OM. The PCA showed that
6 enrichment by allochthonous carbon had a strong effect on the CDOM curve between
7 300 and 400 nm, while the increase of autochthonous carbon increased absorption at
8 shorter wavelengths. In turn, the concentration of OM from different sources was linked
9 to the availability of nutrients and allochthonous OM, which under natural conditions
10 vary seasonally with rainfall and mixing conditions.

11 Keywords: DOC, DOM, mesocosm, carbon source, spectral indexes

12 **Acknowledgments**

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17 scholarship. We would like to thank Gustavo Turci, Ralph Thomé, Patrícia Ferreira and
18 Marcelo Ávila for field support and to Marcelo Costa for nutrient analysis.

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1 **Introduction**

2 Organic matter (OM) is a central theme in ecology and influence many processes from
3 bacterial metabolism to primary production. In aquatic systems OM consists of
4 particulate organic matter (POM; organic compounds represented by aquatic
5 communities and detritus), and dissolved organic matter (DOM – in most of DOM
6 studies it is the compounds smaller than 0.2 or 0.7 μm) which is far the largest pool of
7 organic carbon in most of aquatic systems (Hedges 1992). A better understanding of
8 carbon cycling in aquatic systems and their regional and global importance therefore
9 requires knowledge of the physical, chemical and biological processes involved in the
10 production and the mineralization of DOM (Bertilsson and Tranvik 2000; Johannessen
11 et al. 2007; Tranvik et al. 2009).

12 The main sources of DOM in aquatic ecosystems can be divided into two main pools
13 according to their sources. The allochthonous pool which comes from terrestrial
14 vegetation and soil sources (Kieber et al. 2006; Miller et al. 2009) and the
15 autochthonous pool which is produced mainly by aquatic primary producers (Kritzberg
16 et al. 2004; Guillemette and Del Giorgio 2012). These two pools of DOM have
17 fundamental differences in their optical and chemical characteristics which in turns
18 influence the mechanisms by which DOM is degraded (Wetzel et al. 1995; Bertilsson
19 and Tranvik 2000).

20 The allochthonous DOM is in general more susceptible to photodegradation because it
21 contains aromatic compounds of higher molecular weights which strongly absorb UV
22 light (Amon and Benner 1994; McKnight et al. 1994; Benner 2002; Helms et al. 2008).

23 The autochthonous DOM originating from phytoplankton mainly consists of simple
24 molecules (carbohydrates, proteins, amino acids) of low molecular weight and it is more

1 labile for microbial community (Farjalla et al. 2009; Fonte et al. 2013). However,
2 autochthonous DOM originating from macrophytes or macroalgae can be as aromatic
3 and recalcitrant as DOM from terrestrial vegetation (Catalán et al. 2013).

4 Photochemical and microbial degradation are important processes driving the dynamics
5 of DOM in aquatic ecosystems (Roland et al. 2010; Mopper et al. 2015). The rate of
6 photodegradation depends on a combination of available sunlight and the chemical
7 quality of DOM (Benner 2002) whereas the microbial degradation rate depends on
8 DOM bioavailability and utilization efficiency of the bacterial community (Catalán et
9 al. 2013; Asmala et al. 2014). Photodegradation is also an important process
10 transforming DOM to ammonia and other highly bioavailable inorganic compounds
11 (Aarnos et al. 2012) which can be an important nutrient supply for both phytoplankton
12 (Hessen and Tranvik 1998) and heterotrophic bacterial communities (Kieber et al. 1989;
13 Miller et al. 2002; Lønborg et al. 2010). Exposure to light transforms DOM into smaller
14 molecules and depending on its origin the photoreactions can reduce (autochthonous
15 DOM) or increase (allochthonous DOM) bioavailability (Pérez and Sommaruga 2007;
16 Vahatalo and Wetzel 2008; Catalán et al. 2013). The mineralization of CDOM also
17 causes direct CO₂ evasion to atmosphere influencing the global carbon cycle (Arrigo et
18 al. 2011) and it is an important parameter to the management of water quality and
19 eutrophication control (Zhang et al. 2011).

20 Microbial degradation is another pathway by which heterotrophic organisms convert
21 DOM in to POM which can be assimilated by protozooplankton via the microbial loop
22 (Azam et al. 1983). However, the ability of bacteria to degrade recalcitrant DOM is
23 expected to increase with availability of labile molecules due to the “priming effect”
24 phenomenon (Bianchi 2011), although some studies have contradicted this (Catalán et
25 al. 2015). Additionally, biodegradation is stimulated by inorganic nutrients, mainly

1 nitrogen and phosphorus, which increase the bacterial growth efficiency (Zweifel et al.
2 1995; Asmala et al. 2013) by reducing the energetic cost of nutrient acquisition
3 (Hopkinson et al. 1998).

4 Some recent studies have demonstrated that the effect of photodegradation and
5 biodegradation on the absorption properties of CDOM was not constant over the
6 spectral range, thus influencing the shape of the absorption curve (Helms et al. 2013;
7 Reader et al. 2015). As modifications in the spectral shape reflect underlying changes in
8 the carbon compounds at the molecular level, studies on biological and chemical effects
9 on CDOM spectra allows a better understanding of the DOM transformations and how
10 this links to overall carbon cycling in aquatic ecosystems (Stubbins et al. 2014).

11 Tropical regions are subjected to intense solar radiation and high temperatures
12 throughout the year, which accelerates the biodegradation and photodegradation
13 processes. However, there are only few studies on the photochemical (Teixeira et al.
14 2013; Bittar et al. 2015) and bacterial (Farjalla et al. 2002; 2009; Roland et al. 2010)
15 degradation of DOM in tropical environments compared to temperate freshwater
16 systems and estuaries (Anesio and Granéli 2003; Bertilsson and Tranvik 2000; Boreen
17 et al. 2008; Asmala et al. 2014; Attermeyer et al. 2015).

18 In this study, we performed a mesocosms experiment in the tropical lake, manipulating
19 nutrients, OM and light conditions, in order to answer the following main question:
20 what are the effects of increased inputs of allochthonous and autochthonous DOM on
21 changes in lacustrine CDOM spectra? Also, we evaluated the effects of
22 photodegradation on the different sources of DOM by manipulating the availability of
23 sunlight. We expect that addition of nutrients causes algal growth, which increases the
24 production of autochthonous DOM (Schindler 1977; Lean and Pick 1981) until nutrients

1 become exhausted. In contrast, the addition of extracted organic matter from leaves of
2 the native forest surrounding the lake is done to increase the allochthonous DOM
3 concentration. Variable levels of OM with different origin are finally expected to
4 change the CDOM absorption spectra and the associated indexes over time. Moreover,
5 we expect that the light reduction will decrease the DOM degradation because of the
6 reduction of photodegradation.

7 Methods

8 *Study area*

9 This study was conducted in Carioca Lake (19°45'26.0"S; 42°37'06.2"W) located in the
10 southern part of the State Park of Rio Doce (PERD, Minas Gerais, Brazil) which is the
11 largest remnant of the Atlantic Forest in Minas Gerais (36000 ha forest) with lakes
12 occupying 9.8% (3530 ha) of its total area. This lake is surrounded by secondary
13 Atlantic Forest and is a warm-monomictic lake with a mixing period during the dry
14 winter (May to August) and thermal stratification during the rest of the year (September
15 to April). It is a mesotrophic lake (total phosphorus: 5.6-21.4 $\mu\text{g L}^{-1}$, epilimnion annual
16 average 13.9 $\mu\text{g L}^{-1}$; chlorophyll-a: 1.3-16.6 $\mu\text{g L}^{-1}$, epilimnion annual average 7.7 $\mu\text{g L}^{-1}$)
17 (Petrucio et al. 2006; Brighenti 2014), with 1718.6 m of perimeter, 14.1 ha, 671×10^3
18 m^3 , and maximum depth of 11.8 and average depth of 4.8 m (Bezerra-Neto et al. 2010).
19 Carioca is one of the lakes that have been monitored for water quality and aquatic biota
20 since 2000 through the Brasil-LTER Programme (PELD-CNPq Proc. 403698/2012-0).
21 It is known that the input of DOM and nutrients in Carioca Lake occurs via runoff
22 during the rainy season, but most of them remain below the thermocline due to
23 temperature differences (Reynolds 2009). During the mixing period, DOC, CDOM and
24 nutrients become distributed in the water column, increasing their concentrations and

1 availability in the surface layer, thus increasing primary production and respiration rates
2 (Brighenti et al. 2015).

3 *Experimental design and measurements*

4 To test the effect of organic matter inputs, sunlight, and nutrients on DOM degradation,
5 we conducted an *in situ* experiment using a total of 16 cylindrical mesocosms (diameter
6 1.3 m, height 1.5 m and volume 2 m³) with eight different combinations (two replicates
7 for each combination). The sampling has been carried out between January 20th and
8 February 1st 2015 with daily measurements occurring between 10:00 am and 12:00 pm.
9 Water samples (3L at 0.5 m from the surface) from mesocosms were collected every
10 three days. The mesocosms setup was based on a 2³ factorial design as follows: 1) with
11 and without addition of organic matter extracted from leaves surrounding the lake (see
12 next paragraph), 2) with and without addition of nutrients (NaNO₃, K₂HPO₄, NH₄Cl)
13 and with and without 50% of shading of solar radiation (Figure 1).

14 The organic matter added in some mesocosms was obtained as follows: leaves of the
15 native vegetation present in the soil around the lake were placed in buckets with
16 distilled water for decomposition and disposed in the laboratory under room
17 temperature (~ 25 °C). After a week this water was filtered at 20 µm, the concentration
18 of DOC was measured using TOC analyzer and then a known concentration of DOC
19 was added in the mesocosms in order to increase DOC concentrations with an organic
20 matter from allochthonous origin.

21 Mesocosms were placed at the surface of the lake and filled with lake water.
22 Mesocosms with light reduction (SH) were shaded with spectrally neutral shading
23 screens (50%) and only opened for quick samplings and measurements. Every day, the

1 mesocosms were gently stirred and measured for water temperature using a probe
2 Hydrolab DS5 (Hach Inc.).

3 Water samples were filtered immediately after sampling for Chl-a and nutrients (0.47
4 μm filter) and TSM (AP040 filter). The filters were kept frozen until analysis. Water
5 samples were also filtered for analysis of DOC and CDOM (0.22 μm Millipore glass
6 microfiber filter) and stored in amber glass bottles (pre-washed with distilled water and
7 hydrochloric acid 10%) at 4°C in the dark. The Chl-a concentration corrected by
8 pheophytin ($\mu\text{g L}^{-1}$) was obtained by acetone extraction (90%) measured in a
9 spectrophotometer (UV-VIS Shimadzu) at 665 and 750 nm and calculated using the
10 protocol provided in APHA (1998). The TSM (mg L^{-1}) were determined by the
11 gravimetric method, considering the difference between the dry weights of AP40
12 Millipore filters (105 °C for 2 hours) before and after the filtration of water samples
13 (APHA 1998). The DOC concentration (mg L^{-1}) was obtained by catalytic oxidation
14 method of high temperature using TOC Analyzer (Shimadzu TOC – 5000A). Filtered
15 water samples were taken for dissolved nutrients (ammonia, nitrate, nitrite and
16 phosphate; $\mu\text{g L}^{-1}$) and frozen until analyzes with an auto-analyzer (Metrohm 8000 IC-
17 Plus).

18 *CDOM optical properties*

19 Absorption spectra of CDOM were obtained between 250 and 700 nm at 1 nm intervals
20 with a spectrophotometer (UV-VIS Shimadzu) using 5 cm quartz cuvette and a Milli-Q
21 water sample as blank reference. The absorption spectra of each sample were measured
22 in replicate (standard deviation < 0.01). The absorption coefficients ($a_{\text{CDOM}}(\lambda)$; m^{-1})
23 were derived from absorbance measurements according to the equation $a_{\text{CDOM}}(\lambda) =$
24 $2.303A(\lambda)L^{-1}$, where $A(\lambda)$ is the absorbance measured at wavelength λ and L is the

1 optical path of the cuvette (in meters). Absorption coefficients were corrected for
2 backscattering by subtracting the value of the coefficient at 700 nm. The absorption
3 coefficient at 254 nm ($a_{\text{CDOM}254}$) was used as an index of CDOM UV-absorption and at
4 440 nm ($a_{\text{CDOM}440}$) as a CDOM PAR-absorption.

5 We used a simple exponential curve to model the decrease in absorption with increasing
6 wavelength using the equation (Jerlov 1968; Bricaud et al. 1981; Stedmon and
7 Markager 2001): $a_{\text{CDOM}}(\lambda) = a_{\text{CDOM}}(\lambda_0) e^{-S(\lambda - \lambda_0)} + K$, where a_{CDOM} is the absorption
8 coefficient (m^{-1}), λ is the wavelength (nm), λ_0 is a reference wavelength (nm), K is a
9 background constant (m^{-1}) accounting for scatter in the cuvette and drift of the
10 instrument and S is the spectral slope (nm^{-1}) that describes the approximate exponential
11 rate of decrease absorption with increasing wavelength. Furthermore, we calculated the
12 spectral slope between 275-295 nm ($S_{275-295}$) and 350-400 nm ($S_{350-400}$). The slope ratio
13 (S_R , Helms et al. 2008) was obtained by dividing $S_{275-295\text{nm}}$ by $S_{350-400\text{nm}}$. These metrics
14 have been calculated using the *cdom* R package (Massicotte 2016). We also calculated
15 $S_{250-450}$ as this wavelength to be used in the principal component analysis, and we
16 created this new index to measure changes in the entire spectrum including UV and
17 PAR-absorbing (we limit to 400 nm because after this wavelength the noise reading was
18 high). The specific UV absorbance (SUVA_{254}) ($\text{m}^2 \text{mg}^{-1} \text{C}$) was calculated dividing the
19 value of the absorption coefficient at 254 nm (m^{-1}) by the concentration of DOC (mg L^{-1})
20 ($\text{Weishaar et al. 2003}$).

21 *Data analysis*

22 Relative changes (%) of the parameters over time were calculated dividing the value at
23 the end by the value at the beginning (day 0) of the experiment, after subtracting this
24 result from 1 and multiplying by 100 [(1 - end/start) x 100]. Negative values of relative

1 changes indicate decrease and positive values indicate increase compared to initial
2 values.

3 To compare if there were differences in the quantity ($a_{\text{CDOM}254}$, $a_{\text{CDOM}440}$ and DOC) and
4 quality (SUVA₂₅₄ and spectral slopes) of DOM between treatments with and without
5 additions of OM we used *t*-test (for the parametric data) and Mann-Whitney (for non
6 parametric data) (significant $p < 0.05$). To compare if there was difference in
7 phytoplankton density (measured by Chl-a concentration) between treatments with and
8 without nutrients we used *t*-test (significant $p < 0.05$). Differences between the
9 treatments with and without shading were also tested by *t*-test for each parameter
10 measured (significant $p < 0.05$).

11 A principal component analysis (PCA) was generated using absorption spectra on a $n \times$
12 p matrix where n is the number of observation in the dataset ($n = 80$) and p is the
13 wavelength number ($250 \leq p \leq 450$). The PCA was performed on scaled data (0 mean
14 and unit variance) as suggested by Borcard et al. (2011). Based on the Kaiser-Guttman
15 criterion (Kaiser 1960), the first two principal components were kept for subsequent
16 analyzes. PCA scores of first two axes (PC1 and PC2) were correlated against
17 environmental variables (shade, nutrients, OM, Chl-a, TSM, DOC, S_R and SUVA₂₅₄)
18 using a redundancy analysis (RDA) to help to understand their interpretation. All
19 statistical analyzes were performed using R Software (R core development team 2011).

20 Results

21 *Overall characteristics of the treatments*

22 The water temperature of the mesocosms ranged between 28.4 and 31.3 °C (average
23 30.8 °C) throughout the experiment. As hypothesized Chl-a levels were significantly
24 higher for treatments with addition of nutrients (Table 1). Concentrations of DOC,

1 $a_{\text{CDOM}254}$ and $a_{\text{CDOM}440}$ were significantly higher in all the treatments with addition of
2 OM. Higher molecular size (lower $S_{275-295}$) and more aromatics compounds (higher
3 SUVA_{254}) were observed also in the treatments with OM added as expected due to the
4 addition of allochthonous DOM derived from the degradation of forest leaves
5 surrounding the lake (Table 1). There were no significant differences between
6 treatments with and without shading to any of the parameters above (t -test, $p > 0.05$).

7 *Temporal changes in the mesocosms*

8 Phytoplankton biomass (Chl-a) increased over time only in the treatments with addition
9 of nutrients (from average of 3.9 in the day 0 to 19.1 $\mu\text{g L}^{-1}$ in the 12th day) and OM
10 (from minimum average 3.4 in day 0 to maximum of 12.1 $\mu\text{g L}^{-1}$ in 9th day) (Figure 2A-
11 B). DOC increased in the experimental units with added nutrients until day 9 (maximum
12 8.7 mg L^{-1}). In the treatments with addition of OM, DOC was higher as expected with a
13 minimum of 8.3 mg L^{-1} in 3th day and maximum of 8.9 mg L^{-1} in 9th day, while in the
14 treatments without addition of OM the DOC had a maximum of 8.2 mg L^{-1} in 6th day
15 (Figure 2C-D). $S_{250-450}$ and slope ratio (S_R) had the same pattern in the treatments with
16 and without nutrients addition, but decreased consistently in treatments with OM
17 addition (Figure 2 E-H).

18 The relative changes in CDOM absorption along the spectral range were different for
19 each sampling day (Figure 3A-E). On the initial day, only treatments with and without
20 addition of OM had distinct absorption curves, especially in the UV range below 400
21 nm, and the absorption spectra for each treatment group in the day 0 are shown in
22 Figure 3A. To evaluate treatment effects we determined the change in light absorption
23 spectra for the other sampling days relative to the initial day (Figure 3B-E). On the 3rd
24 day, treatments with and without nutrients added were quite similar, while those with
25 and without OM and with and without shading showed opposing changes. Loss of

1 absorption occurred only in treatments with full light (less than 5% between 300-420
2 nm) and in those without OM addition (the loss of absorption increased with the
3 increase of wavelength) (Figure 3B). On the 6th day all treatments showed an increased
4 absorption especially after 350 nm (higher increase with shade: ~ 40% at 450 nm),
5 except the treatments with full light that still had a loss of absorption (Figure 3C). After
6 the 9th day, all treatments had a loss of CDOM absorption with increasing wavelengths,
7 especially for the full light treatment (Figure 3D-E).

8 The concentrations of DIP and DIN ($\mu\text{g L}^{-1}$) and the molar ratio between them
9 (DIN:DIP) were higher in treatments with added nutrients as expected (Figure 4A-C). In
10 the treatments without nutrients addition, DIP had concentrations between 3 and 9 $\mu\text{g L}^{-1}$
11 and DIN between 1.7 and 100.4 $\mu\text{g L}^{-1}$, with DIN:DIP ratios ranging between 7.6 (12th
12 day) and 21.3 (day 0) and below 13.9 after 3rd day indicating that phytoplankton
13 community were mostly limited by nitrogen in these treatments (Redfield 1958;
14 Reynolds 1999). All the dissolved nutrients decreased along the days in treatments with
15 nutrients addition (DIP decreased from an average of 153 to 59 $\mu\text{g L}^{-1}$ and DIN from
16 2600 to 400 $\mu\text{g L}^{-1}$). The DIN:DIP molar ratio ranged from 55.3 (3rd day) to 12.2 (12th
17 day) in the units with addition of nutrients.

18 *Principal component analysis results*

19 The first principal component of the redundancy analysis (Figure 5A) was mostly
20 associated with availability of OM. Samples presenting high score on the first principal
21 component furthermore tend to have high values of DOC and SUVA₂₅₄ but low values
22 of S_R. High scores on the second principal component were correlated positively with
23 Chl-a, nutrients and TSM and negatively with shading.

24 Exploration of spectral PCA loadings (Figure 5B-C) revealed that principal component 1

1 (PC1) had the strongest effect on the shape of CDOM absorbance between 300 and 400
2 nm. Principal component 2 (PC2) loadings showed a quasi-linear decrease with
3 increasing wavelength suggesting that phytoplankton enrichment had a stronger effect at
4 lower wavelengths. Furthermore, loading values were negative after ~340 nm,
5 indicating that phytoplankton was on average lowering CDOM absorption after this
6 threshold. Based on RDA, PC1 was renamed “allochthonous carbon enrichment”
7 whereas PC2 was renamed “autochthonous carbon enrichment”.

8 **Discussion**

9 *Temporal responses in DOM and Chl-a*

10 Our study supports previous studies showing that addition of allochthonous matter
11 causes DOM to be dominated by more aromatic carbon and with higher molecular
12 weight (Bertilsson and Tranvik 2000; Benner 2002) and lower spectral slopes (Helms et
13 al. 2008; Figure 2F,H). Addition of nutrients also affected DOM quantity and quality
14 related to autochthonous production of DOM by phytoplankton growth, and it is known
15 that phytoplankton in high densities can be an important source of DOM (Zhang et al.
16 2009; Zhang et al. 2013; Brandão et al. 2016). In the treatments without addition of
17 nutrients the phytoplankton community was limited by nitrogen since the beginning of
18 the experiment (DIN concentrations below 100 $\mu\text{g L}^{-1}$; Reynolds 1999). The DIN:DIP
19 molar ratio in these treatments indicates that nitrogen was consumed very fast and was
20 limited after the 3rd day (ratio below 13.9; Redfield 1958). With the addition of
21 nutrients, the concentrations of DIN and DIP were higher than that considered a limited
22 condition proposed by Reynolds (1999), but the molar ratio values proposed by
23 Redfield (1958) suggested limitation by phosphorus (molar ratios between 21.3 and
24 55.3) during the experiment and only in the 12th day became limited by nitrogen again

1 (molar ratio 12.2). Moreover, these suggestive limitation indicated by molar ratio values
2 did not restrict the phytoplankton growth in the treatments with addition of nutrients. In
3 addition to increased nutrient availability (mainly nitrogen and phosphate) and
4 phytoplankton growth (Schindler 1977; Lean and Pick 1981), the increase of
5 allochthonous OM also caused an increase in Chl-a (Figure 2B). This suggest
6 development and prevalence of mixotrophic phytoplankton species, present in Carioca
7 Lake (Barros et al. 2010) which are known to be favored by organic matter (Cohen and
8 Post 1993). Additionally, the light attenuation caused by enhanced light absorption from
9 added allochthonous OM (Kirk 1994), may have favored growth of phytoplankton by
10 reducing photoinhibition, known to occur in Lake Carioca (Brighenti et al. 2015). The
11 increase of phytoplankton in these treatments therefore suggest that algal growth was
12 stimulated by a combination of increases in nutrients availability due to degradation of
13 the OM added (Hessen and Tranvik 1998) as well as a shift towards mixotrophic
14 phytoplankton species.

15 Although additions of allochthonous OM and nutrients both contributed to increase
16 DOC concentrations, divergent effects of these additions were evident in the quality of
17 carbon assessed by optical indexes ($S_{250-450}$ and S_R). $S_{250-450}$ and S_R decreased
18 significantly after addition of allochthonous OM (Figure 2F, H). The decrease in the
19 slope $S_{250-450}$ occurred due to the increase of a higher molecular weight carbon, which
20 lowered the values of $S_{275-295}$ and consequently of S_R (Helms et al. 2008). Addition of
21 nutrients, however, had little effect on these metrics, which we interpret as a
22 consequence of autochthonous production of DOM. Thus, both spectral indexes were
23 more influenced by allochthonous contribution than by the produced autochthonous
24 DOM. This is likely because these indices derive from slope intervals in the ultraviolet
25 range (250-400 nm) known to be influenced by carbon with higher molecular weight

1 and aromatic compounds capable of absorbing energy at shorter wavelengths
2 (Bertilsson and Tranvik 2000; Benner 2002; Helms et al. 2008).

3 Manipulations with nutrients, allochthonous OM and shading caused visual changes in
4 the spectral curves of CDOM over the sampling days (Figure 3). Several studies have
5 shown that aromatic carbon, typically of terrestrial origin, has a higher absorption in the
6 ultraviolet range (Bertilsson and Tranvik 2000; Benner 2002; Helms et al. 2008). This
7 explains the initial (day 0) effects of allochthonous OM addition on elevated CDOM
8 absorption primarily below 350 nm (Figure 3A). We interpret the following (day 3 and
9 6) increase in the CDOM absorption (especially above 350 nm) for most treatments to
10 result from autochthonous DOM related to phytoplankton growth. Increases in
11 absorption in the PAR range (Figure 3B-C), is known to be related to increases in
12 carbon of algal origin, composed of labile molecules of simple structures and with low
13 molecular weight (Amon and Benner 1994; McKnight et al. 1994; Benner 2002; Helms
14 et al. 2008). The relative changes in the spectral curves shown in Figure 3 therefore
15 reflect the final result of gains by autochthonous production and the loss of absorption
16 by photodegradation and/or biodegradation, although these processes were not directly
17 investigated in this work. We could only estimate the importance of photodegradation in
18 this lake comparing the treatments exposed to full light (orange solid lines in Figure 3),
19 which were the ones that lost absorption in all sampling days, and the shaded treatments
20 (orange dashed lines), which were the ones that less had lost of absorption in the
21 spectra. After the 9th day (Figure 3D-E) the absorption loss was larger than the gain by
22 the autochthonous production in all treatments. Such spectral changes with loss of
23 absorption with increasing wavelength, was described by Asmala et al. (2014) as a
24 typical biodegradation curve. Thus, we believe that after the 9th experimental day the
25 observed loss of DOM was mostly caused by biodegradation.

1 *Effect of allochthonous and autochthonous DOM on CDOM spectra*

2 The results obtained from the PCA and RDA demonstrated that the increase of
3 allochthonous OM had greater effect between the wavelengths 300 and 400 nm,
4 increasing the absorption in this range (PC1, Figure 5B). Several studies have shown
5 that photodegradation is more pronounced at shorter wavelengths (300-400 nm) due to
6 absorption of aromatic carbon compounds (Helms et al. 2008; Helms et al. 2013)
7 typically related to degradation of either terrestrial vegetation (Bertilsson and Tranvik
8 2000; Benner 2002; Helms et al. 2008) or aquatic macrophytes (Catalán et al. 2013). It
9 is possible to assess the importance of photodegradation on CDOM absorption through the
10 spectral slope between 275 and 295 nm ($S_{275-295}$) proposed by Helms et al. (2008). We
11 noticed a decrease in the CDOM absorption below 300 nm, suggesting a greater
12 degradation by photodegradation in these compounds from allochthonous origin
13 affecting the absorption at shorter wavelengths and increasing the absorption between
14 300 and 400 nm.

15 In contrast, the increase of autochthonous OM coming from the high phytoplankton
16 growth caused an increase of absorption in the UV range and a loss of absorption at
17 wavelengths beyond 350 nm (PC2; Figure 5C). The loss of absorption above 350 nm
18 indicates degradation by microorganisms which have greater impact on the PAR
19 absorption. Substances that absorb in this range are typically non-aromatic compounds
20 originating from algal source with high lability for bacterial degradation (Baines and
21 Pace 1990; Berggren et al. 2009).

22 *Conclusions*

23 Additions of terrestrial OM and inorganic nutrients to a tropical lake mesocosms caused
24 fast changes in the production and transformation of OM pools as well as distinct

1 changes in the absorption spectra of CDOM. The factorial design of the mesocosms
2 experiment simulated seasonal changes commonly observed related to changes in
3 primary production and water clarity caused by changes in temperature, stratification,
4 and rains which affect the input of terrestrial OM and nutrients (Brighenti et al. 2015;
5 Brandão et al. 2016). Increased production of autochthonous OM caused an increase of
6 CDOM absorption in the UV range and a reduction in the PAR range, which we
7 interpret to result primarily from bacterial degradation of highly labile algal material
8 (Baines and Pace 1990; Berggren et al. 2009). In contrast, the additions of
9 allochthonous OM caused increased absorption of CDOM, especially between 300 and
10 400 nm. The new index created $S_{250-450}$ was an effective tool to evaluate the spectral
11 changes in general from short to long wavelengths, not restricting spectrum ranges
12 where we can only perceive changes in some regions of UV-absorbing ($S_{275-295}$, S_{350-}
13 400). Although the non-shaded treatments showed apparent effects of photodegradation,
14 changes in CDOM absorption curves after the 9th day suggest that biodegradation was
15 overall responsible for the largest percentage of OM degradation in these experiments.
16 The 50% reduction of sunlight in some mesocosms accordingly had minor effects on
17 overall changes in OM concentration and composition, likely due to the large amount of
18 light still reaching the DOM in the shaded units.

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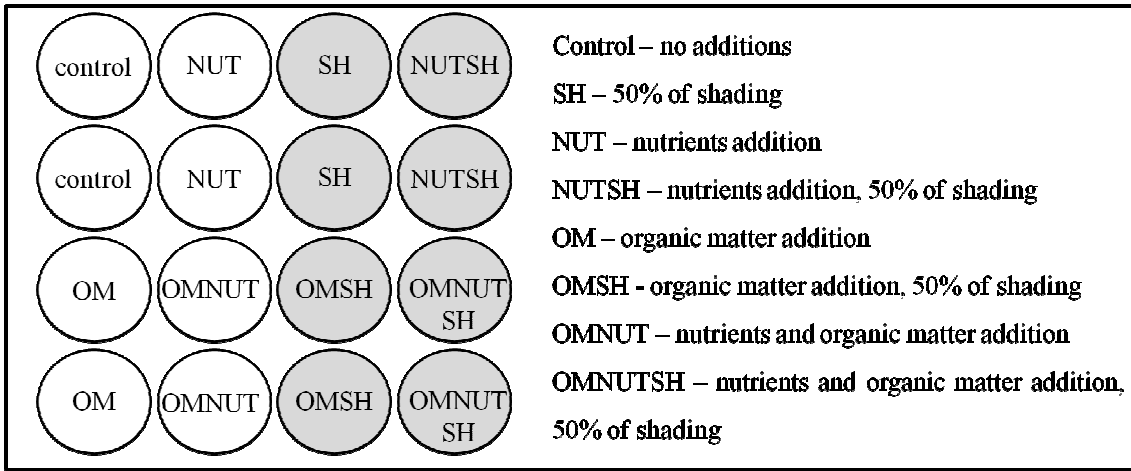
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17 **Figures and Tables**



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2 **Fig. 1** Schematic figure of the factorial design of mesocosms experiments

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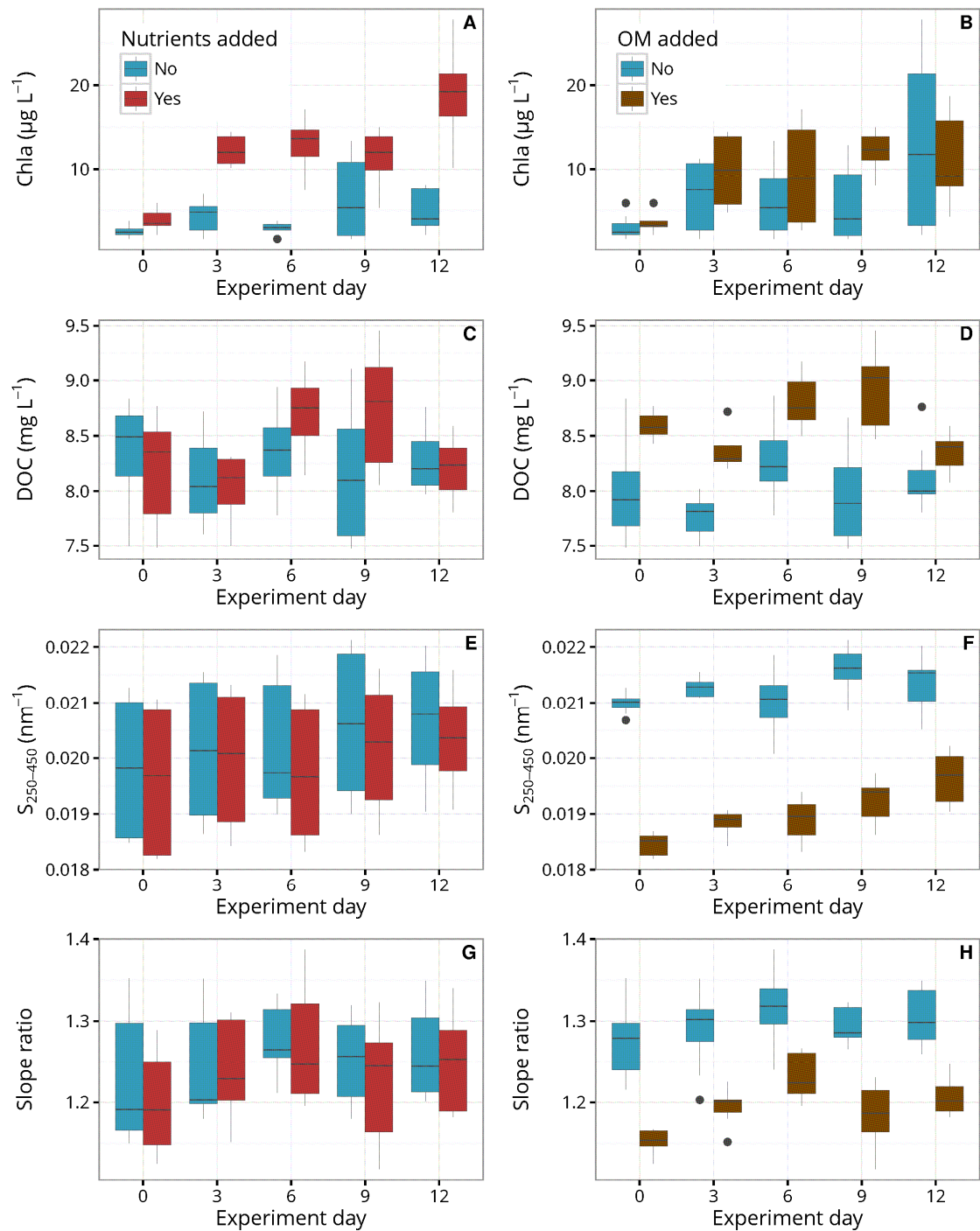
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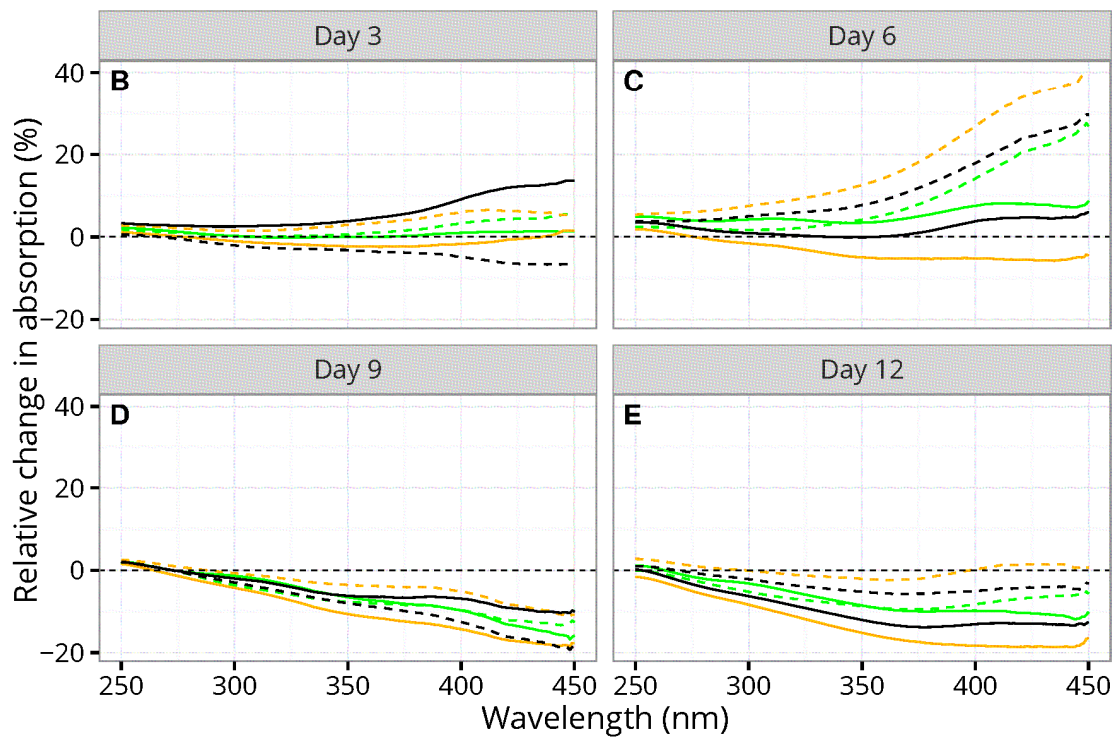
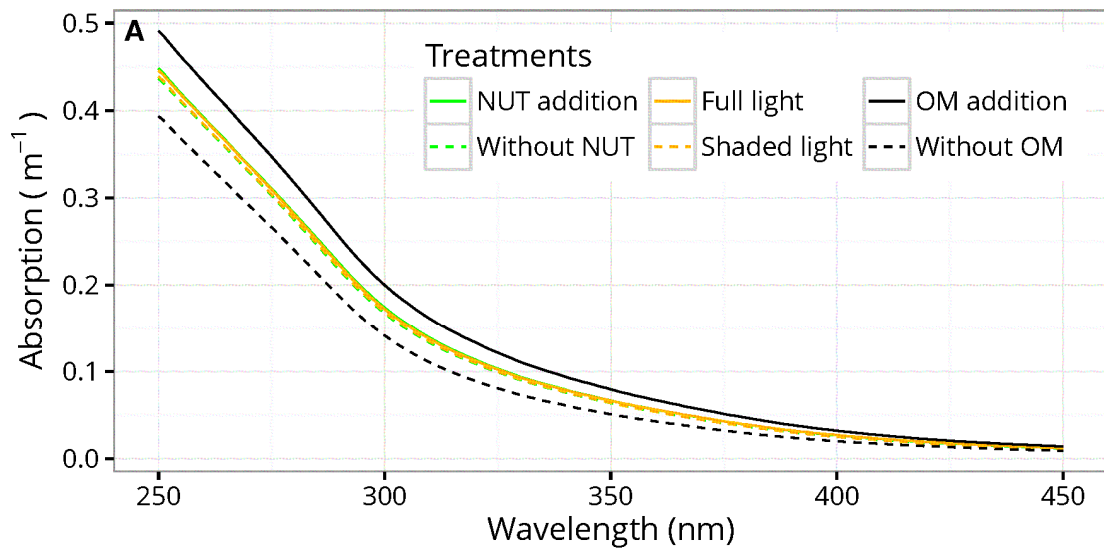
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2 **Fig. 2** Temporal variation in the mesocosms units with and without nutrients (left panel)

3 and organic matter additions (right panel) for Chl-a (A,B), DOC (C,D), $S_{250-450}$ (E,F), S_R

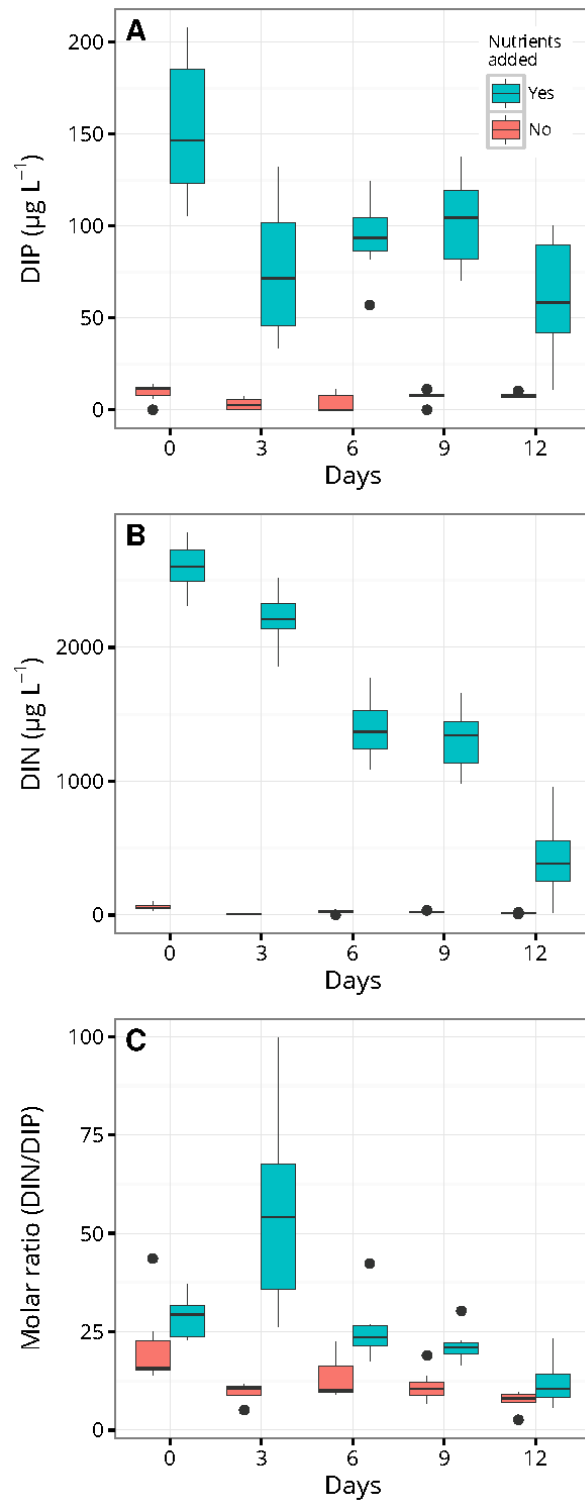
4 (G,H) (center line-median, 5th/95th percentiles, 95% confidence level, black circles-

5 outlier)

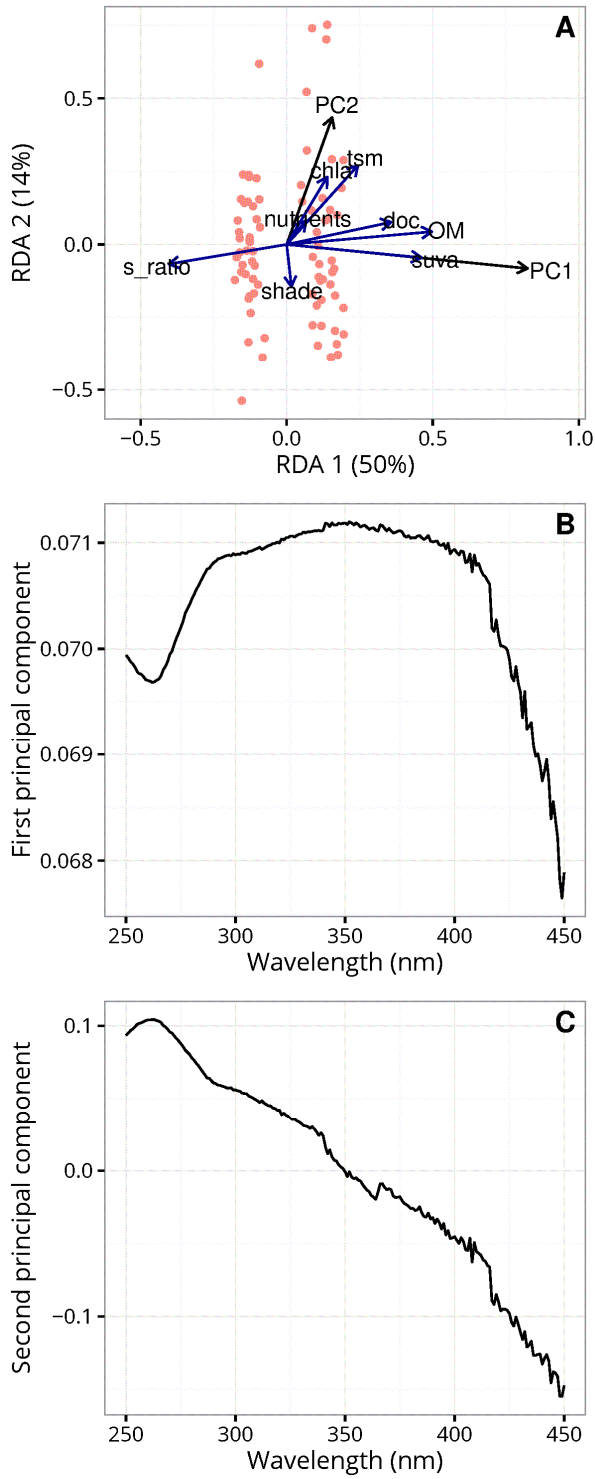


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 2 **Fig. 3** Spectral absorption curves of CDOM in the different mesocosms units for the
 3 initial day (A) and the relative changes for each sampling day compared with the initial
 4 day (3rd day – B, 6th day – C, 9th day – D, 12th day – E)

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 2 **Fig. 4** Temporal variation in the mesocosms units with and without nutrients additions
 3 for dissolved inorganic phosphorus-DIP (A), dissolved inorganic nitrogen-DIN (B) and
 4 the ratio DIN/DIP (C) (center line-median, 5th/95th percentiles, 95% confidence level,
 5 black circles-outlier)



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 2 **Fig. 5** Results of redundancy analysis-RDA (A) and the first two principal components
 3 obtained from PCA analysis, PC1 (B) and PC2 (C), plotted against wavelengths

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1 Table 1 - Results of the *t*-test and Mann-Whitney (legend: OM = treatments with
 2 organic matter addition; no OM = treatments without addition of organic matter; NUT =
 3 treatments with nutrients addition; no NUT = treatments without nutrients addition).

Parameters	Analysis		significance p<0.05	
a254 (m ⁻¹)	Mann-Whitney	Median	U	p
OM		21.857	0	<0.001
no OM		17.344		
a440 (m ⁻¹)	Mann-Whitney	Median	U	p
OM		0.616	0	<0.001
no OM		0.377		
DOC (mg L ⁻¹)	<i>t</i> -test	Mean (SD)	t (df)	p
OM		8.633(0.233)	6.797(22)	<0.001
no OM		7.955(0.255)		
SUVA ₂₅₄ (m ² mg ⁻¹ C)	<i>t</i> -test	Mean (SD)	t (df)	p
OM		2.502(0.120)		
no OM		2.167(0.059)	8.677(22)	<0.001
S ₂₇₅₋₂₉₅ (nm ⁻¹)	<i>t</i> -test	Mean (SD)	t (df)	p
OM		0.023(0.001)	-7.431(26)	<0.001
no OM		0.026(0.000)		
Chl-a (µg L ⁻¹)	<i>t</i> -test	Mean (SD)	t (df)	p
NUT		9.628(3.192)	5.152(26)	<0.001
no NUT		4.697(1.625)		

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1 Supplementary material

2 Table - Absolute values and relative changes of all the parameters measured during 12
3 days of experiment.

Treatment	Days	DOC (mg L ⁻¹)	a ₂₅₄ (m ⁻¹)	a ₄₄₀ (m ⁻¹)	SUVA ₂₅₄ (m ² mg ⁻¹ C)	S ₂₇₅₋₂₉₅ (nm ⁻¹)	S ₃₅₀₋₄₀₀ (nm ⁻¹)	S ₂₅₀₋₄₅₀ (nm ⁻¹)	S _R	Chl- <i>a</i> (µg L ⁻¹)	TSM (mg L ⁻¹)
control	0	8.44	17.38	0.46	2.064	0.025	0.019	0.021	1.336	1.87	1.27
	3	7.76	16.88	0.38	2.176	0.026	0.020	0.022	1.338	3.74	1.87
	6	8.01	17.24	0.38	2.152	0.026	0.020	0.022	1.318	2.67	1.41
	9	7.66	17.30	0.36	2.258	0.027	0.021	0.022	1.317	1.60	1.21
	12	8.37	17.25	0.39	2.064	0.027	0.020	0.022	1.343	2.67	1.61
	<i>relative change (%)</i>	<i>-0.81</i>	<i>-0.73</i>	<i>-15.00</i>	<i>-0.02</i>	<i>5.75</i>	<i>5.21</i>	<i>3.45</i>	<i>0.50</i>	<i>42.86</i>	<i>26.88</i>
NUT	0	7.98	17.38	0.45	2.180	0.025	0.019	0.021	1.279	4.54	1.12
	3	7.97	17.55	0.41	2.202	0.025	0.020	0.021	1.268	10.42	2.15
	6	8.69	17.64	0.40	2.029	0.026	0.019	0.021	1.348	10.16	2.73
	9	8.36	17.68	0.30	2.116	0.026	0.020	0.022	1.294	9.62	2.62
	12	7.90	17.46	0.37	2.211	0.027	0.021	0.021	1.300	23.79	3.60
	<i>relative change (%)</i>	<i>-1.00</i>	<i>0.46</i>	<i>-17.95</i>	<i>1.43</i>	<i>7.90</i>	<i>6.27</i>	<i>1.76</i>	<i>1.64</i>	<i>423.53</i>	<i>220.62</i>
SH	0	7.83	16.67	0.40	2.131	0.025	0.020	0.021	1.253	2.41	1.19
	3	7.73	17.00	0.36	2.200	0.026	0.021	0.021	1.246	2.14	0.78
	6	8.06	17.58	0.54	2.180	0.024	0.019	0.021	1.287	2.41	1.71
	9	7.53	17.18	0.37	2.283	0.026	0.021	0.022	1.286	2.41	1.61
	12	8.05	16.57	0.38	2.058	0.026	0.020	0.022	1.277	3.47	1.64
	<i>relative change (%)</i>	<i>2.79</i>	<i>-0.62</i>	<i>-5.71</i>	<i>-3.43</i>	<i>3.61</i>	<i>1.65</i>	<i>2.15</i>	<i>1.89</i>	<i>44.44</i>	<i>37.89</i>
NUTSH	0	7.65	17.01	0.40	2.226	0.025	0.020	0.021	1.235	3.21	1.42
	3	7.64	17.30	0.41	2.266	0.026	0.020	0.021	1.306	10.96	1.84
	6	8.29	17.91	0.47	2.161	0.025	0.019	0.021	1.308	10.42	2.65
	9	8.24	17.78	0.40	2.159	0.026	0.020	0.021	1.277	9.09	2.29
	12	8.20	17.46	0.45	2.132	0.026	0.020	0.021	1.294	22.45	2.54
	<i>relative change (%)</i>	<i>7.16</i>	<i>2.64</i>	<i>11.43</i>	<i>-4.25</i>	<i>3.21</i>	<i>-1.43</i>	<i>-0.63</i>	<i>4.71</i>	<i>600.00</i>	<i>79.25</i>
OM	0	8.56	21.25	0.64	2.482	0.022	0.019	0.019	1.167	2.67	1.26
	3	8.72	22.38	0.75	2.567	0.022	0.018	0.019	1.202	5.08	2.43
	6	8.72	21.42	0.66	2.457	0.023	0.019	0.019	1.238	3.47	1.87
	9	8.82	22.39	0.68	2.539	0.023	0.019	0.019	1.212	9.36	2.18
	12	8.38	20.84	0.55	2.487	0.024	0.020	0.020	1.222	5.88	2.45
	<i>relative change (%)</i>	<i>-2.12</i>	<i>-1.90</i>	<i>-14.29</i>	<i>0.20</i>	<i>10.58</i>	<i>5.54</i>	<i>7.95</i>	<i>4.77</i>	<i>120.00</i>	<i>93.89</i>
OMNUT	0	8.56	21.67	0.68	2.532	0.022	0.019	0.018	1.146	3.21	3.22
	3	8.30	22.19	0.74	2.675	0.022	0.018	0.019	1.213	13.90	3.00
	6	9.16	22.77	0.67	2.486	0.023	0.019	0.019	1.204	14.17	3.66
	9	9.33	22.23	0.58	2.383	0.023	0.020	0.020	1.196	12.56	4.14
	12	8.48	20.90	0.53	2.464	0.025	0.020	0.020	1.218	14.43	3.81
	<i>relative change (%)</i>	<i>-0.89</i>	<i>-3.56</i>	<i>-22.03</i>	<i>-2.70</i>	<i>12.85</i>	<i>6.22</i>	<i>8.74</i>	<i>6.33</i>	<i>350.00</i>	<i>18.36</i>
OMSH	0	8.61	20.95	0.63	2.432	0.022	0.019	0.019	1.158	3.21	1.30
	3	8.24	21.54	0.71	2.614	0.022	0.019	0.019	1.185	6.42	2.02
	6	8.61	21.46	0.71	2.494	0.023	0.018	0.019	1.263	3.21	2.15
	9	8.54	21.30	0.45	2.494	0.023	0.019	0.019	1.196	12.30	2.61
	12	8.26	21.73	0.63	2.616	0.023	0.019	0.019	1.202	8.02	2.32
	<i>relative change (%)</i>	<i>-4.12</i>	<i>3.74</i>	<i>0.00</i>	<i>7.58</i>	<i>5.16</i>	<i>1.29</i>	<i>3.59</i>	<i>3.82</i>	<i>150.00</i>	<i>78.98</i>
OMNUTSH	0	8.64	21.90	0.70	2.536	0.022	0.019	0.018	1.139	4.81	1.27
	3	8.26	22.93	0.77	2.775	0.022	0.018	0.019	1.177	13.63	2.68
	6	8.75	22.81	0.82	2.606	0.022	0.018	0.018	1.221	16.31	3.89
	9	9.02	22.26	0.68	2.467	0.022	0.019	0.019	1.134	14.43	3.86
	12	8.27	22.91	0.66	2.773	0.023	0.019	0.019	1.185	16.04	6.47
	<i>relative change (%)</i>	<i>-4.31</i>	<i>4.63</i>	<i>-6.56</i>	<i>9.32</i>	<i>7.03</i>	<i>2.86</i>	<i>5.16</i>	<i>4.06</i>	<i>233.33</i>	<i>408.49</i>

4 Negative values in relative changes indicate decrease and positive indicate increase compared to initial values [relative changes = -1*(1 - end/start) x 100]

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CONSIDERAÇÕES FINAIS

1
2 Os estudos das propriedades ópticas aparentes e inerentes são importantes para a
3 compreensão do funcionamento aquático em escala espacial e temporal. Através de
4 simples análise da transparência das águas para a radiação PAR e UV, é possível
5 estimar a distribuição dos produtores primários na coluna d'água, que somado a uma
6 análise qualitativa destes, também se pode ter uma ideia da produtividade ecossistêmica.
7 Avaliando também a análise da concentração e qualidade de DOM é possível
8 compreender a dinâmica sazonal de entrada alóctone, produção autóctone e os
9 principais meios de degradação de compostos de carbono nos ambientes.

10 A utilização de índices obtidos através de análises espectrais de CDOM para a sua
11 caracterização possui a vantagem de ser relativamente de baixo custo e com resultados
12 rápidos. Apesar da crescente utilização desta abordagem em ambientes temperados,
13 principalmente após o trabalho de Helms et al. (2008), ainda são comuns estudos que
14 avaliam apenas a concentração de DOC e CDOM, e bastante escassos nos trópicos e no
15 Brasil aqueles que consideram também a qualidade dos compostos de carbono. Nesta
16 tese, foi demonstrado que incluir a análise da qualidade espectral de CDOM fornece
17 resultados extremamente importantes (quanto a fontes, tamanhos das moléculas,
18 susceptibilidade aos processos de degradação, etc) para o entendimento dos processos
19 biogeoquímicos e ecológicos, não obtidos apenas pela análise quantitativa. Inúmeras
20 são as vantagens do estudo da qualidade de DOM através da análise espectral de
21 CDOM, no entanto, é preciso cautela no uso e interpretação dos índices, pois alguns
22 podem variar entre pesquisadores sem explicações bem definidas (por exemplo os
23 intervalos de comprimentos de onda para cálculo dos “spectral slopes”).

24 As amostragens no lago Dom Helvécio (MG, Brasil) contemplaram um período de
25 redução na pluviosidade regional do Médio Rio Doce, e alguns parâmetros (por
26 exemplo clorofila-a, DOC, nutrientes) puderam ser comparados com o observado
27 anteriormente durante 10 anos de monitoramento pelo programa PELD. Além disto,
28 através da análise espectral do CDOM e seus índices também foi possível demonstrar
29 como as recentes mudanças climáticas podem afetar a sazonalidade da entrada de
30 matéria orgânica alóctone no lago e os potenciais efeitos disto para o ecossistema. Os
31 resultados demonstrados nos capítulos 1 e 2 desta tese permitiram ainda estimar qual o

1 efeito na atenuação da radiação PAR em função da redução do regime de chuvas, que
2 por sua vez afeta as dinâmicas dos componentes opticamente ativos.

3 Compreender as fontes de DOM nos ecossistemas aquáticos, bem como a sazonalidade
4 destas renovações de DOM e nutrientes, é de grande auxílio para o entendimento dos
5 processos biogeoquímicos, ecológicos e metabólicos. Somente após esta compreensão é
6 possível investigar como as mudanças climáticas, alterações do uso do solo, aportes
7 esporádicos de esgotos, tempestades, dentre outros fenômenos, podem afetar a dinâmica
8 de DOM e quais as consequências para os ambientes aquáticos.

9 A utilização de experimentos *in situ* com mesocosmos permitiu ainda avaliar como o
10 aumento de DOM autóctone e DOM alóctone podem alterar a qualidade espectral dos
11 compostos de carbono. Dessa forma, é possível extrapolar tais resultados para
12 estimativas das consequências da eutrofização (por exemplo, o reservatório da
13 Pampulha), desmatamento, substituição da vegetação do entorno por eucaliptos ou
14 pastos (comuns no entorno do Parque Estadual do Rio Doce), e outros impactos que
15 interferem na quantidade de DOM autóctone e alóctone disponível nos sistemas.

16 Os resultados apresentados nesta tese contribuem com o conhecimento sobre
17 propriedades ópticas aparentes e inerentes, e qualidade e concentração de DOM em
18 ecossistemas aquáticos tropicais naturais e artificiais, em escala espacial e temporal.
19 Além disso, fornecem novas perspectivas no estudo do efeito das diferentes fontes de
20 carbono na qualidade espectral de CDOM, o que reflete nos processos de degradação e
21 características ópticas dos sistemas. Novos estudos que investiguem especificamente as
22 taxas de fotodegradação e biodegradação nestes ambientes são necessários para melhor
23 entendimento da dinâmica de DOM e do papel destes processos no ciclo global do
24 carbono em ambientes aquáticos tropicais.

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